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2011

The Missing Link: The Ecology of the Serpentine and the Implications for East and North Ponds [Presentation]

Colby Environmental Assessment Team, Colby College

Problems in Environmental Science course (Biology 493), Colby College

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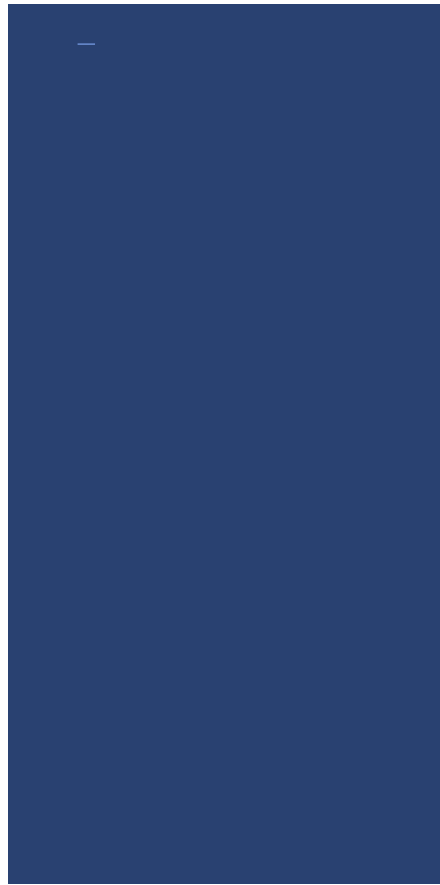
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The Missing Link: The Ecology of the Serpentine and the Implications for East and North Ponds



Colby Environmental Assessment Team
Smithfield Town Hall
December 8, 2011

Importance of Freshwater Ecosystems



■ **Tourism**

- Over 6000 lakes in Maine
- Fishing, boating, swimming, etc.

■ **Lake Economics**

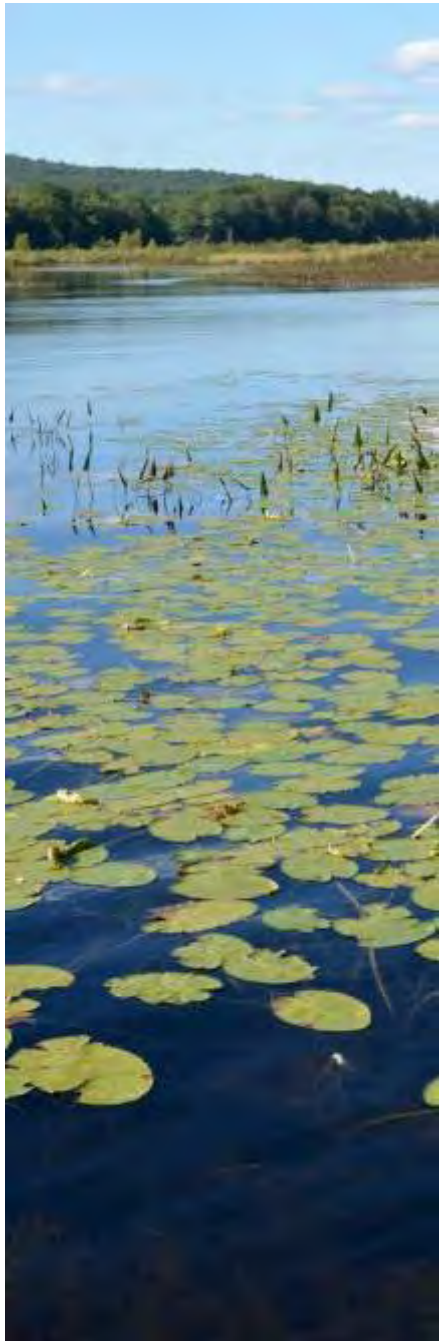
- \$6.7 billion dollars annually

■ **Ecosystem services**

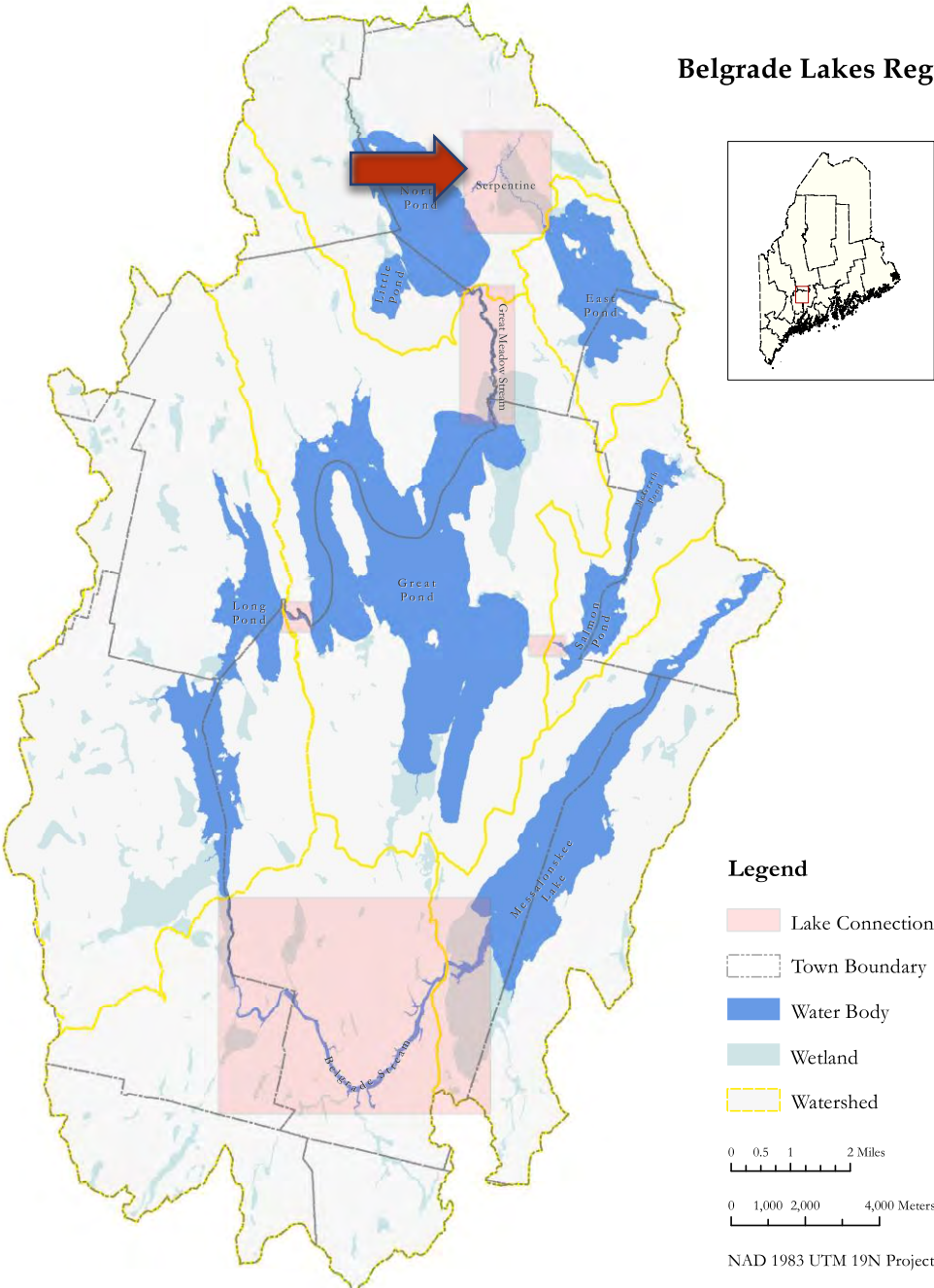
- Water source for municipal and agricultural sectors
- Water filtering
- Flood buffer
- Host of diverse plant, animal & fish species

■ **Eutrophication Threat**





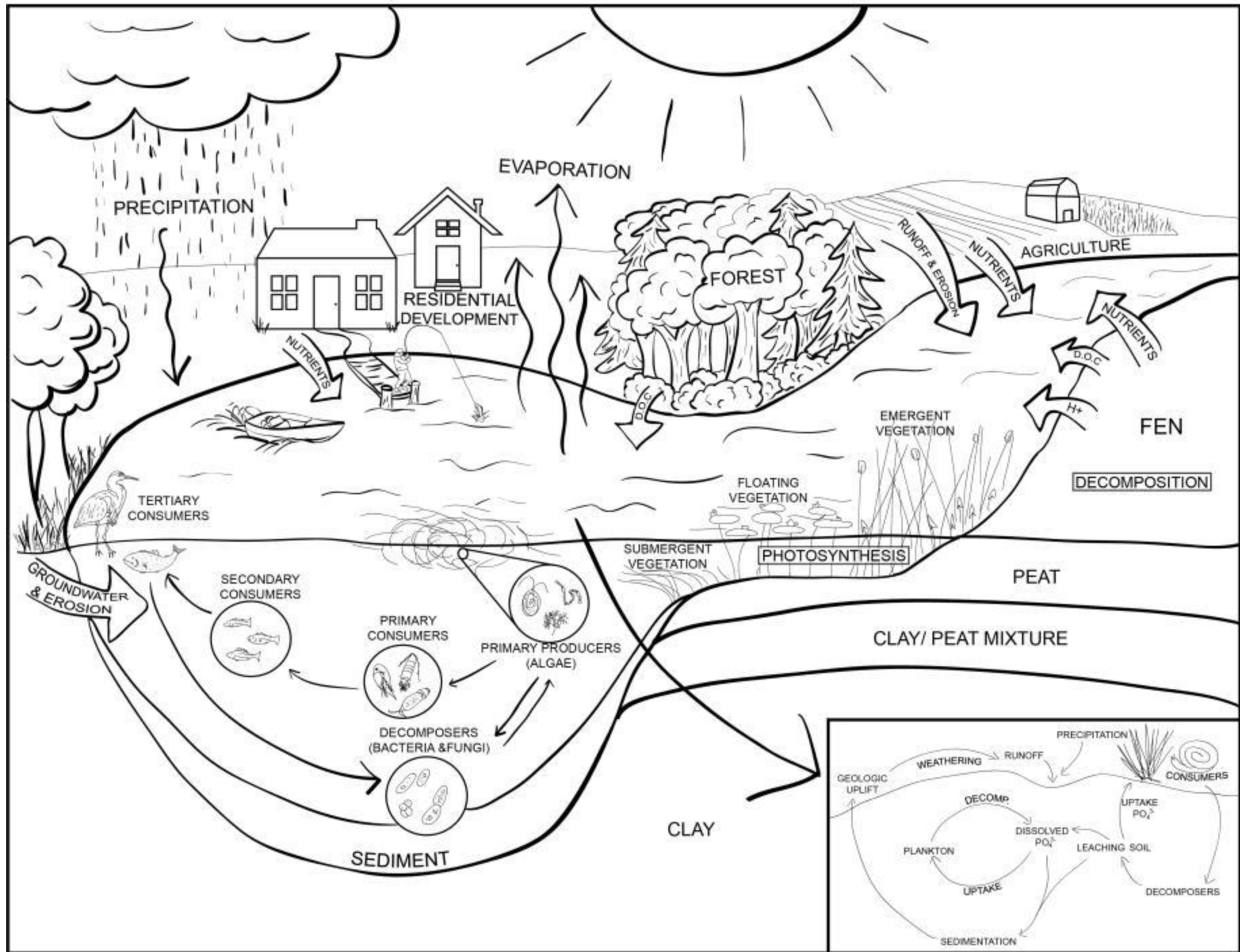
Belgrade Lakes Region



Purpose of Study

- Conduct a broad survey of the Serpentine system
- Understand what effects the Serpentine has on East and North Ponds





Project Organization

Spatial Analysis

Chemistry

Algae

Fish

Plants

Sediment

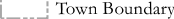

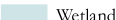
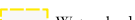





Conclusions

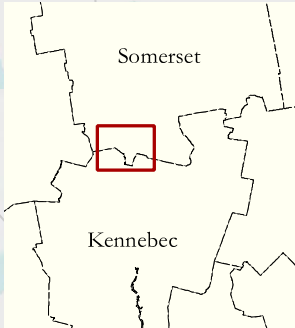
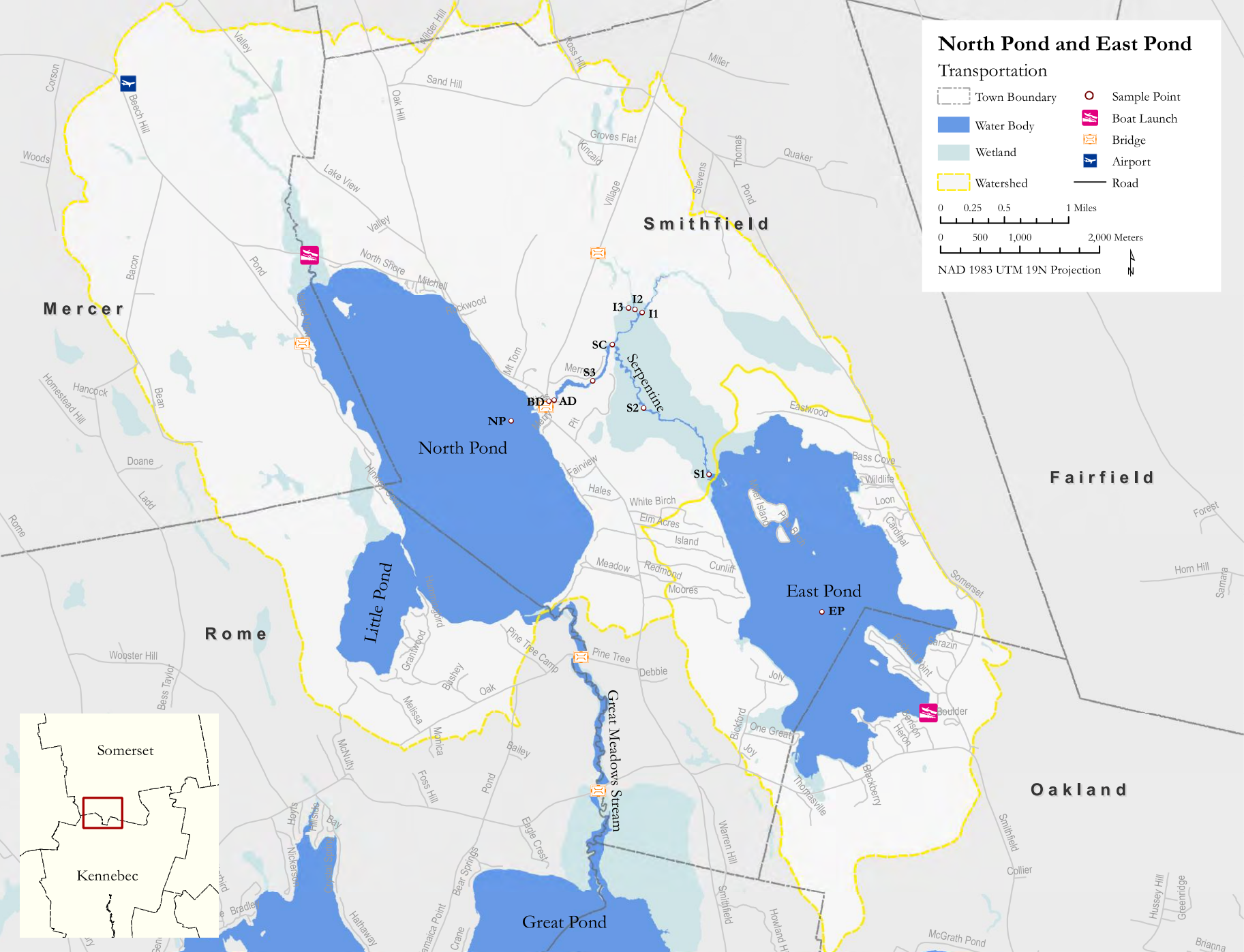
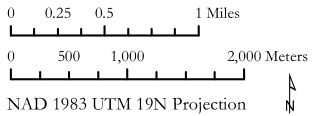
Implications

Questions



North Pond and East Pond Transportation

-  Town Boundary
-  Water Body
-  Wetland
-  Watershed
-  Sample Point
-  Boat Launch
-  Bridge
-  Airport
-  Road



Watershed and Sample Sites



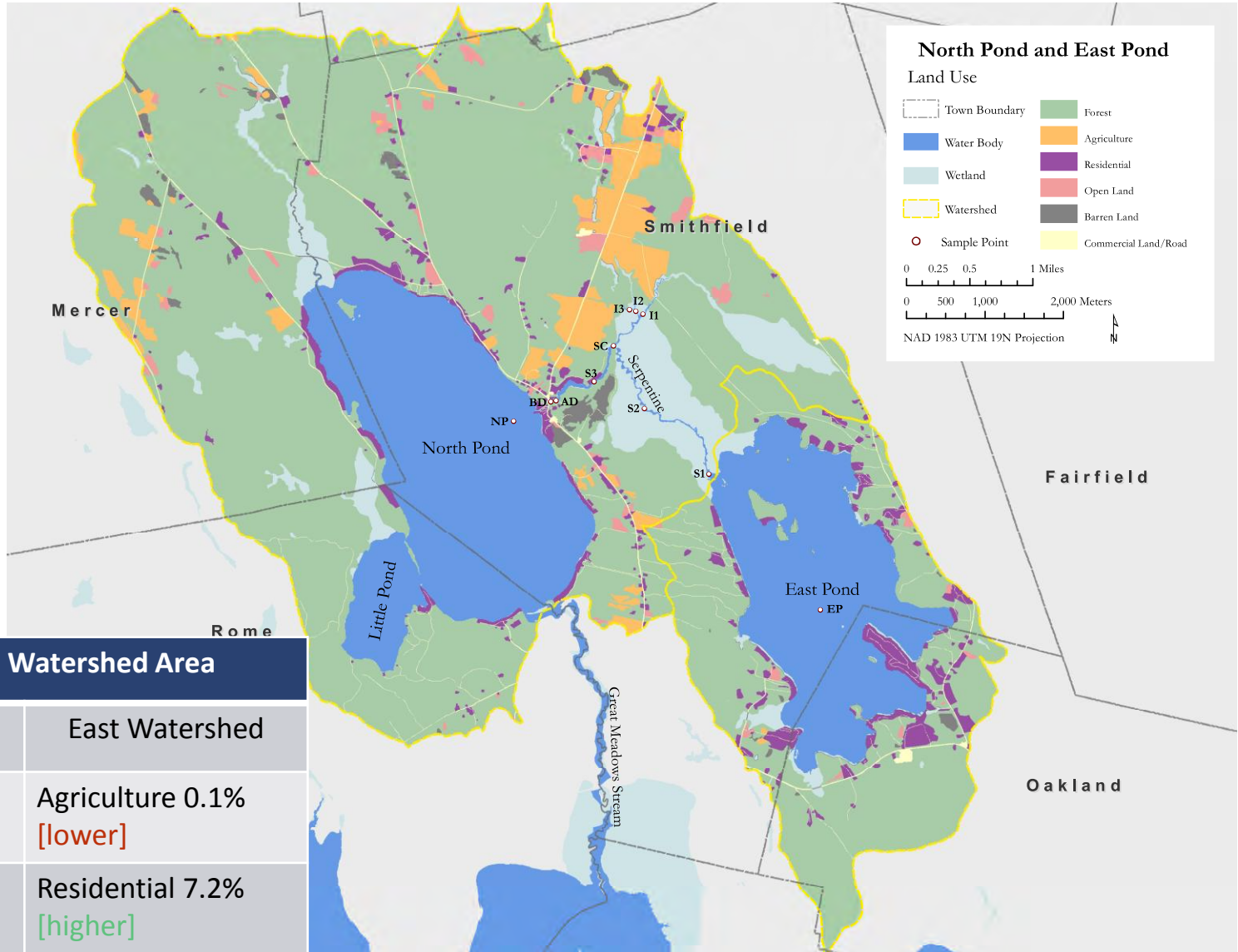


Spatial Analysis

Our objectives:

- Visually display physical parameters of our study area
- Quantify environmental factors contributing to ecosystem health
- Model environmental processes

Land Use

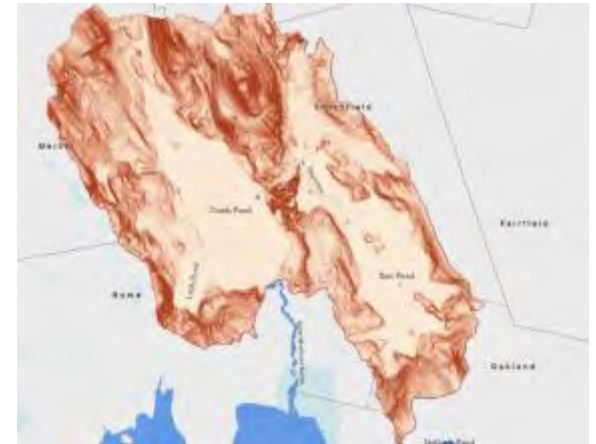
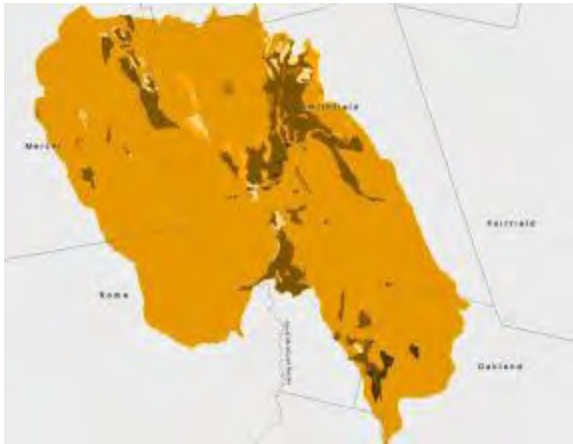


Percent Total Watershed Area

North Watershed	East Watershed
Agriculture 4.9% [higher]	Agriculture 0.1% [lower]
Residential 2.2% [lower]	Residential 7.2% [higher]

Erosion Potential Modeling

Erosion Potential Rating =



$$\text{Soil Type EPR} * 0.4 + \text{Land Use EPR} * 0.3 + \text{Slope EPR} * 0.3$$

North Pond and East Pond

Erosion Potential (Mid)

--- Town Boundary ○ Sample Point

■ Water Body

■ Wetland

■ Watershed

■ High Erosion Potential

Erosion Potential

■ High

■ Low

0 0.25 0.5 1 Miles

0 500 1,000 2,000 Meters

NAD 1983 UTM 19N Projection



Mercer

Smithfield

Fairfield

Rome

Oakland

North Pond

East Pond

Little Pond

Great Meadows Stream

Serpentine

NP

BD

AD

SC

S3

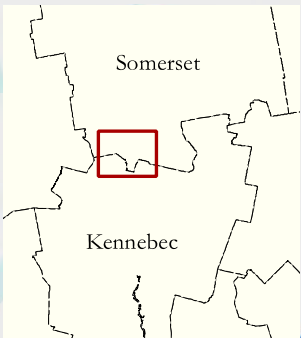
S2

S1

12

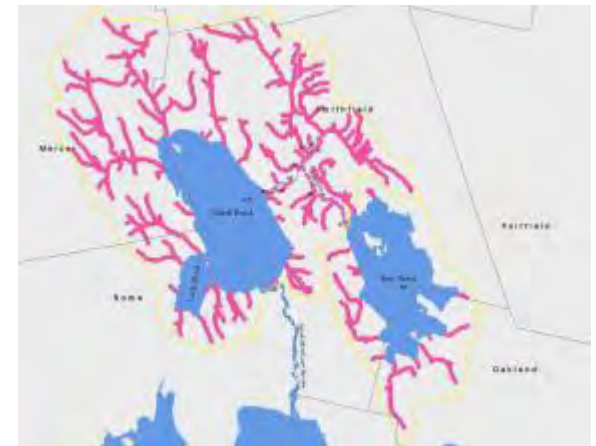
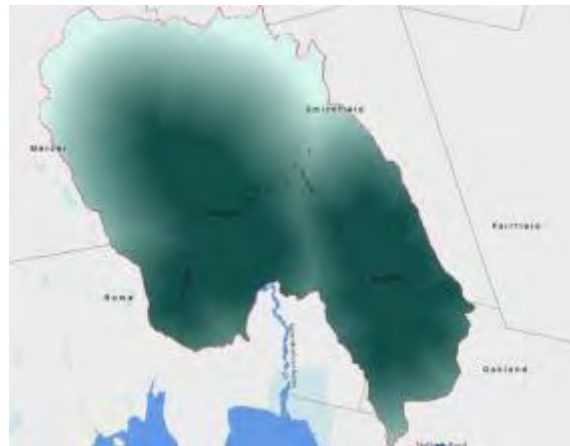
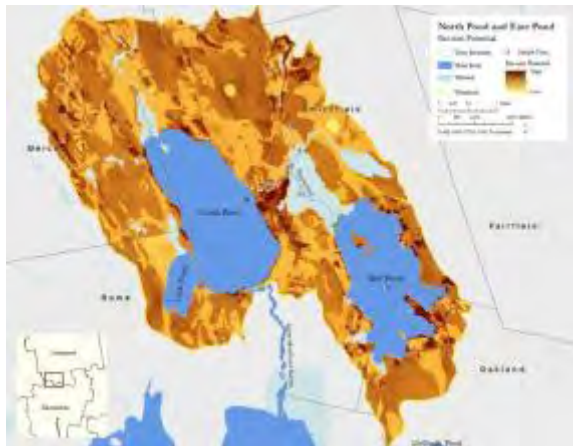
13

11



Erosion Impact Modeling

Erosion Impact Rating =



$$\text{Erosion Potential} * 0.5 + \text{Lake Proximity EPR} * 0.4 + \text{Overland Flow Path Proximity} * 0.1$$

North Pond and East Pond

Erosion Impact (Mid)

--- Town Boundary ○ Sample Point

■ Water Body

■ Wetland

■ Watershed

Erosion Impact

■ High

■ Low

■ High Erosion Impact

0 0.25 0.5 1 Miles

0 500 1,000 2,000 Meters

NAD 1983 UTM 19N Projection



Mercer

Smithfield

Fairfield

Rome

North Pond

East Pond

Little Pond

Great Meadows Stream

Serpentine

NP

BD AD

S1

S2

S3

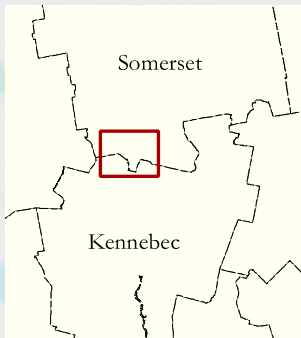
SC

I1

I3

I2

EP





Conclusions and Implications



- Still must maintain buffer zone
 - MEDEP regulations are effective at reducing nutrient transport

- Erosion models highlight areas of concern
 - Land use correlation
 - Proximity correlation

- Look into Lake Smart!

- Future: Targeted upland surveys



Water Chemistry

Our Objective:

- Determine if the water chemistry of the Serpentine influences the water chemistry in North Pond and East Pond

Why is water chemistry important?

- Directly influences timing, magnitude and frequency of algal blooms
- Anthropogenic inputs may cause excessive nutrients and cultural eutrophication
- Excessive nutrients can negatively impact recreation, human health and biotic communities

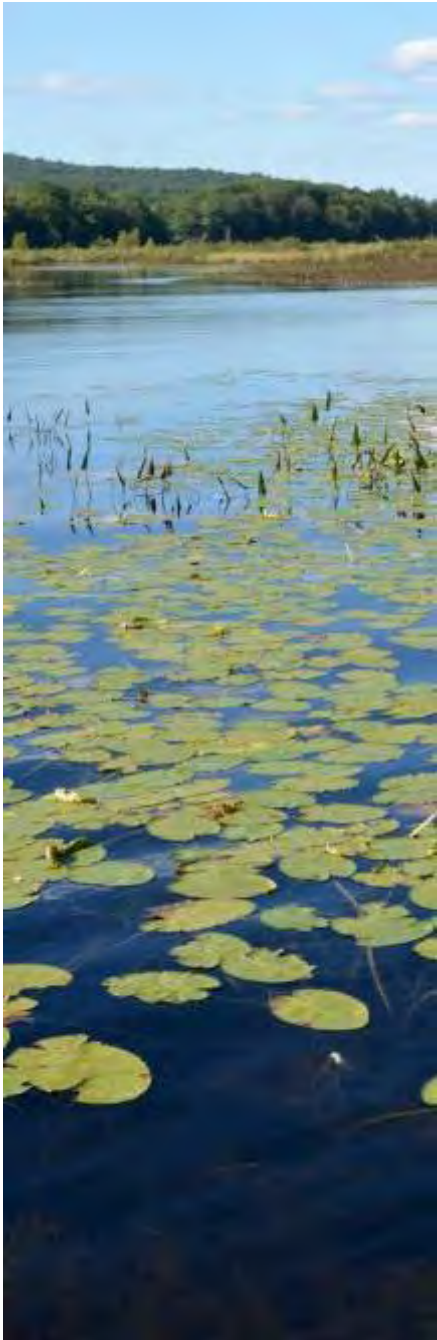


Sampling

- Samples were taken over the course of two weeks on 4 days: September 22, September 29, October 3 and October 6, 2011
- 10 sites in or directly connected to the Serpentine were sampled



Analysis



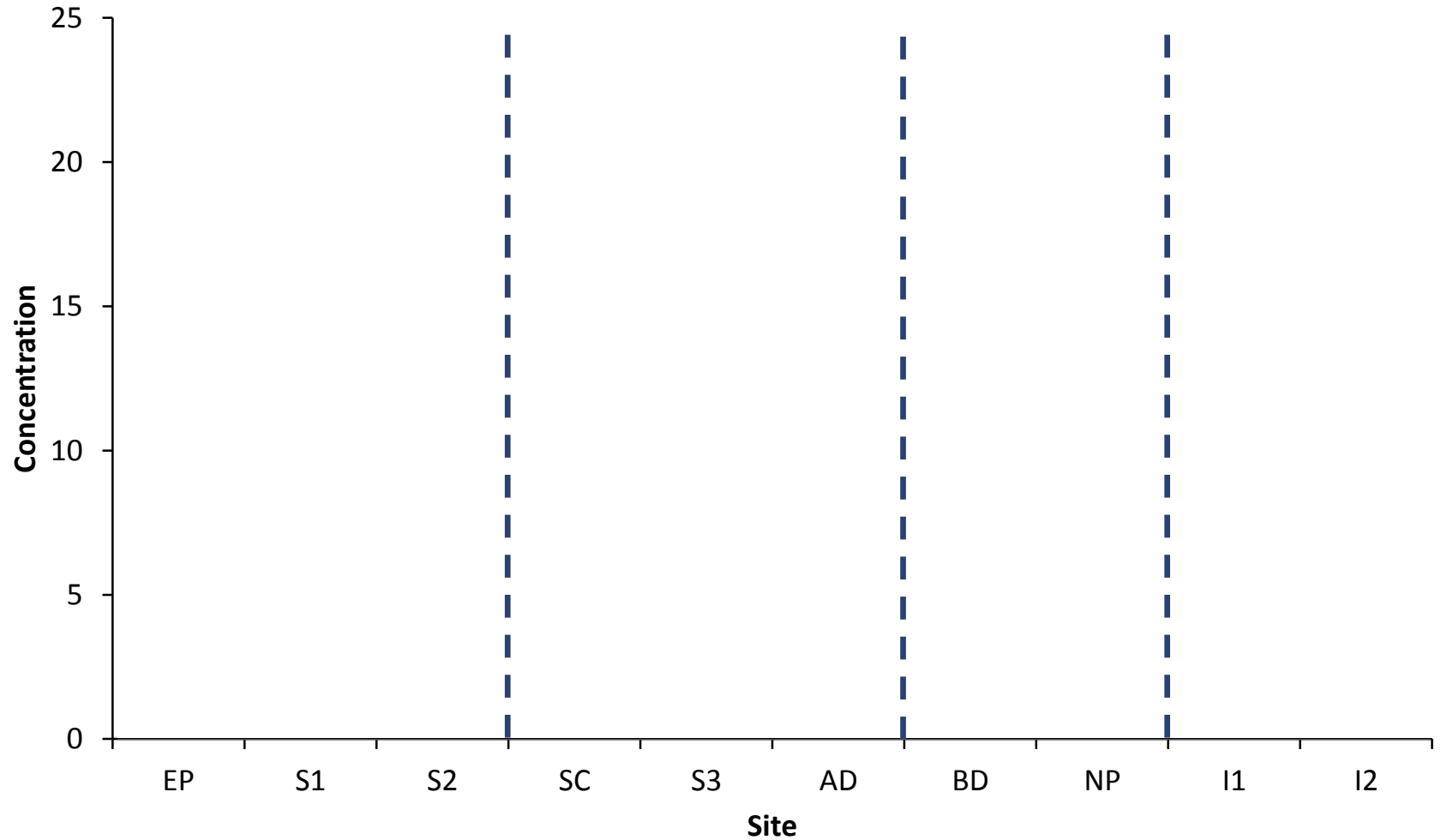
- Measured dissolved oxygen (DO), temperature and pH in the field
- Filtered and unfiltered samples were analyzed using an ICP-AES
 - TP, Fe, Al, Ca, Mg
- DOC and TN concentrations were determined using a Shimadzu TOC analyzer
- A Lachat auto-analyzer was used to determine nitrate concentrations

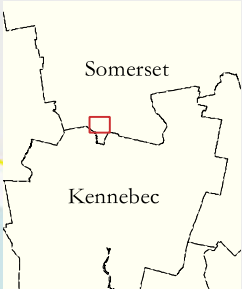
Dissolved Oxygen

- A measure of the amount of oxygen dissolved in the water
- Influenced by temperature, barometric pressure, mixing and decomposition
- DO determines what fauna can survive
- Plays a role in redox chemistry
- Hypoxia is less than 5.0 mg/L
- Anoxia is less than 2.5 mg/L



Example of Results





Serpentine

- Sample Point
- ▭ Watershed

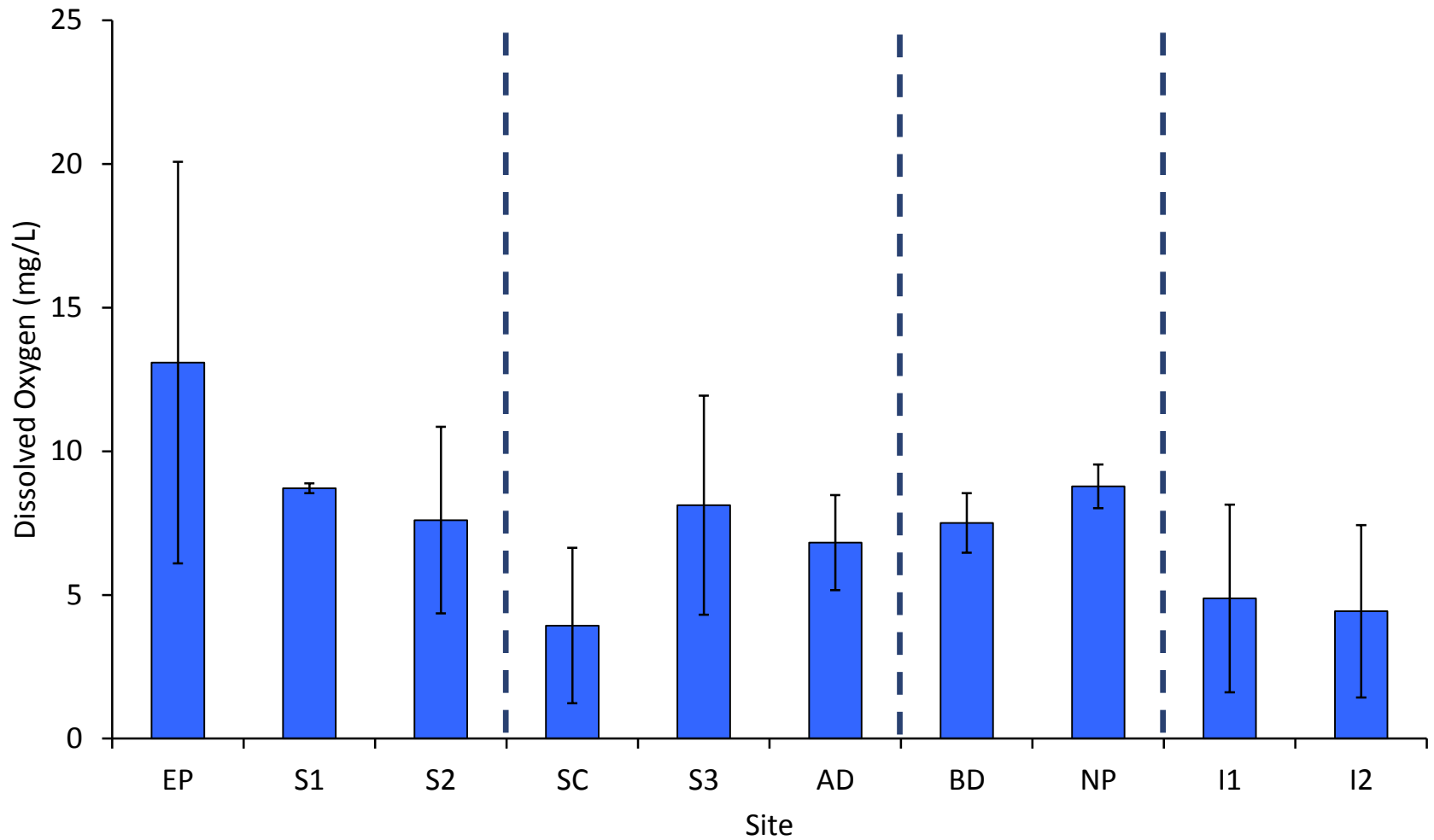
0 487.5 975 1,950 Feet

0 95 190 380 Meters

NAD 1983 UTM 19N Projection

○ EP

Average Dissolved Oxygen by Site

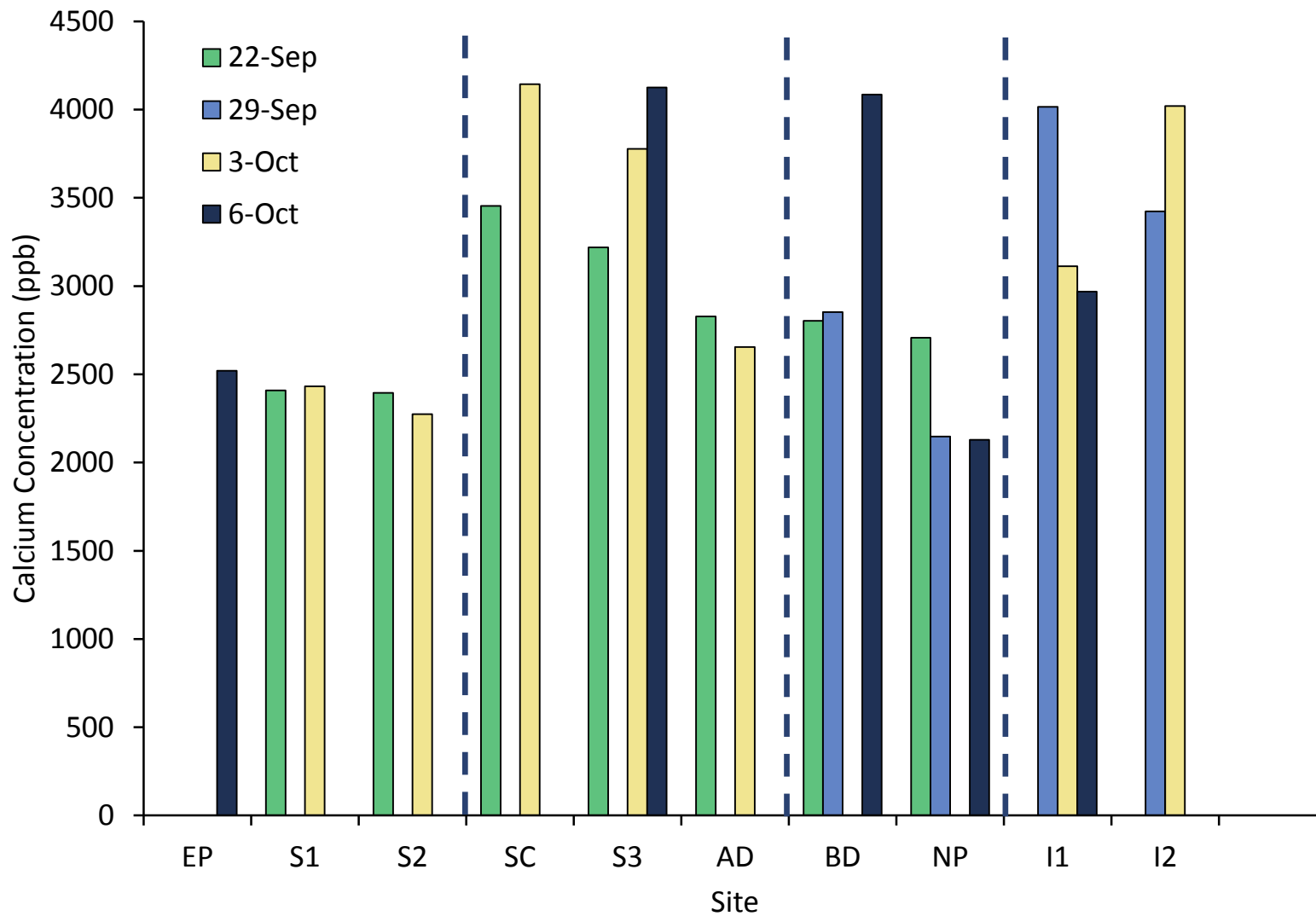


Calcium & Magnesium

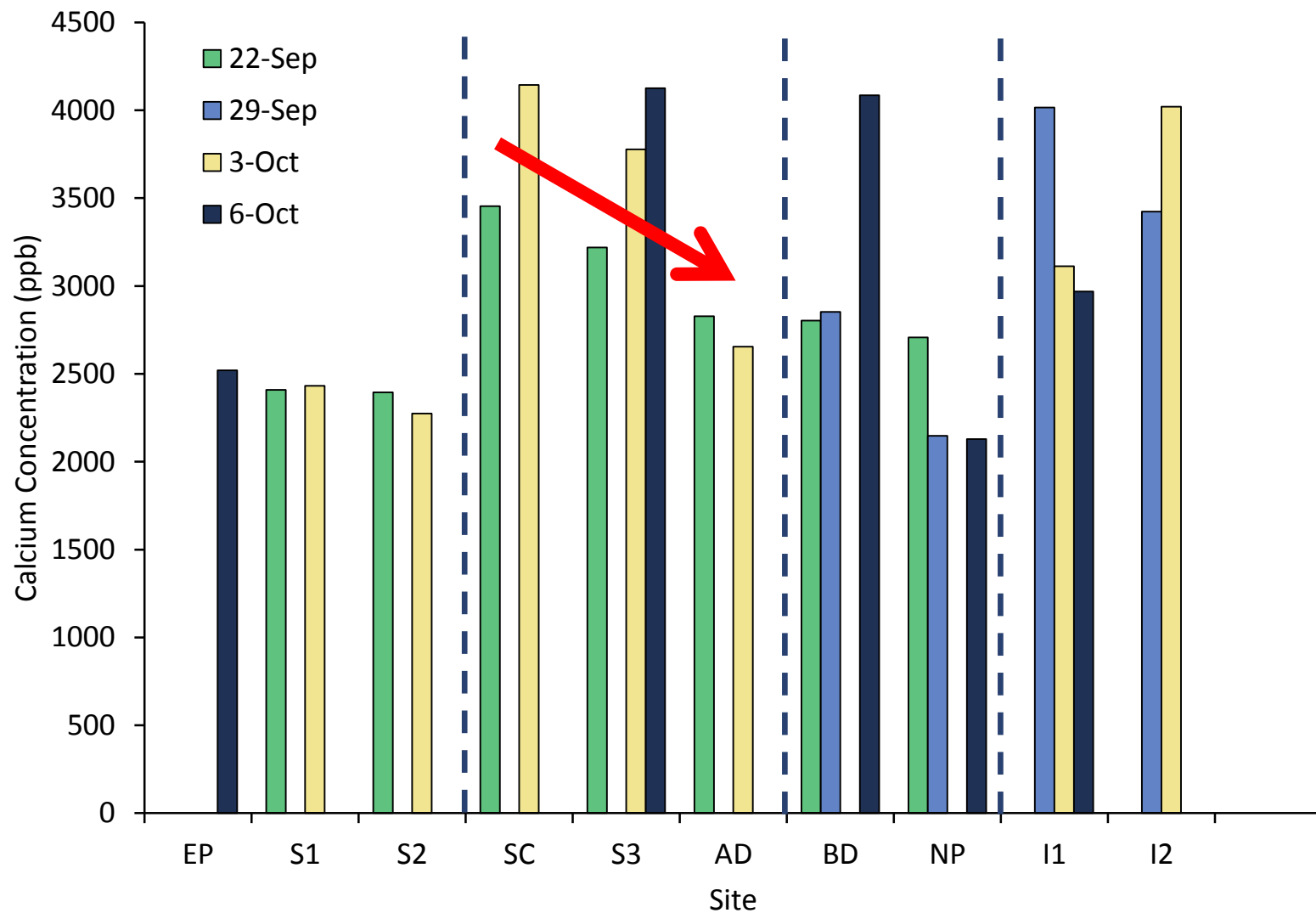
- Considered non-reactive elements in the environment
- Both are found primarily in groundwater, however calcium has significant anthropogenic inputs (industry & agriculture)
- With fewer anthropogenic sources, magnesium is possibly more accurate in determining true groundwater sourcing than calcium
- < 10 ppm characteristic of precipitation or short groundwater flow paths



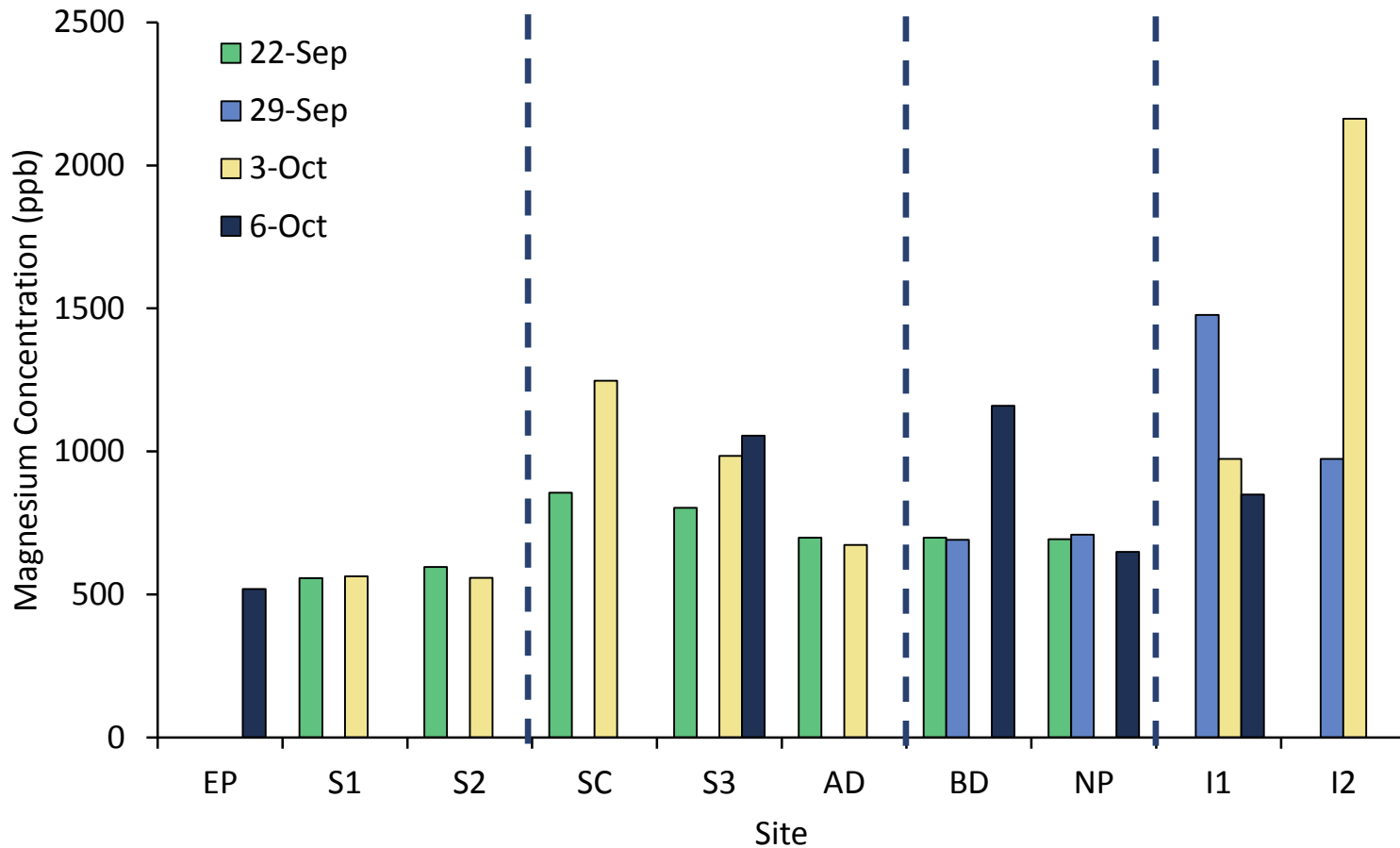
Calcium by Site by Day



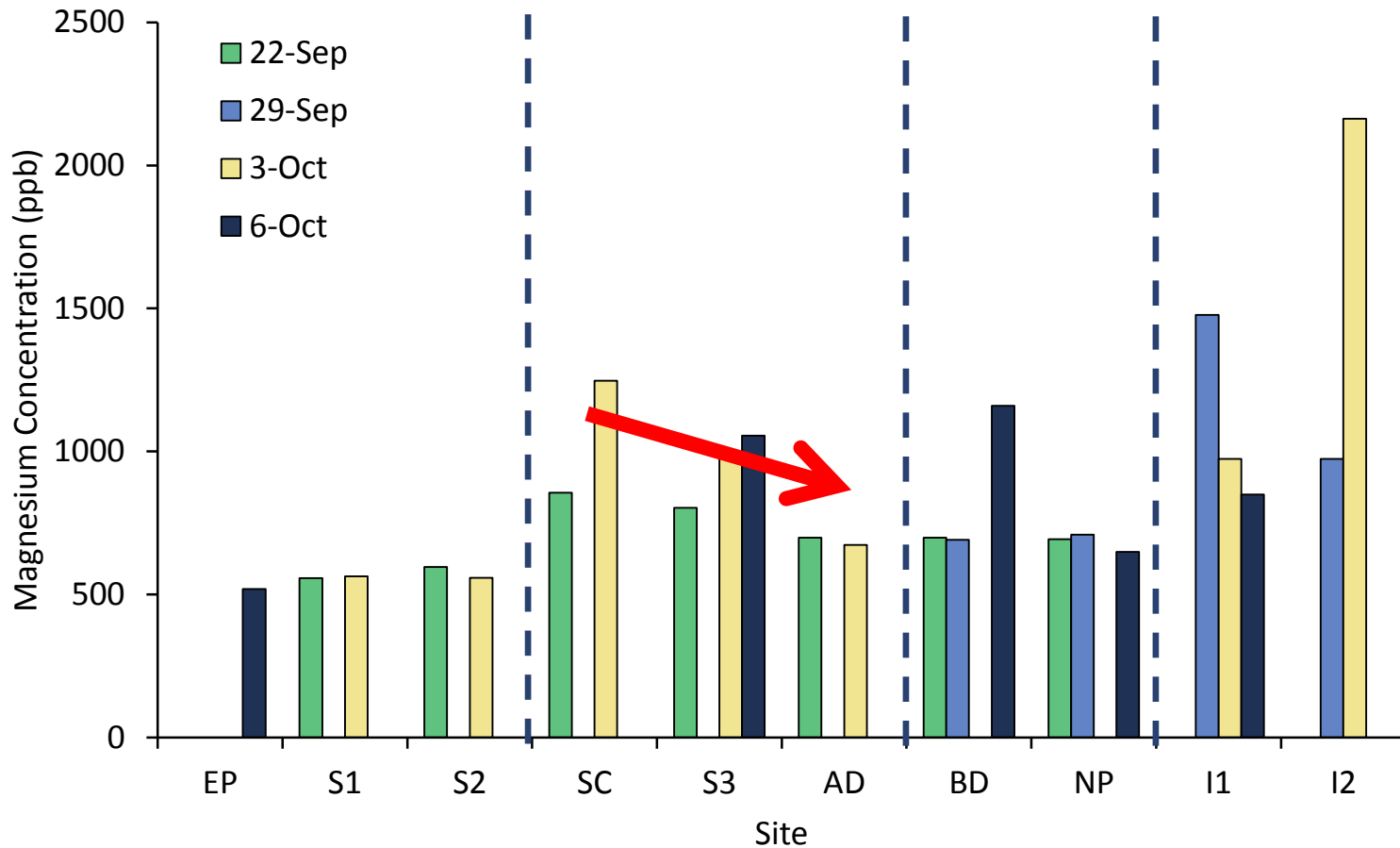
Calcium by Site by Day



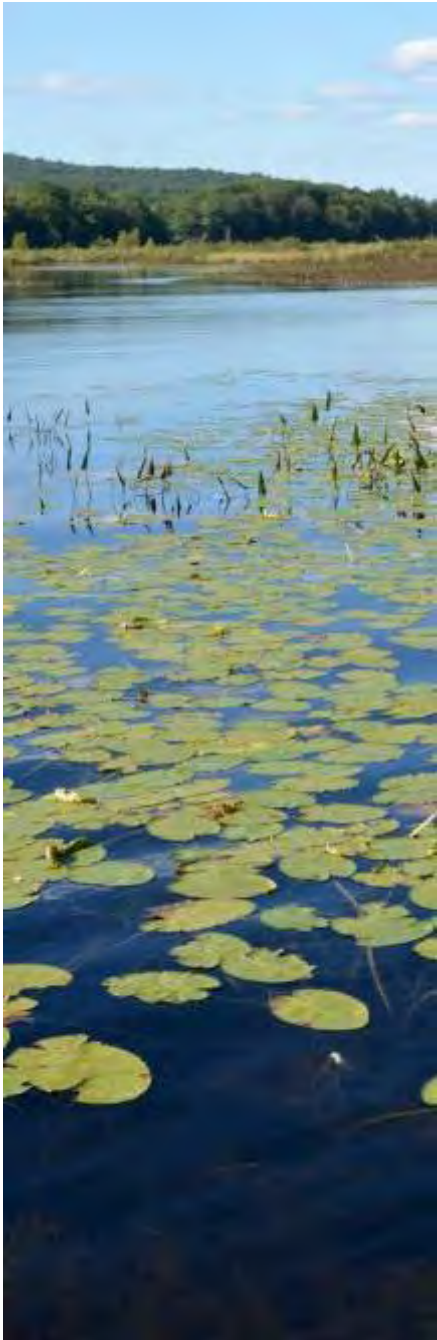
Magnesium by Site by Day



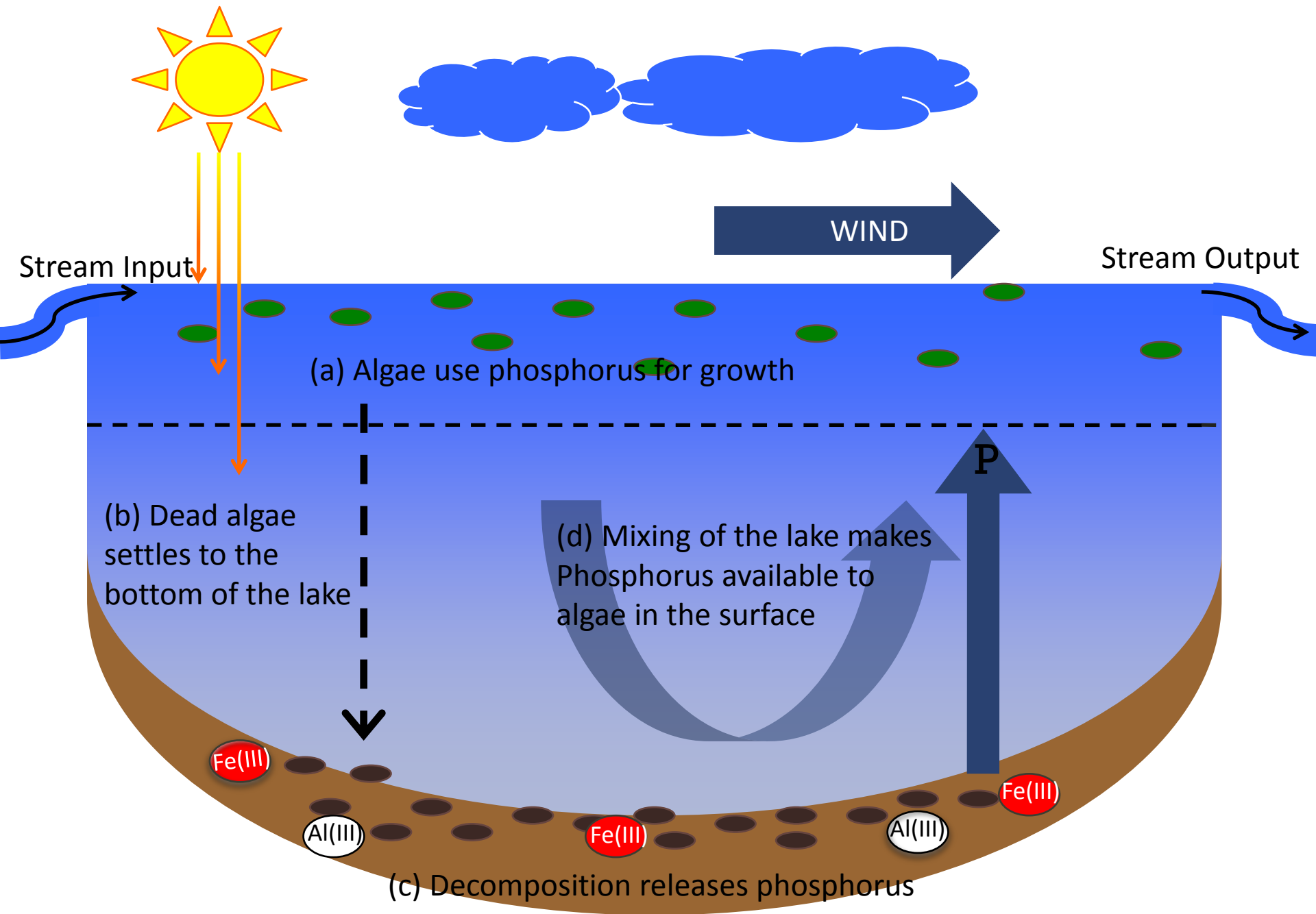
Magnesium by Site by Day



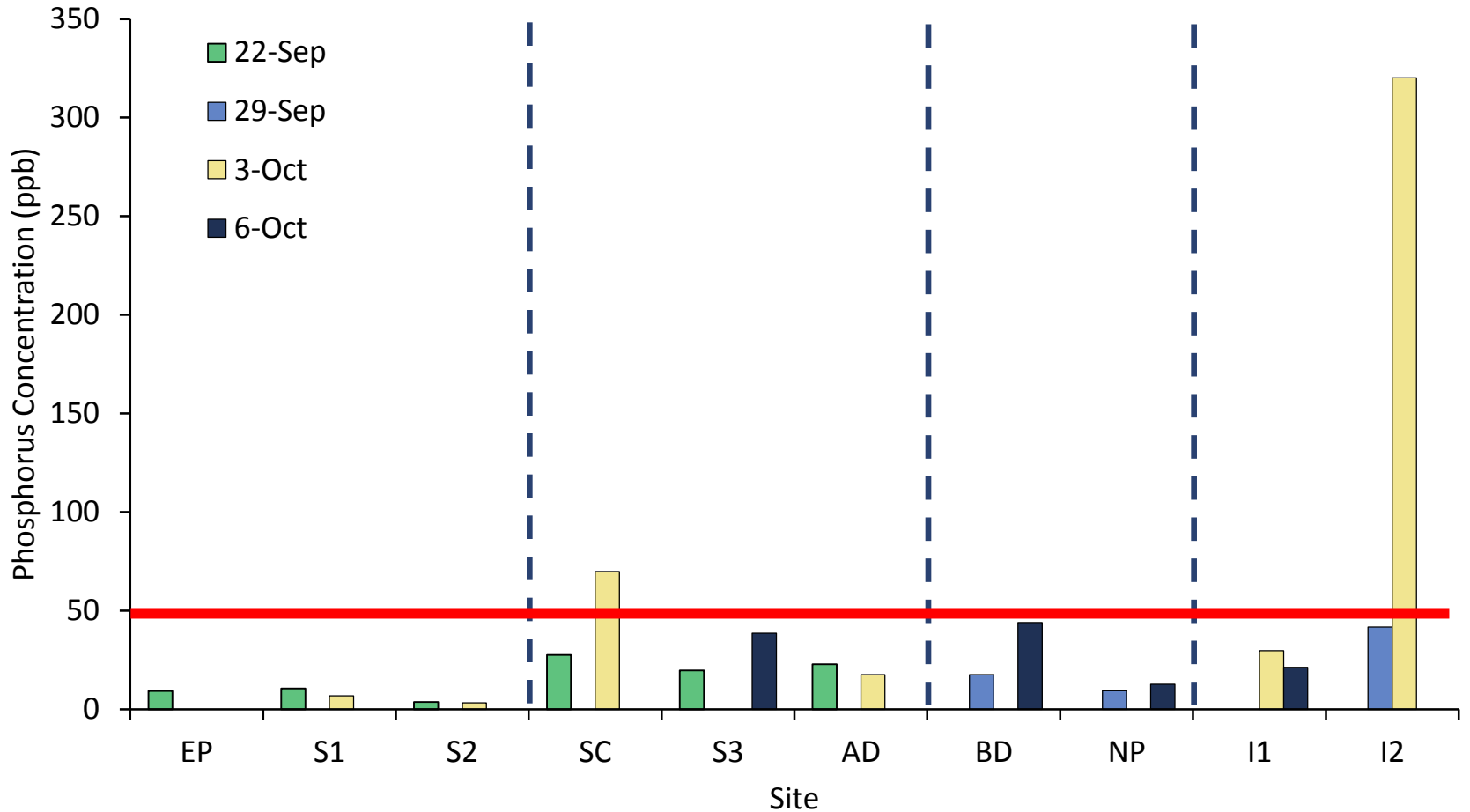
Phosphorus



- An important limiting nutrient, therefore enrichment could lead to eutrophication
- Humans have drastically increased available phosphorus in prior years, altering the nutrient cycling
- It can be introduced through surface runoff and released from internal loading
 - Released from Fe complexes during anoxia



Total Phosphorus by Site by Day

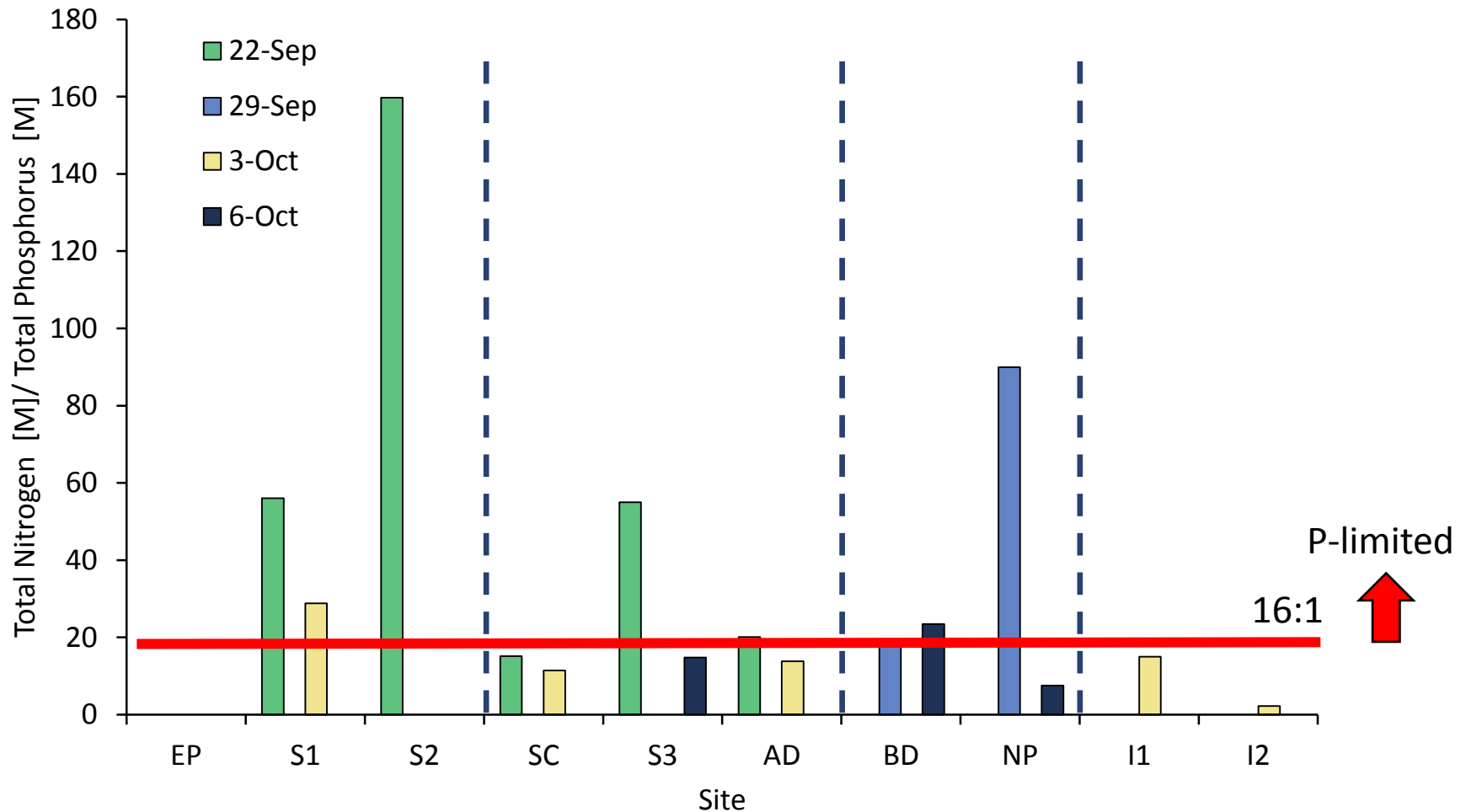


Redfield Ratio & TN:TP Ratio

- The Redfield ratio is a total nitrogen to total phosphorus ratio
 - Defined as 16:1
- The point at which a system shifts from phosphorus limited to nitrogen limited



Total Nitrogen : Total Phosphorus ([M]:[M])

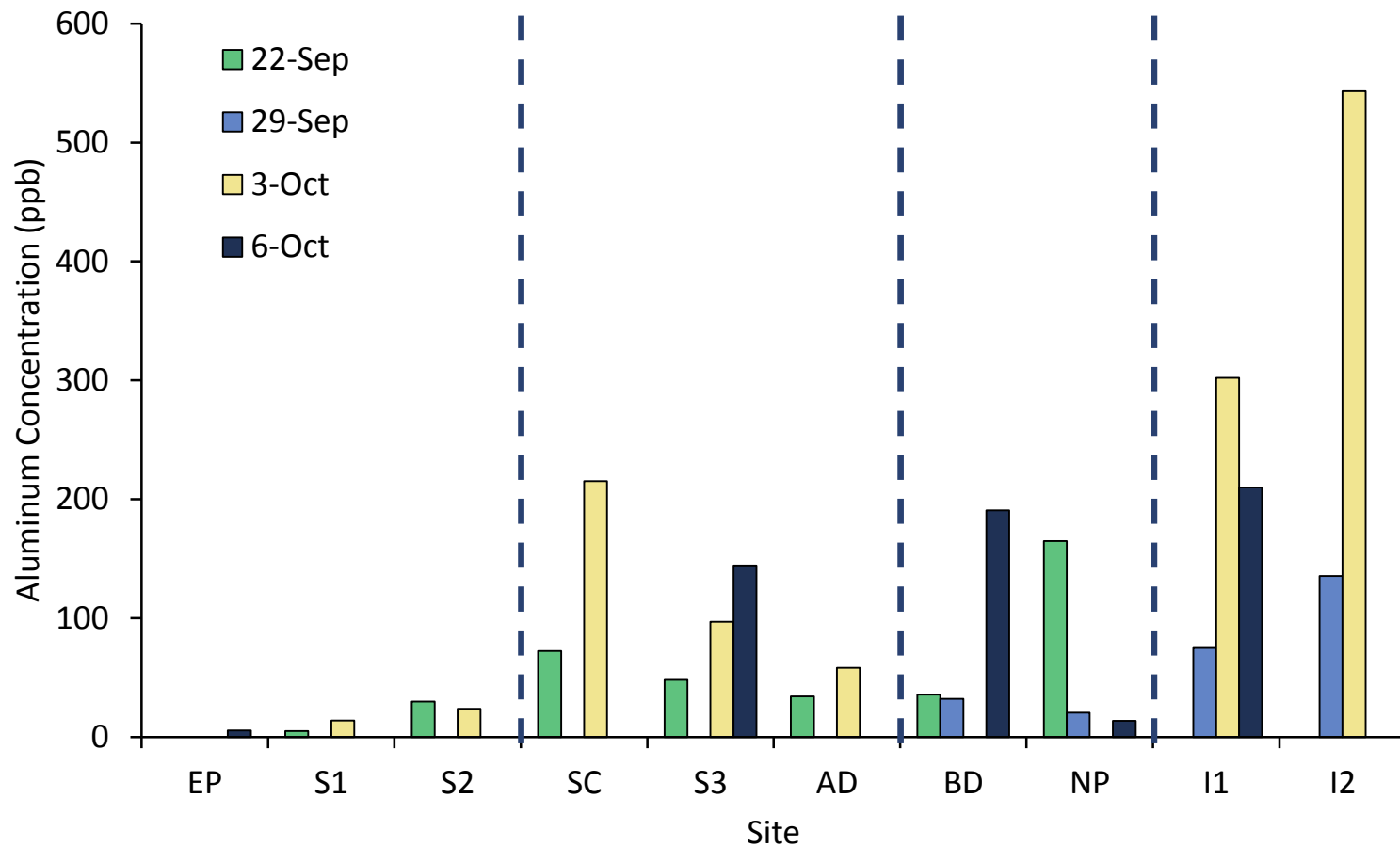


Aluminum

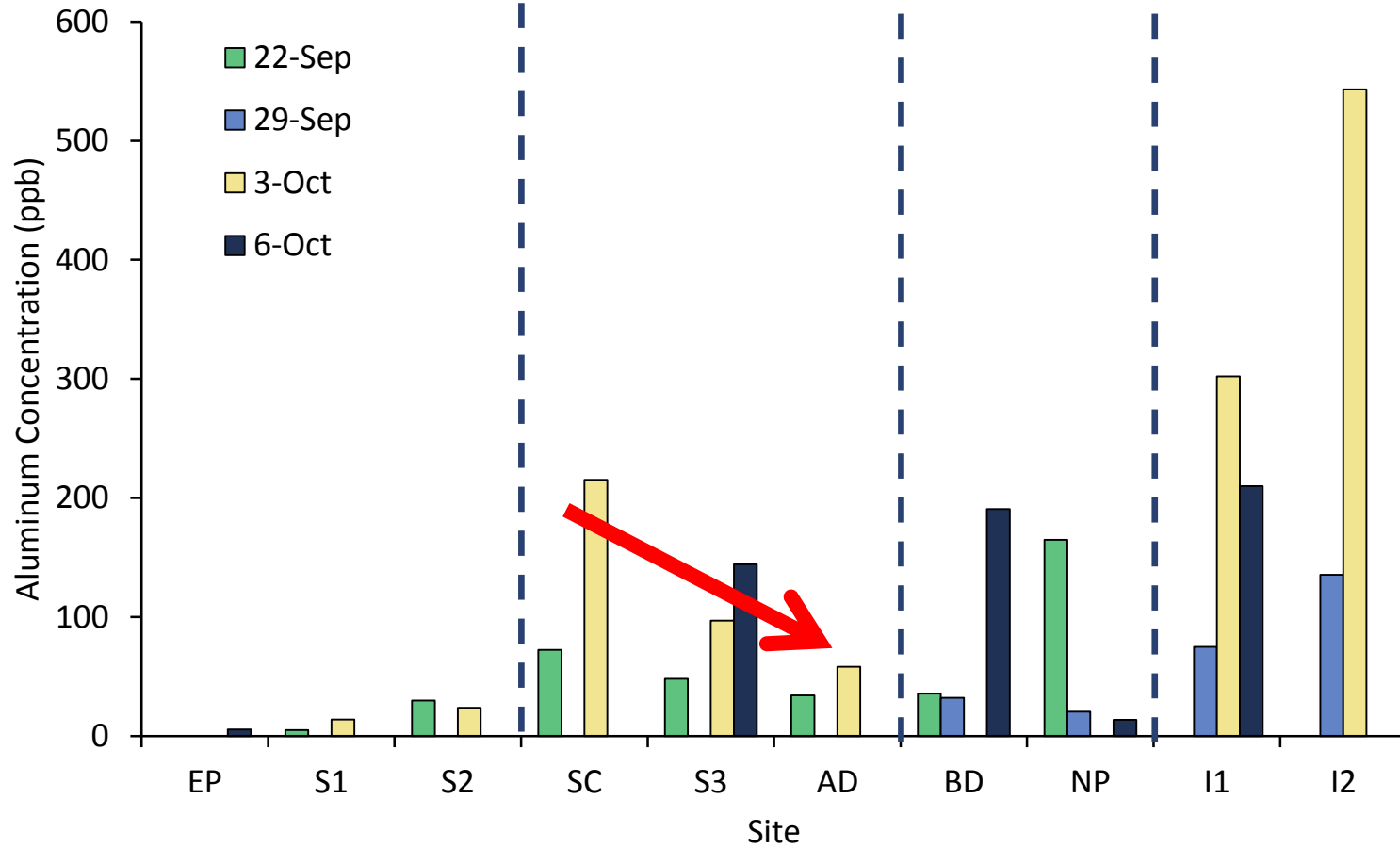
- An important metal which has the ability to permanently bind phosphorus
- Maine lakes had an average total aluminum concentration of 48 ppb (Brakke)
- Streams sampled in September and October averaged 150.5 ppb (Nelson and Johnson 2003)
- All sample sites averaged 112.01 ppb

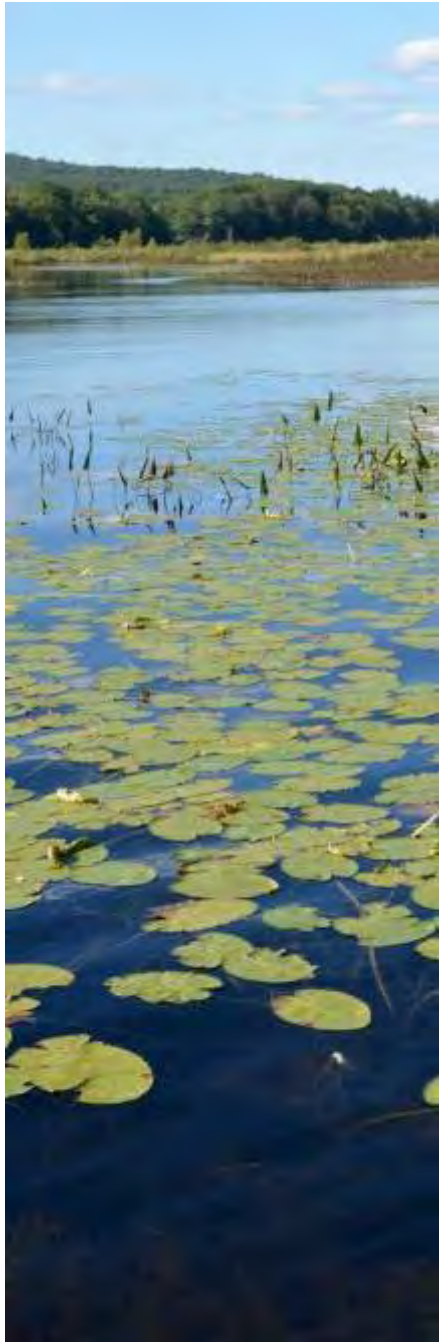


Aluminum by Site by Day



Aluminum by Site by Day

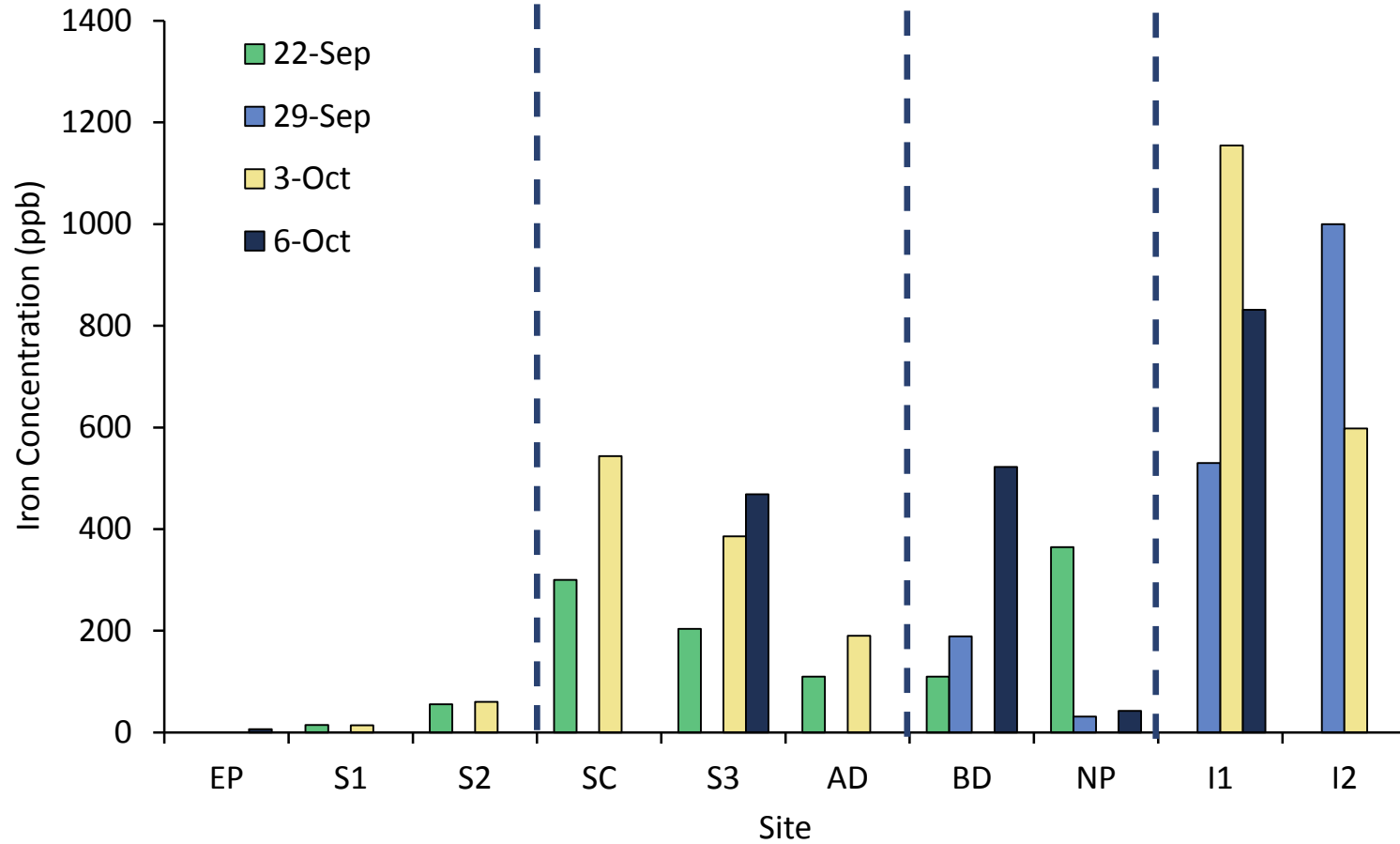




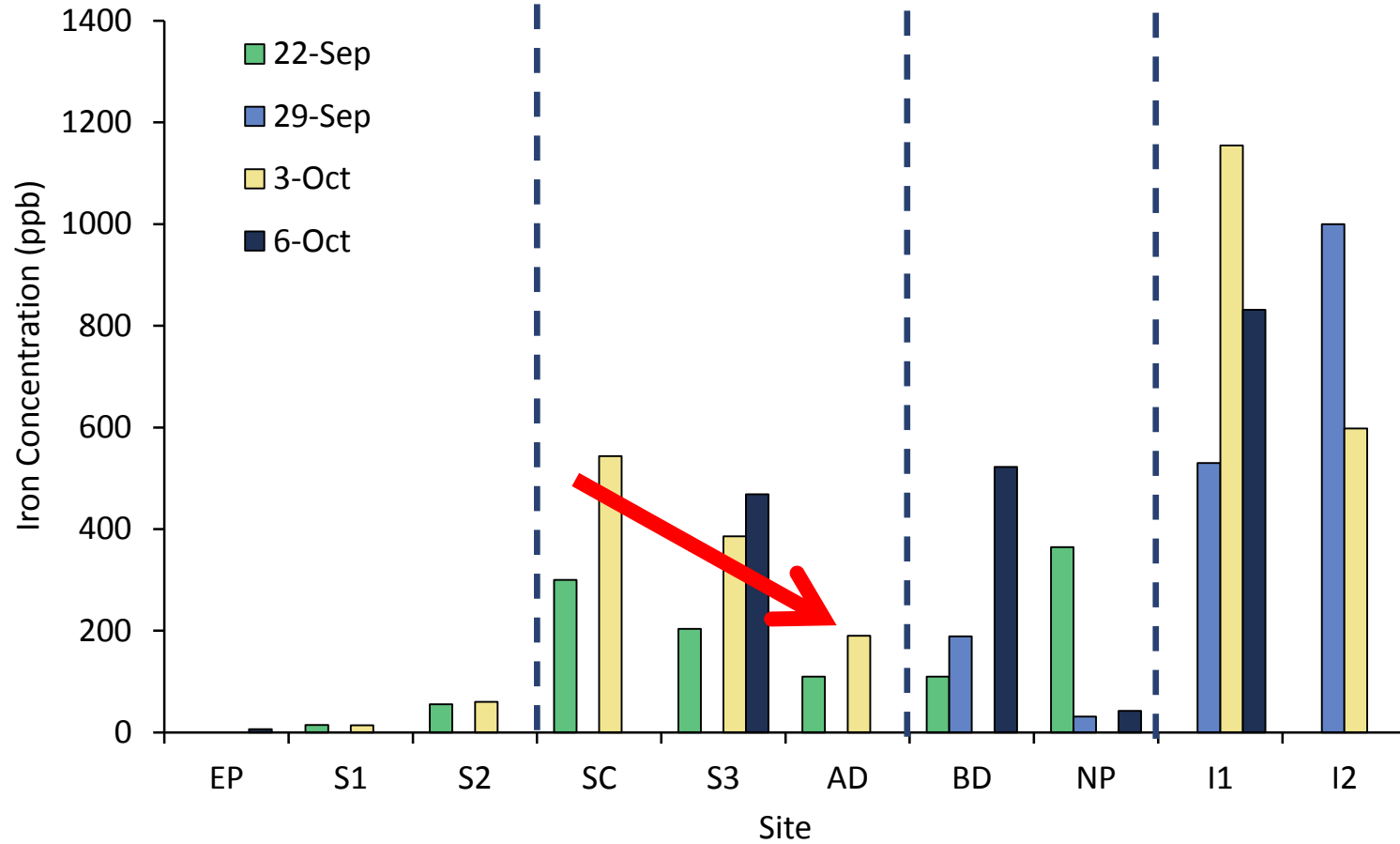
Iron

- An important element for phosphorus sequestration, however it can release phosphorus after binding, unlike aluminum
- Releases phosphorus during periods of hypoxia or anoxia
- Phosphorus is released due to use of iron as a reducing agent during anaerobic respiration
- Average of 40 ppb in Maine lakes (Brakke et al.)

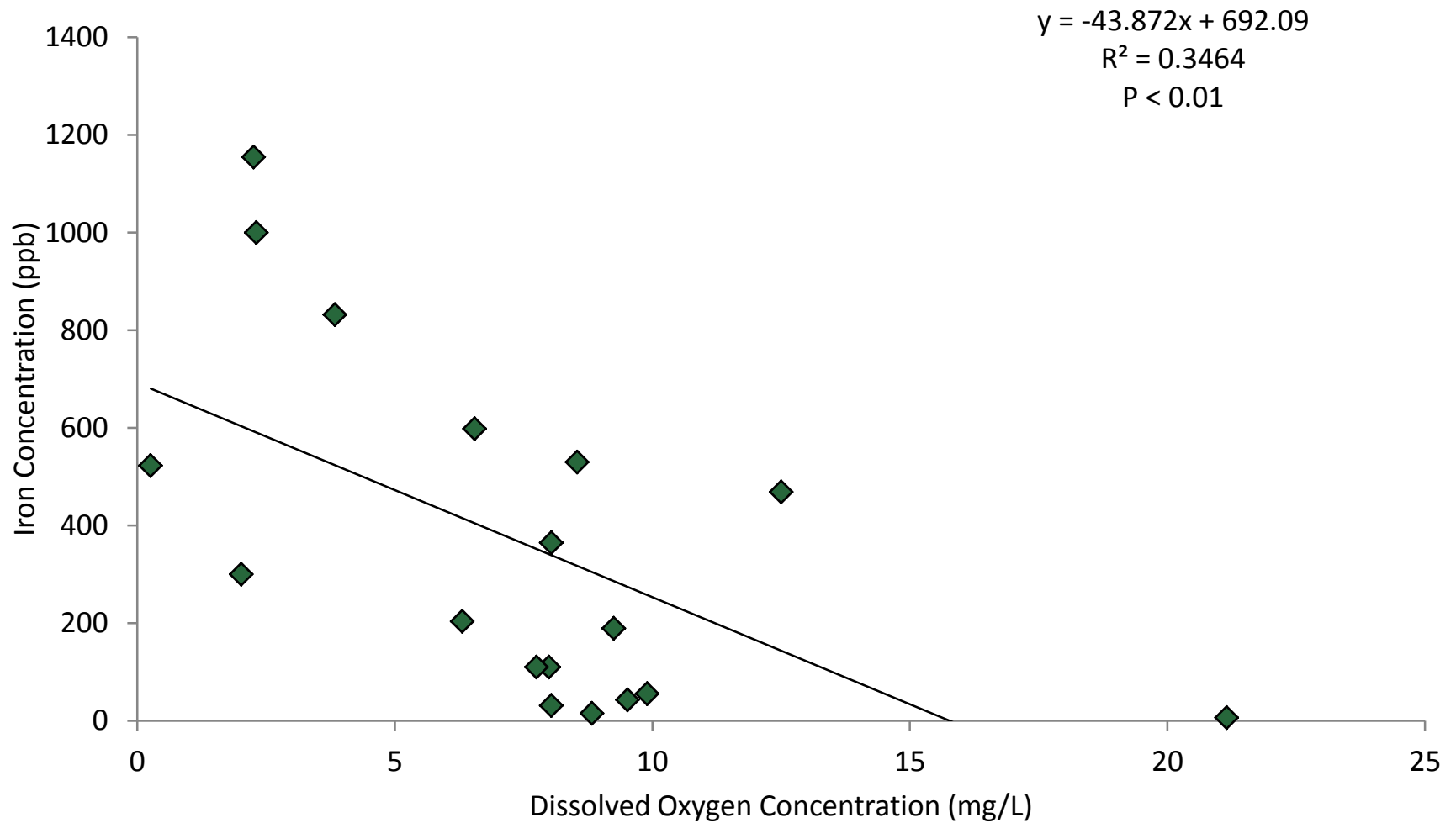
Iron by Site by Day



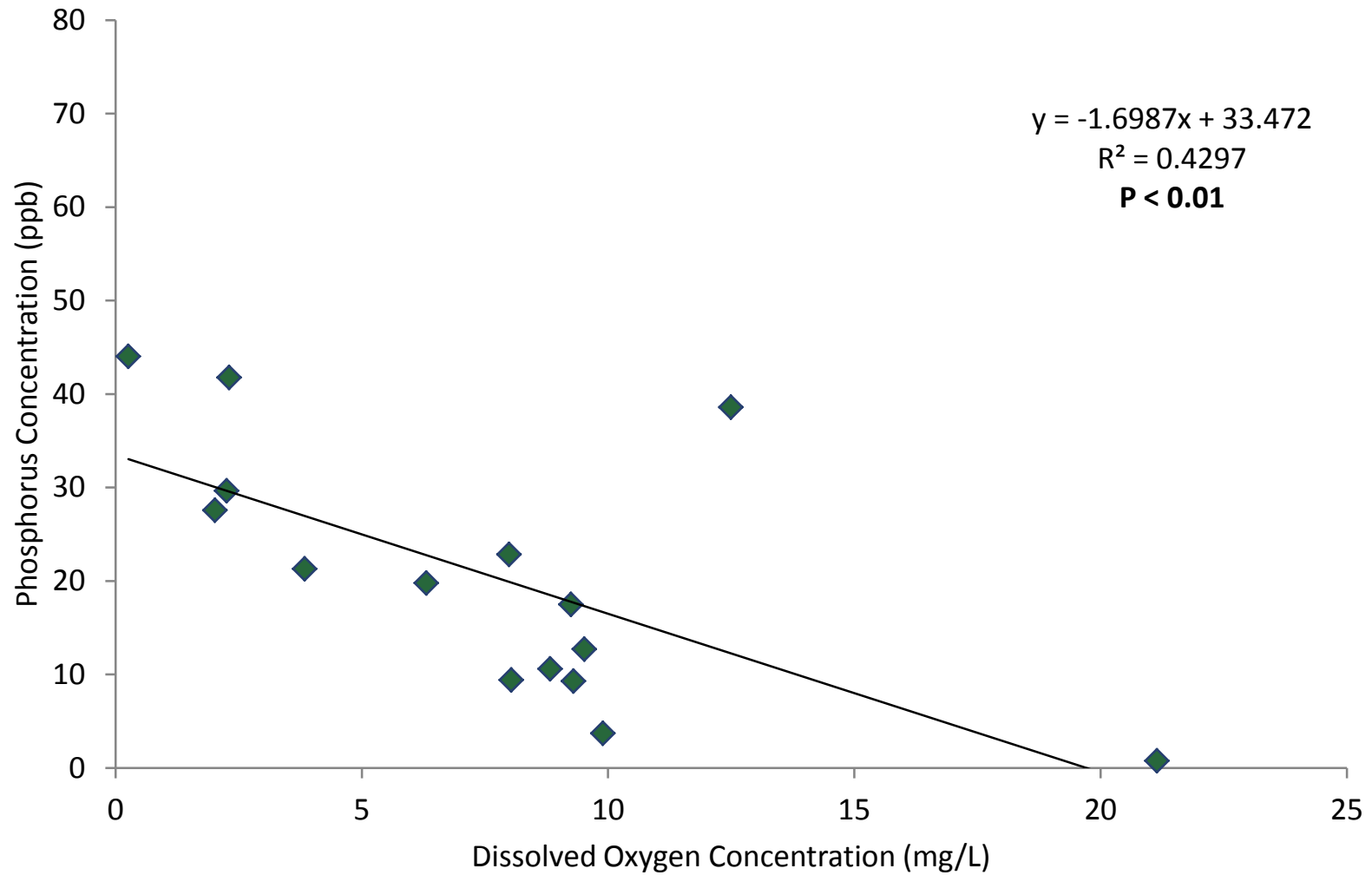
Iron by Site by Day



Iron vs. Dissolved Oxygen



Phosphorus vs. Dissolved Oxygen

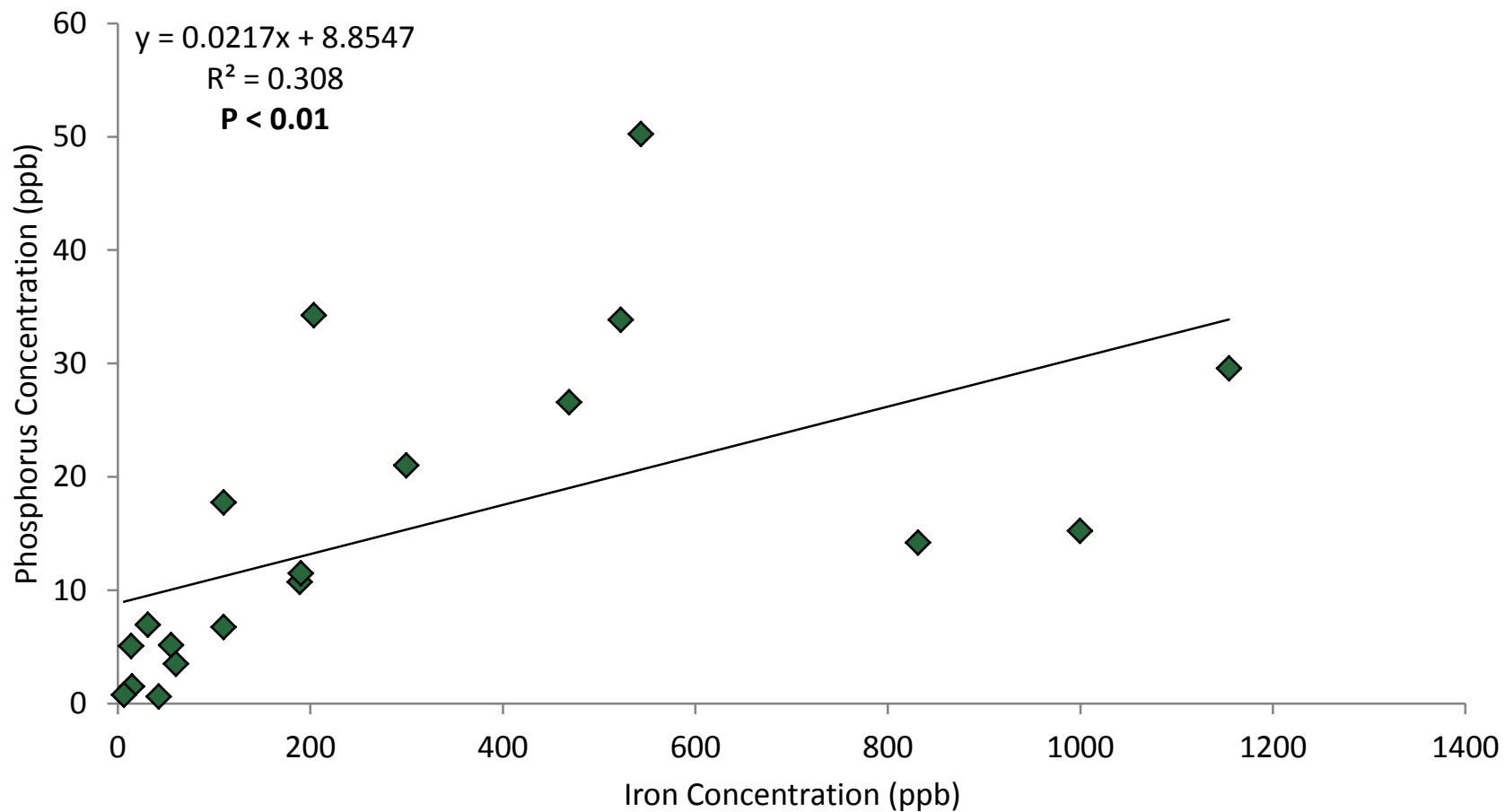


Phosphorus & the Metals

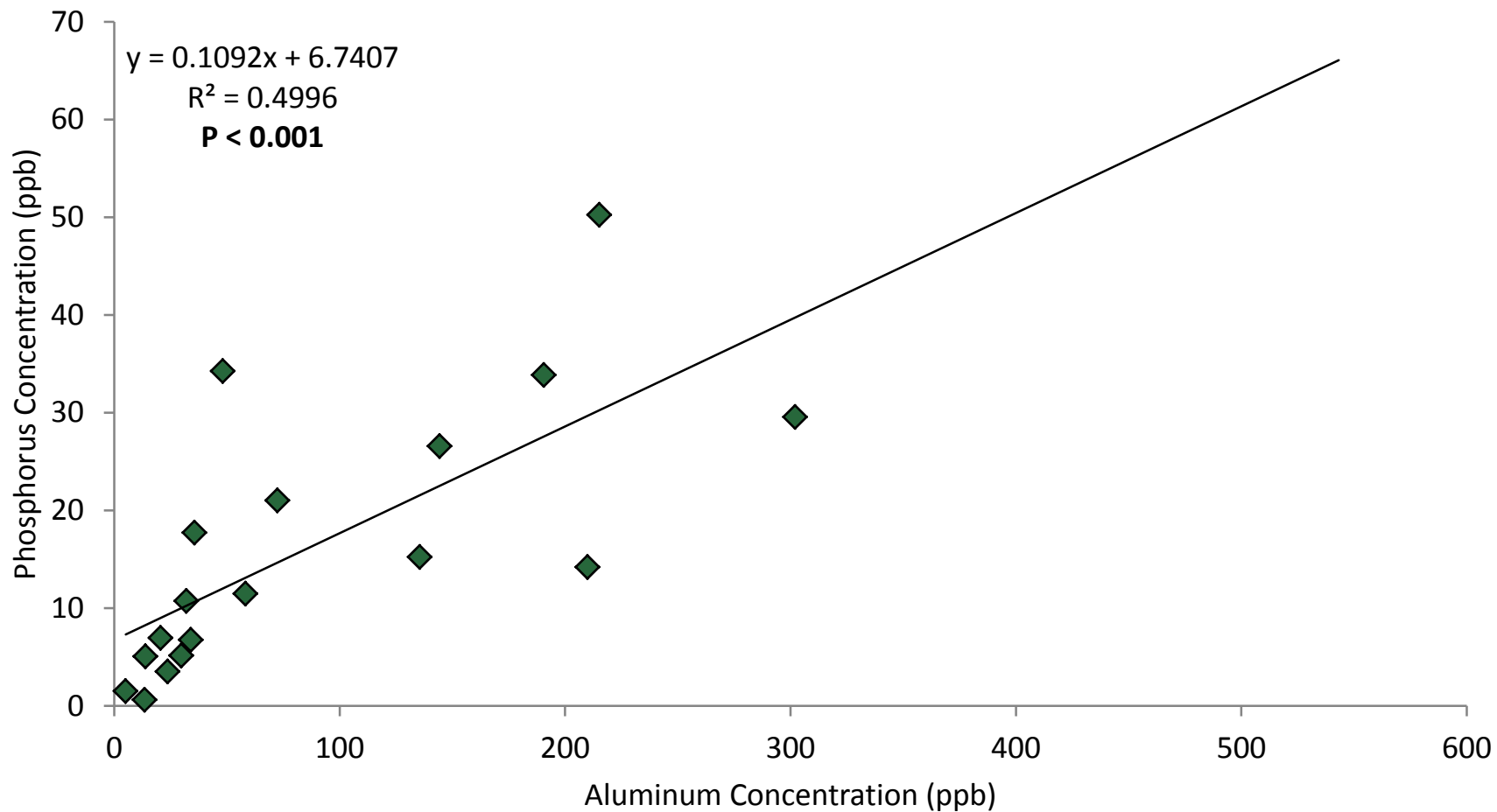
- As phosphorus binds to and is released by redox metals such as iron, we compared iron concentrations to phosphorus concentrations
- Conversely, aluminum binds phosphorus permanently
- Used linear regressions to determine correlation between metals and phosphorus



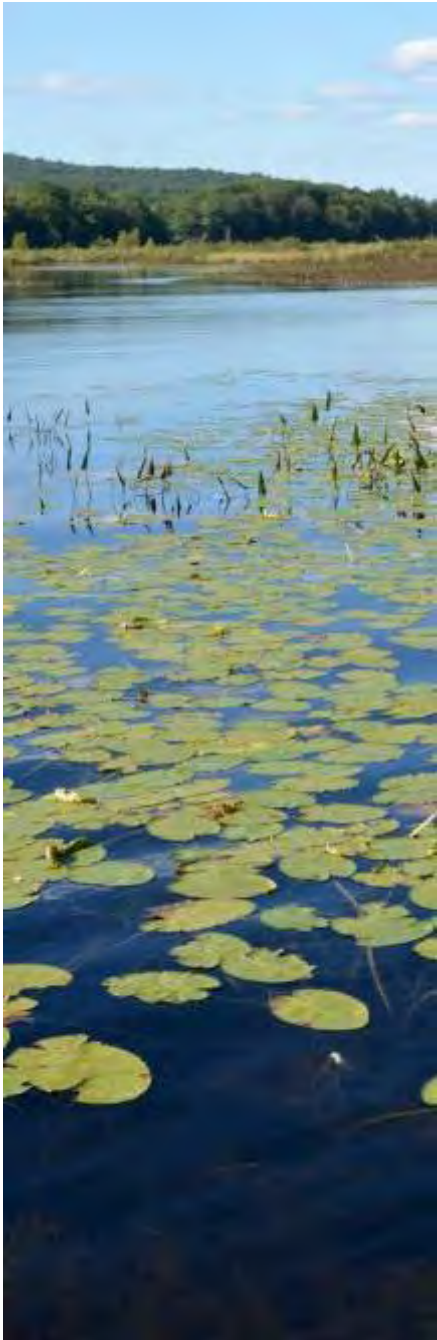
Iron vs. Phosphorus



Aluminum vs. Phosphorus



Take Home Messages



- The water chemistry in East Pond does not appear to be substantially influencing the water chemistry in the Serpentine beyond SC or North Pond
- The most substantial influx of nutrients from the input streams is seen after precipitation events
- Most nutrients seem to enter the Serpentine via the input streams
 - However, those nutrient levels are substantially diminished in North Pond
 - H1: biological uptake
 - H2: binding with Fe or Al
 - H3: dilution

Future Research

- Determine if other inputs contribute substantially to nutrient levels in North Pond
 - Importance of input streams?
- Further investigate three potential hypotheses
 - Does one process reduce nutrients more than others?
 - How might these processes change over time?
- Collect data over a longer time scale
- Determine the impacts of flow on nutrient levels





Algal Sampling

Why do algal blooms occur frequently in East Pond but not North Pond?

Our objectives:

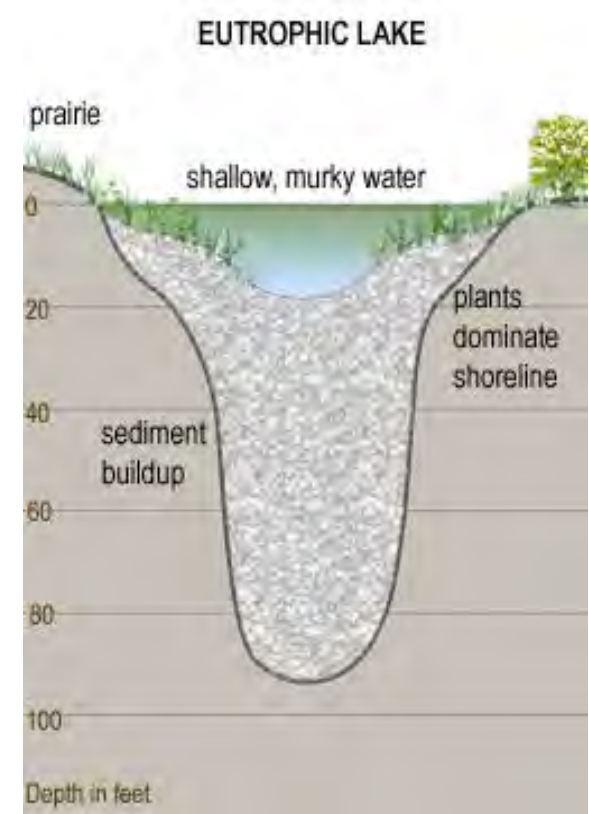
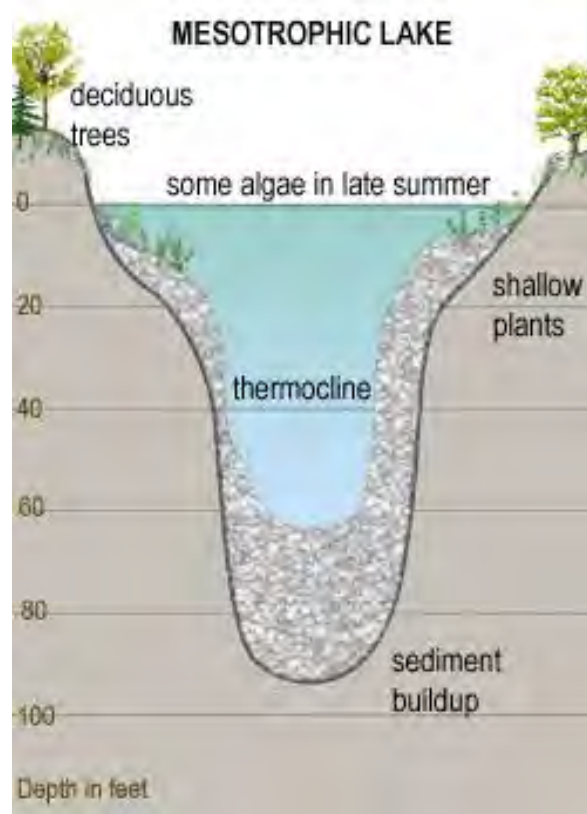
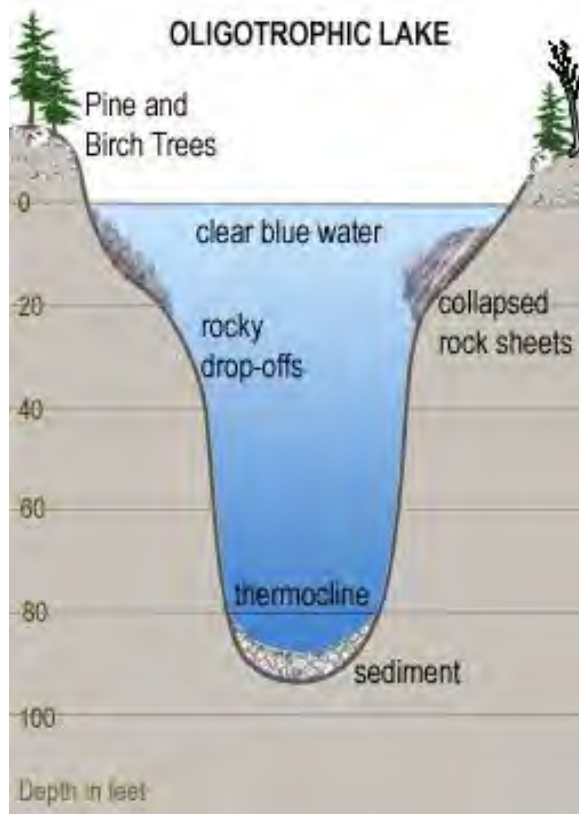
- Establish a baseline for algal species found in the Serpentine, East and North Ponds
- Record algae species that can be bioindicators
 - Can track change in the system over time

Algae as Bioindicators

- Different types of algae grow best in different conditions
 - Water nutrient levels (phosphorus, nitrogen)
 - Water temperatures
- Classified the algae by phylum



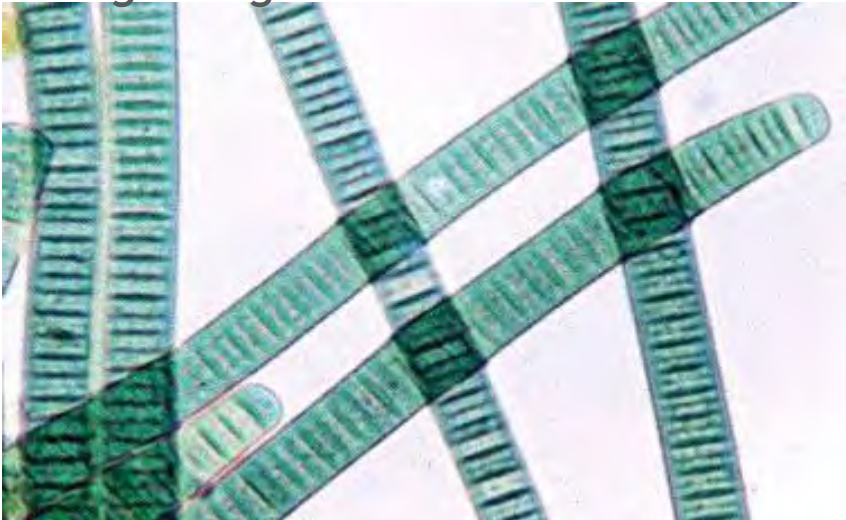
Review: Nutrient Loading



Algal Types (I)

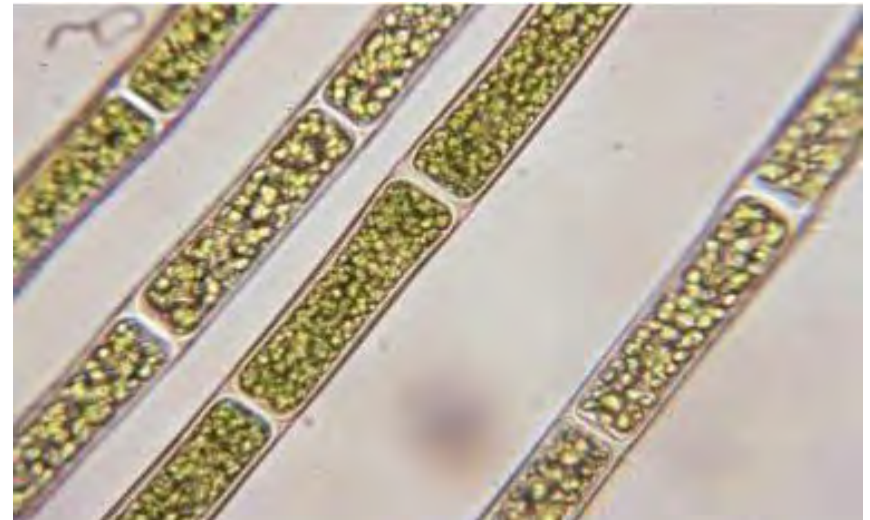
Cyanophyta

- Blue-green algae
- Eutrophic systems—high phosphorus
- Large blooms in late summer (warm water)
- Resistant to zooplankton grazing



Chlorophyta

- Green algae
- Mesotrophic and eutrophic environments
- Not dominant when phosphorus levels are high
- Early summer



Algal Types (II)

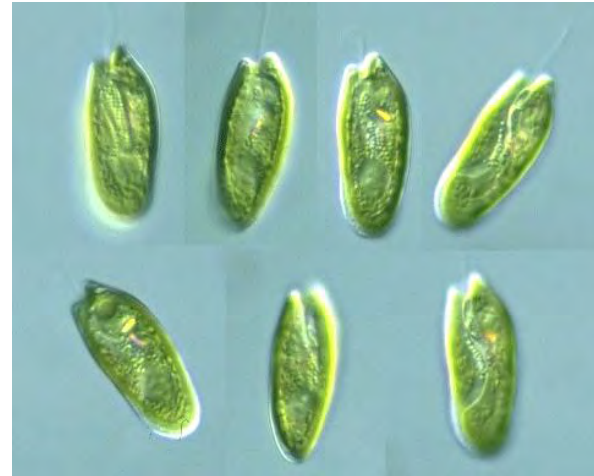
Bacillariophyta

- Diatoms
- Mesotrophic systems
- Mid-level phosphorus concentrations
- Bloom in spring and fall
- Cooler water temperature



Cryptophyta

- Oligotrophic and mesotrophic systems
- Cold water
- Spring blooms

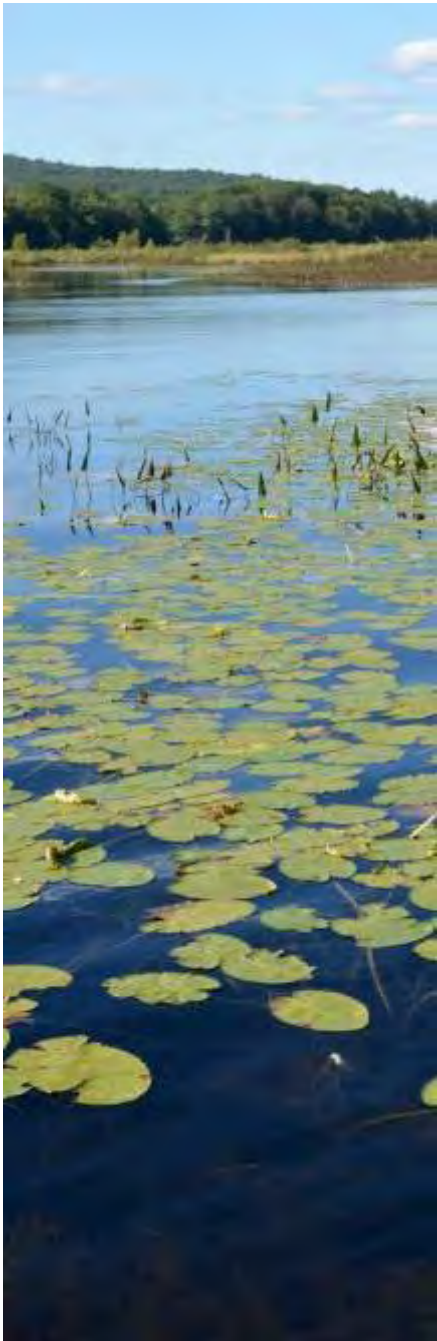


Methods

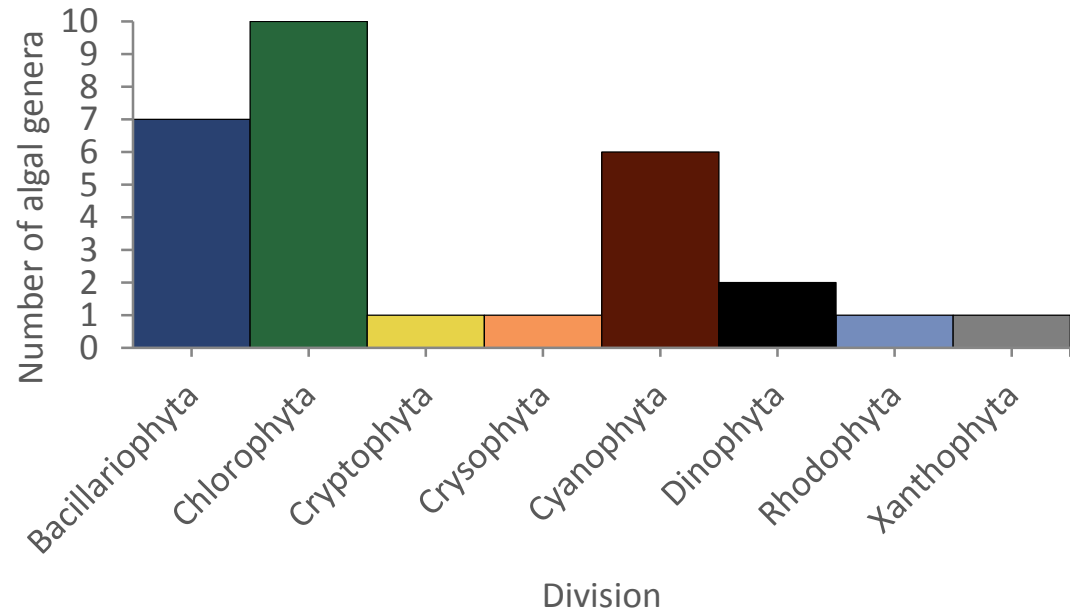


- 5 study sites
- One day of sampling (October 6th, 2011)
- Plankton tow net
- Preserved in ethanol
- Examined 5 slides from each sample through a microscope
- Identified each different specimen

Combined Results

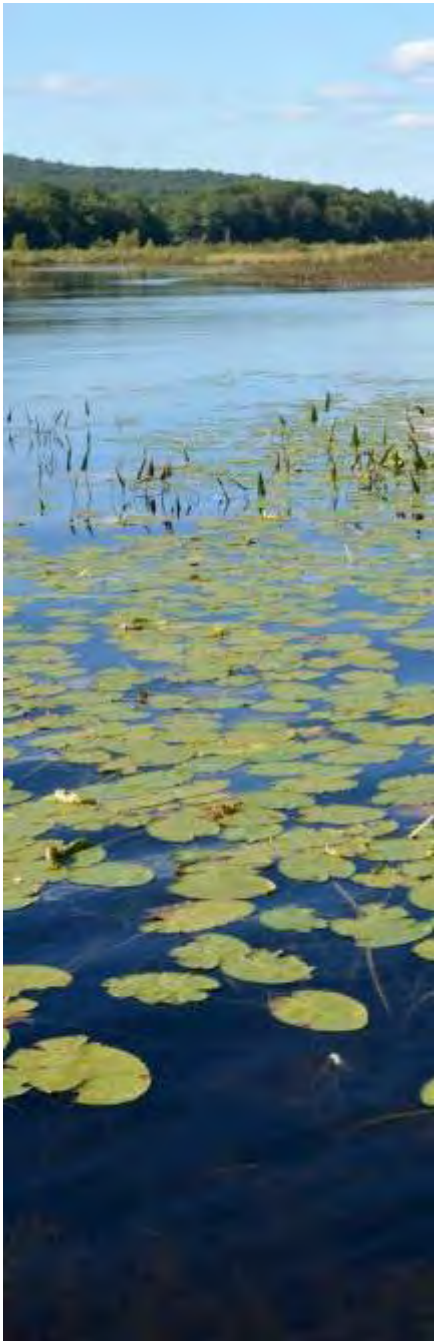


The number of algal genera per division (all sites)

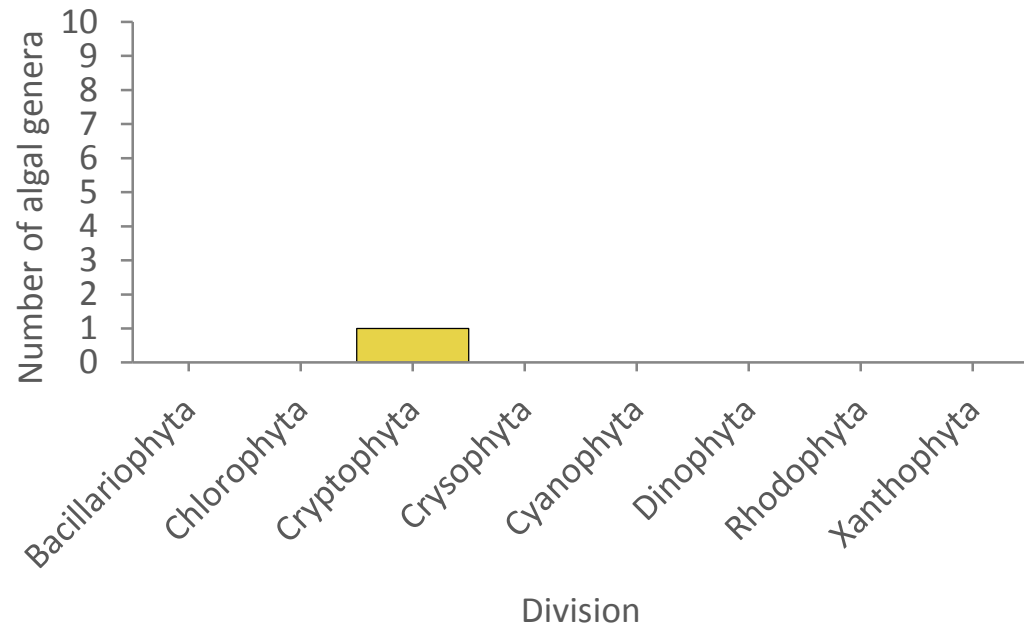


- 62 different genera in 8 phyla
- Chlorophyta, Bacillariophyta, and Cyanophyta were best represented

Serpentine Stream



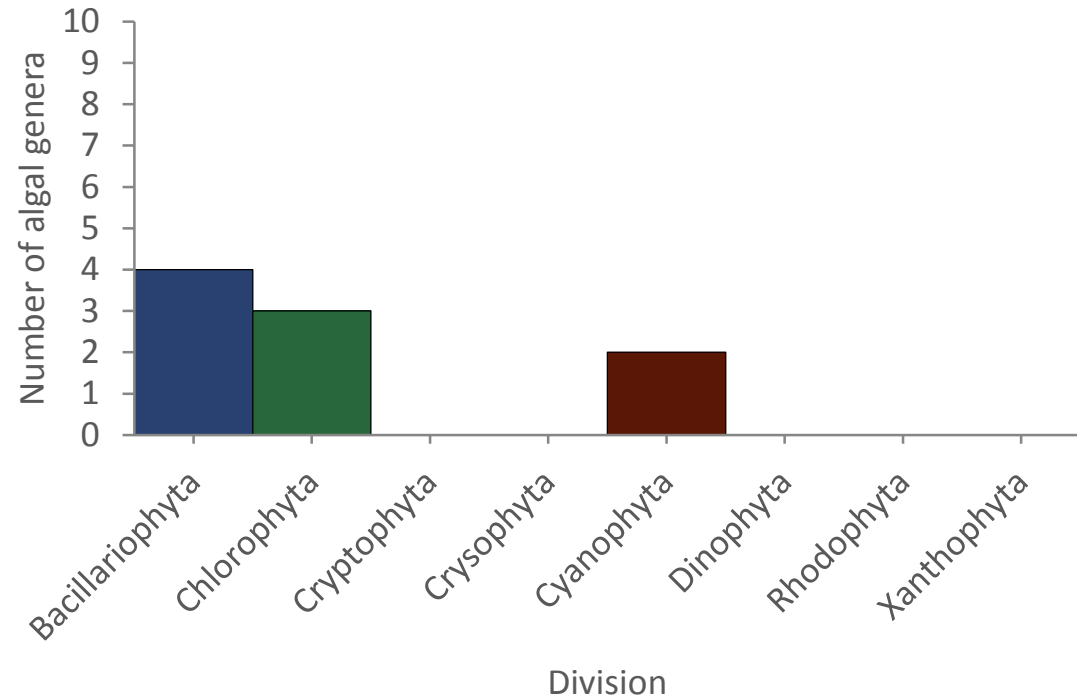
The number of algal genera per division (SC)



- Only one specimen spotted in five slides
- Cryptophyta—cold water species

Input Stream

The number of algal genera per division (12)

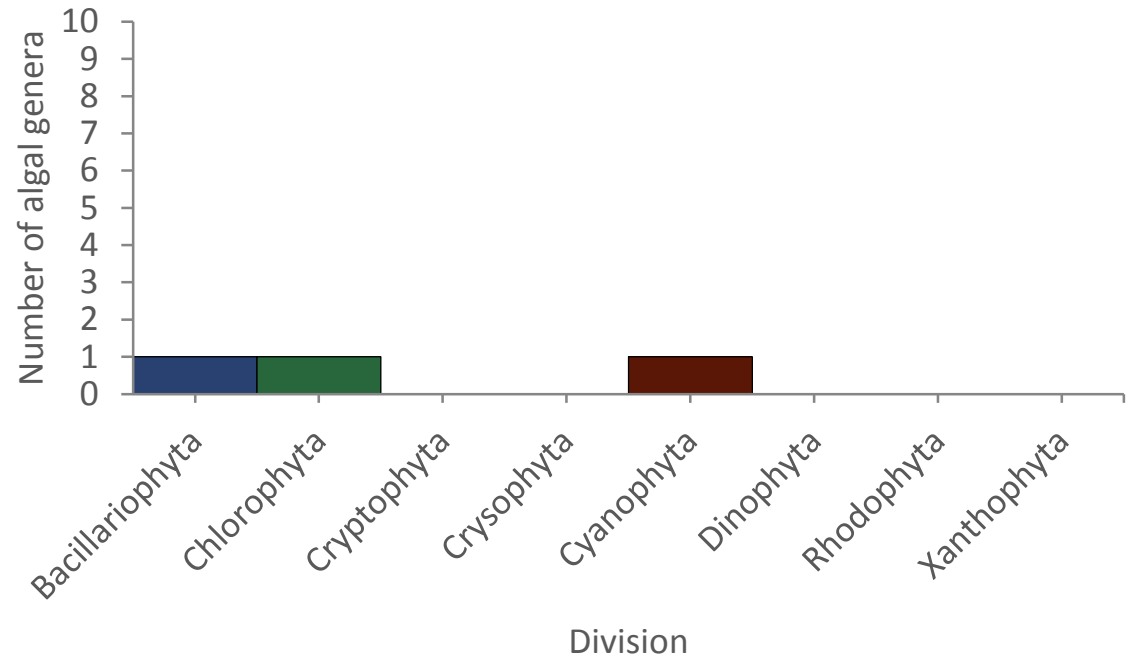


- Bacillariophyta, Chlorophyta, and Cyanophyta
- More Bacillariophytes as a result of cool temperatures?

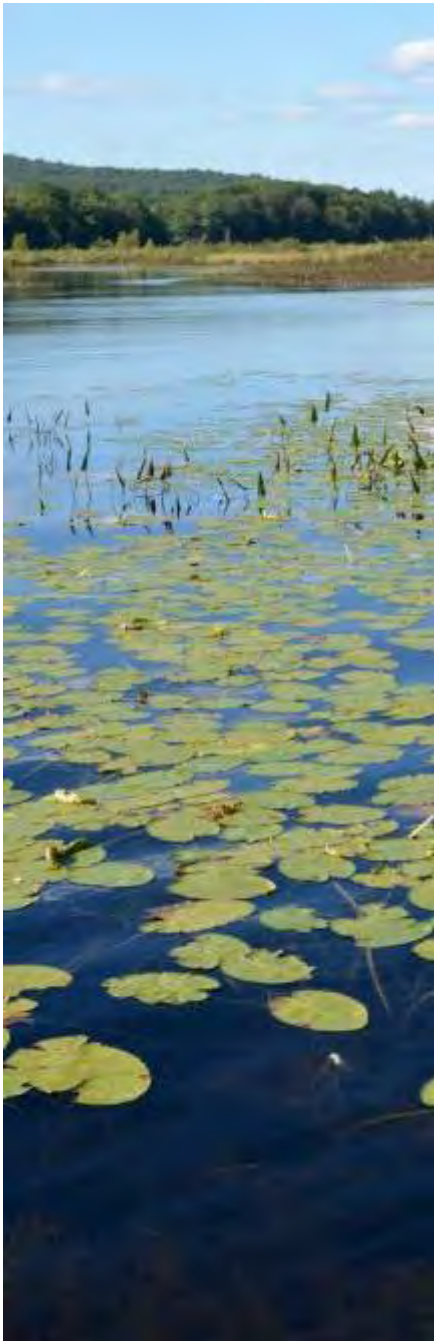


North Pond, Below Dam

The number of algal genera per division (BD)

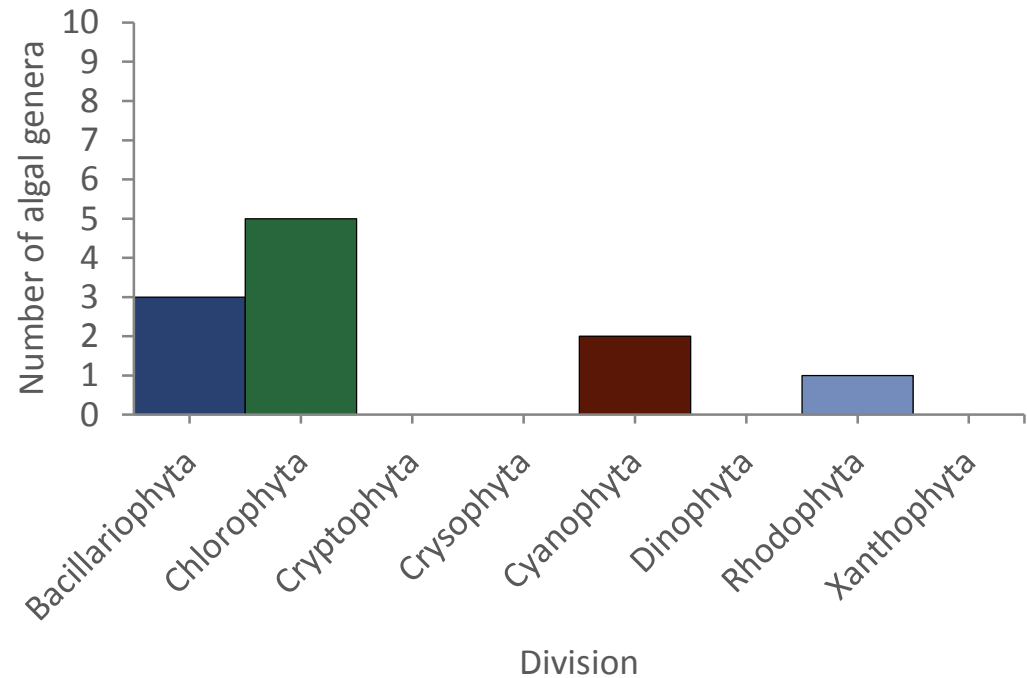


- Bacillariophyta, Chlorophyta, and Cyanophyta



North Pond

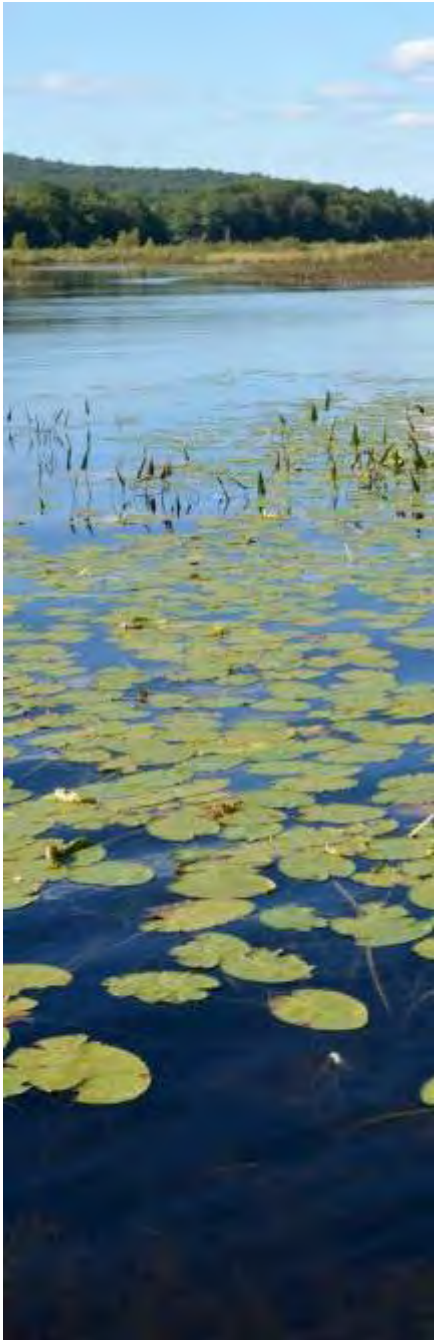
The number of algal genera per division (NP)



- Bacillariophyta, Chlorophyta, and Cyanophyta
- 11 genera in 14 phyla



Implications



- Chlorophyta, Bacillariophyta and Cyanophyta most represented
- Cold water species (Bacillariophyta, Cryptophyta)
- Can indicate:
 - A. System is mesotrophic
 - B. System is eutrophic, but cold water limits Cyanophytes

Future Research

- Limits of sampling method
 - Count abundance of the algal specimens
- Data collected over the whole ice-free season would better represent the system
 - Cyanophyta may dominate late summer algal blooms (warm water)
 - Biomanipulation not effective against Cyanophytes





Fish

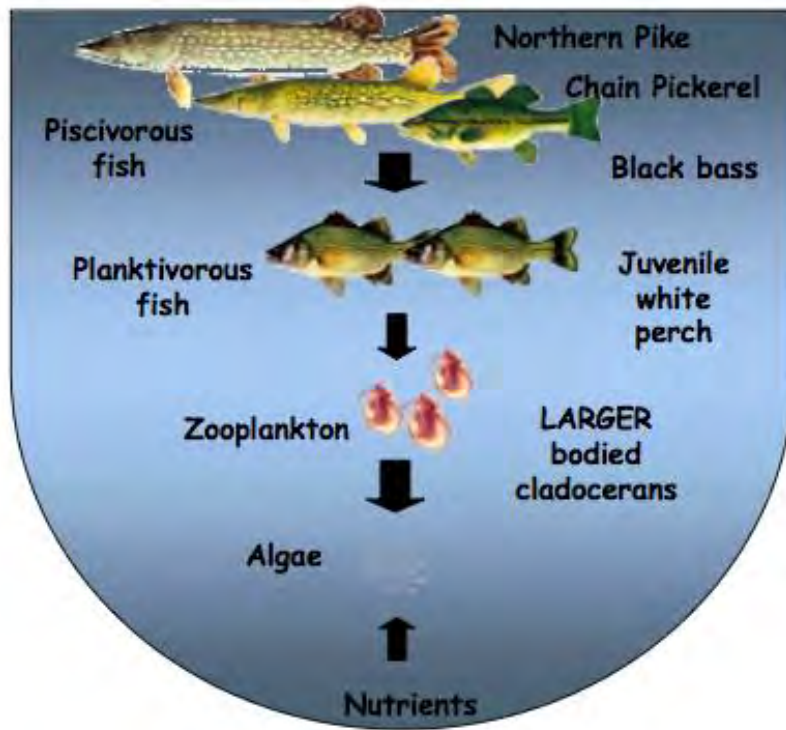
Our objectives:

- Determine which fish species may be present in the Serpentine
- Understand how these fish may be affecting trophic relationships in the Serpentine
- Understand the influence of fish in East pond algal blooms
- Develop potential explanations for the mixed results of the 2008 East pond biomanipulation project

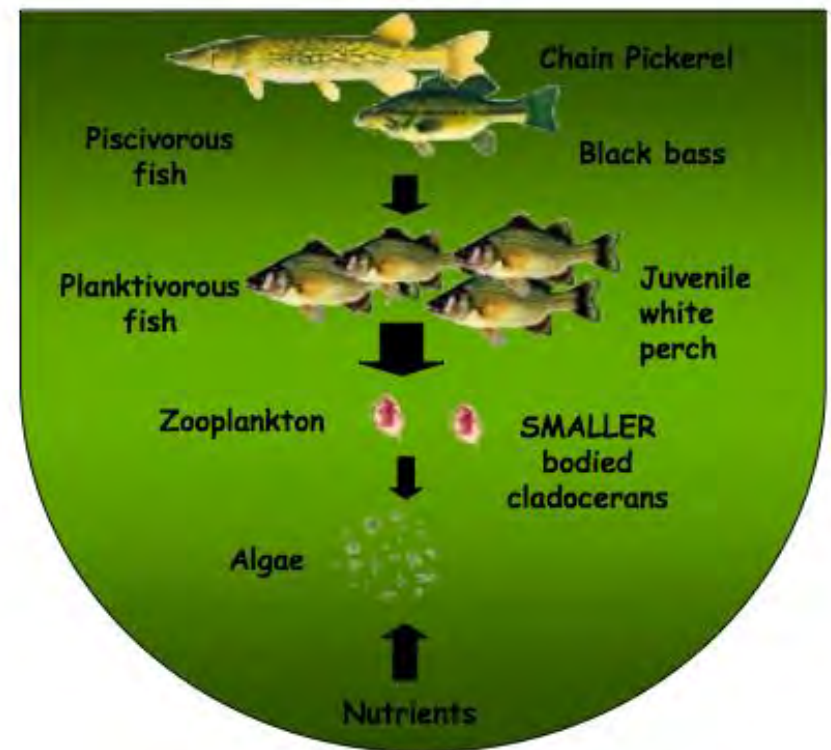
Ecological Role of Fish in Aquatic Ecosystems

Warmwater Shallow Lake Trophic Cascade

North Pond



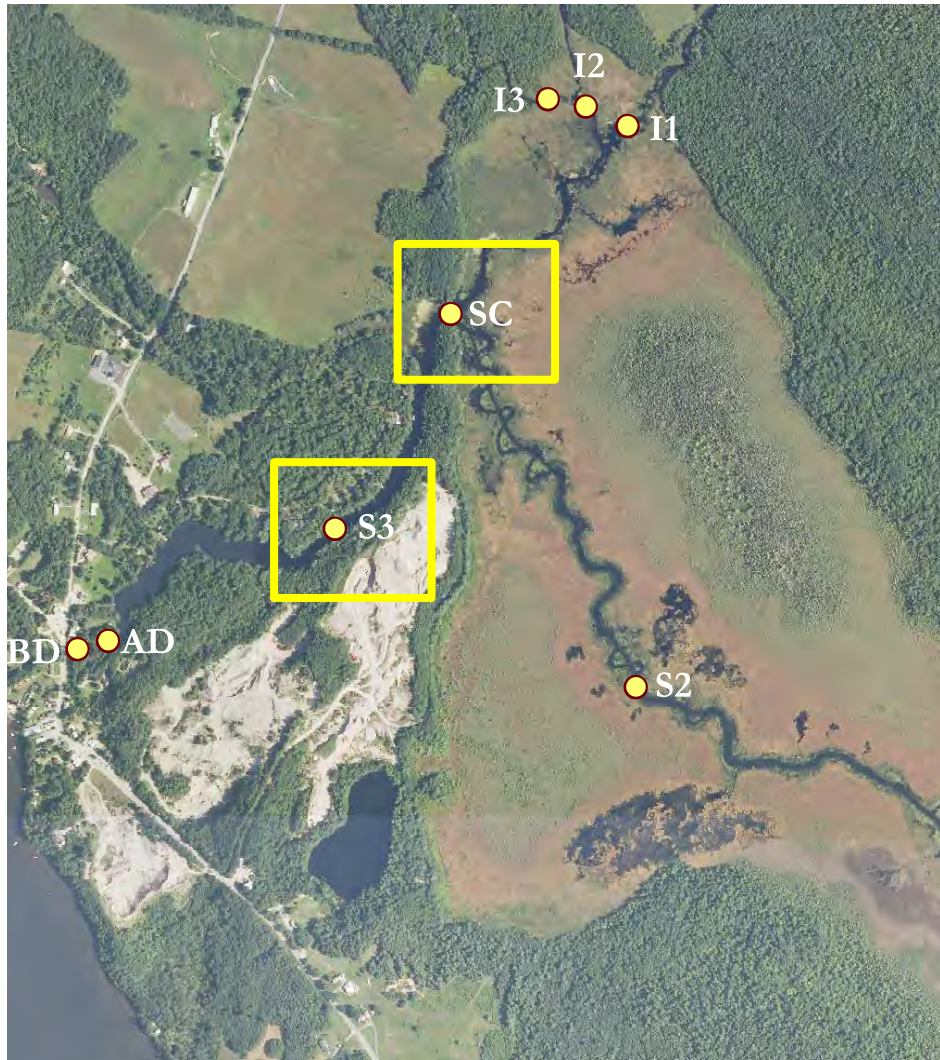
East Pond



Comparison of East & North Pond Fish Stocks

East Pond	Shared	North Pond
Rainbow Smelt	Brown Trout	Northern Pike
Black Crappie	Smallmouth Bass	Banded Killfish
	Largemouth Bass	
	White Perch	
	Yellow Perch	
	Chain Pickerel	
	Golden Shiner	
	White Sucker	
	Hornpout	
	Pumpkinseed Sunfish	

Field Methods



- Angling survey
 - Catch per unit effort
- Diet analysis
 - Species identification
- Fisherman Survey
 - Presence/absence

Results & Discussion

Present	Absent
Yellow Perch	Brown Trout
White Perch	Rainbow Smelt
Chain Pickerel	
Smallmouth Bass	
Largemouth Bass	
Black Crappie	
Bullhead	
White Sucker	
Pumpkinseed Sunfish	



Results & Discussion



- Presence and absence conclusions supported by interviews with fishermen and catch from angling
- Yellow perch dominance and trophic implications

Implications for 2008 East Pond Biomanipulation Project

- Future biomanipulation projects should take into account the Serpentine's potential as a refuge for zooplanktivorous fish



Future Study

- Alternative survey techniques
 - Seine netting
 - Electrofishing
- Full year survey
 - Spawning season





Plants

Our objectives:

- Determine species composition and distribution along length of Serpentine
- Classify types of habitats along Serpentine
- Investigate interactions of plants and sediment/water chemistry

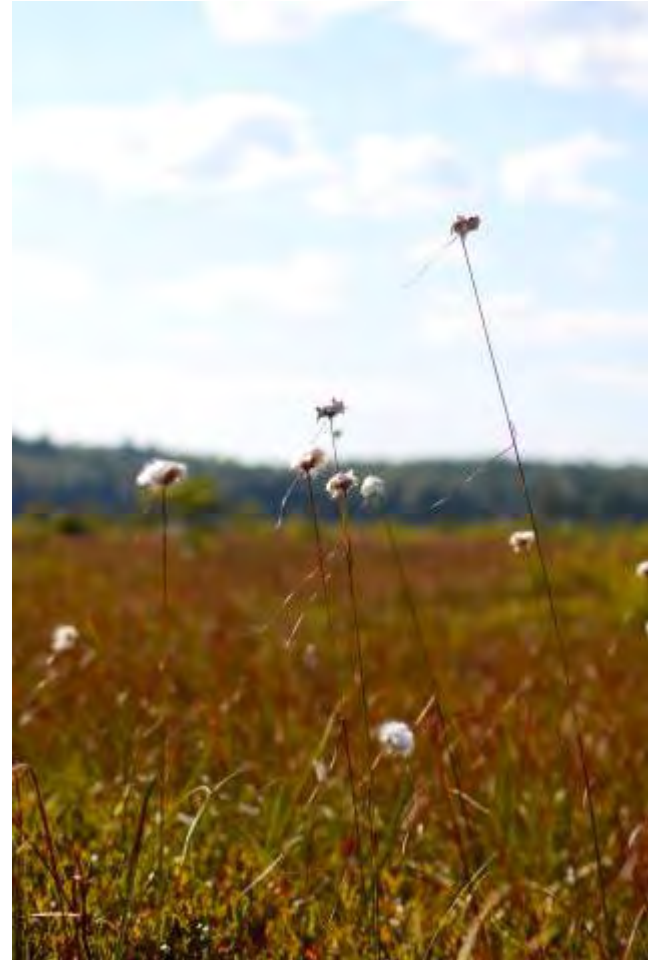
Watershed and Sample Sites



Study Area

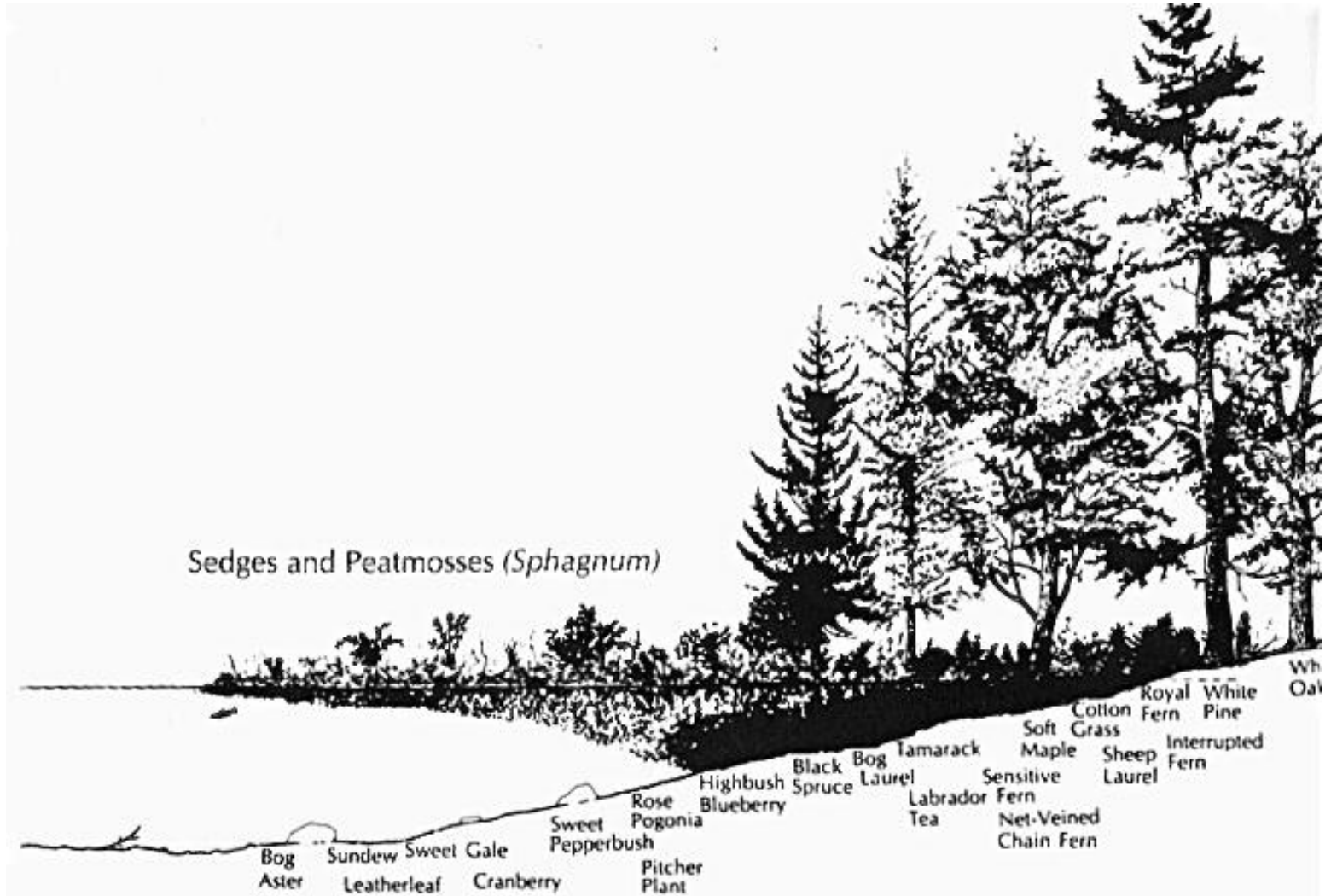


Forest



Wetland

Fen Zonation

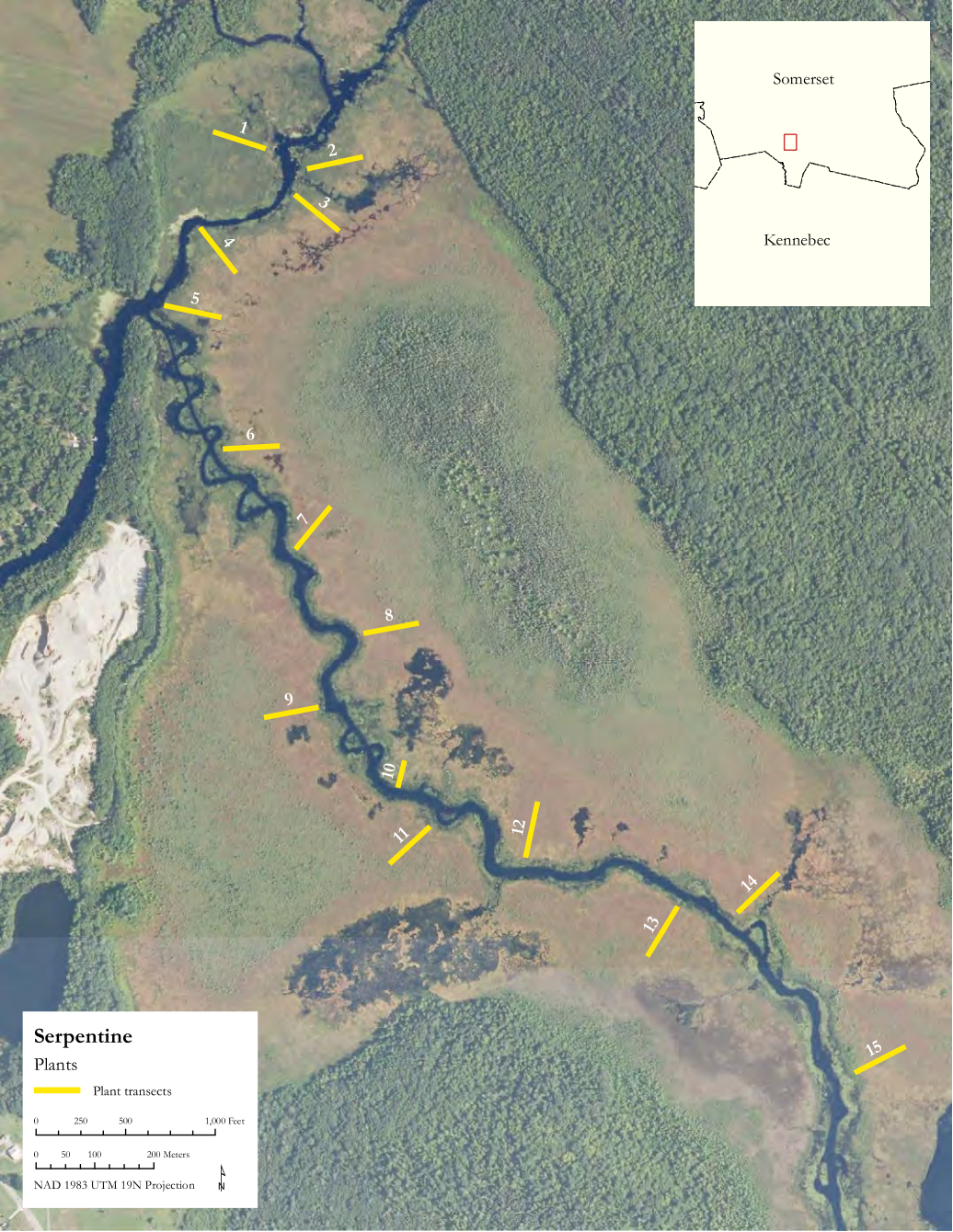


Methods

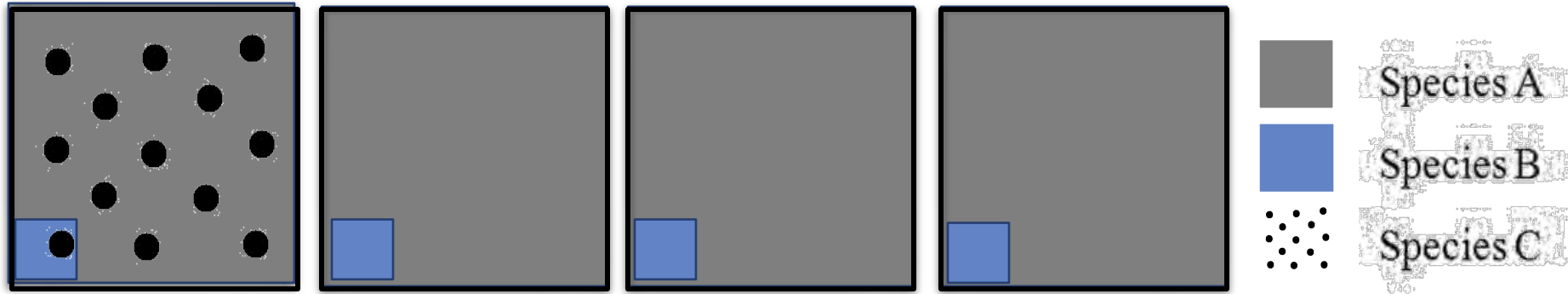
15 transects -- 9 quadrats each

0m • 5m • 10m • 15m • 20m • 25m • 50m • 75m • 100m





Methods: Plant Importance Index



Site 1

A 100%
B 4%
C 100%

Site 2

A 100%
B 4%
C 0%

Site 3

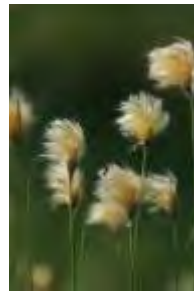
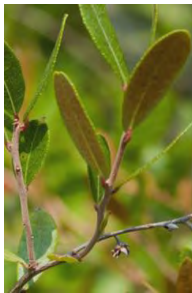
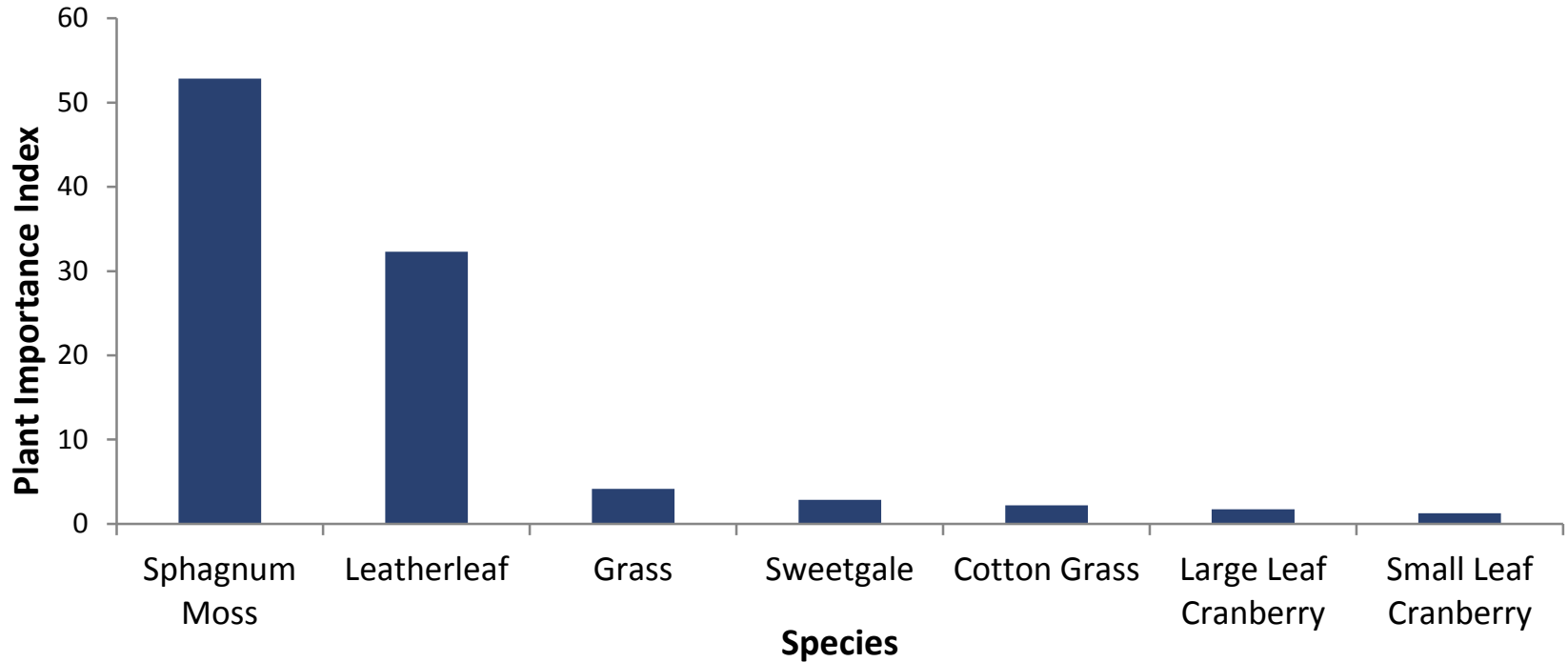
A 100%
B 4%
C 0%

Site 4

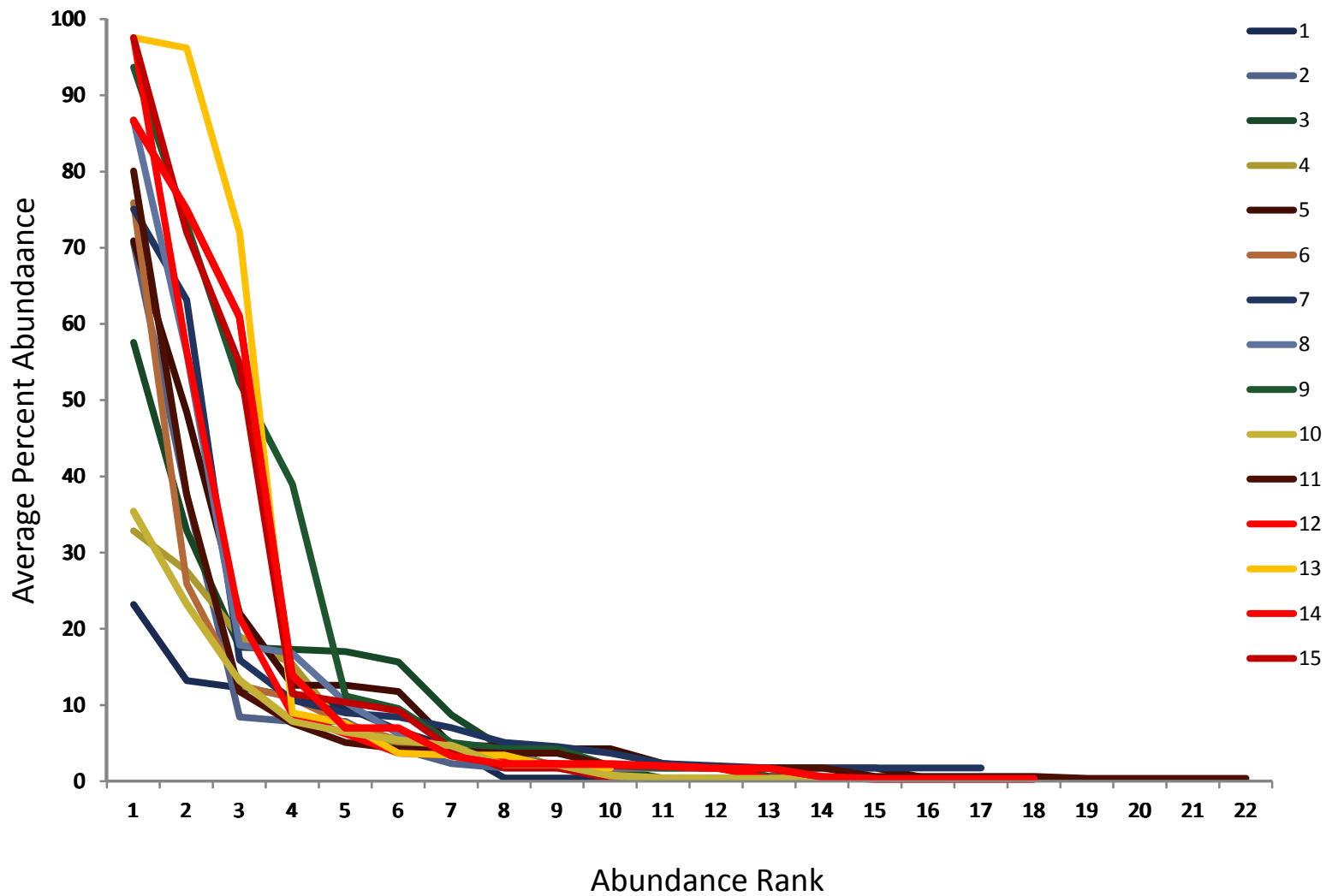
A 100%
B 4%
C 0%

Species	Normalized Cover	Normalized Frequency	Plant Importance Index
A	0.7752	1	0.7752
B	0.0310	1	0.0310
C	0.1938	25	0.0485

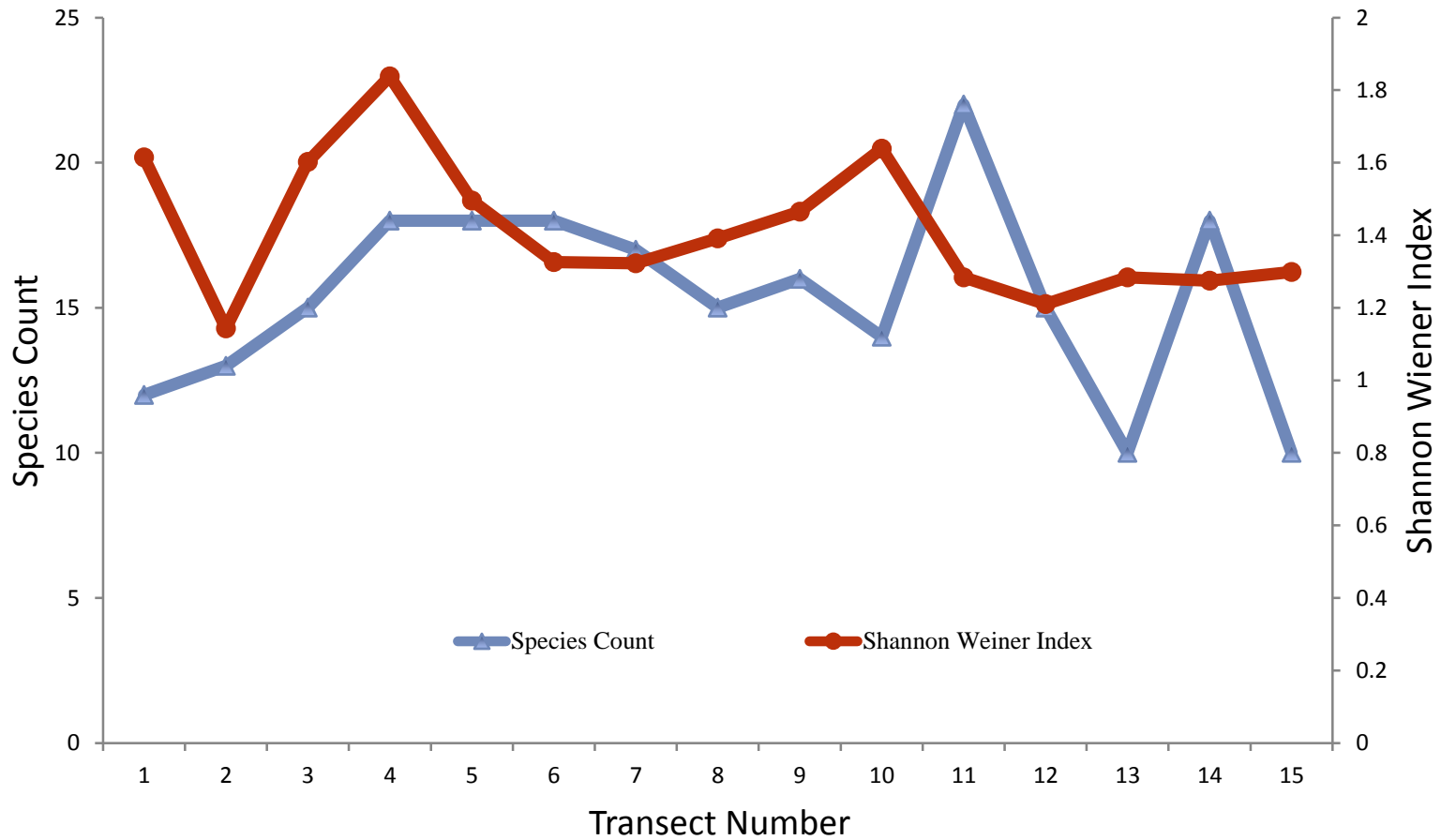
Results: Plant Importance Index



Results: Rank Abundance



Results: Shannon Weiner Index



Inputs → *East Pond*

Conclusions: Sphagnum Moss

- Sphagnol
 - Build up of peat
 - Low decomposition
- Peat as a carbon sink



Conclusions

- Input of nutrients lowers species diversity from inputs into serpentine
- Few dominant species, many uncommon species (typical for fens)
- Importance of biodiversity



Future Research

- Peat as a nutrient sink?

- Water flow through fen (cores)



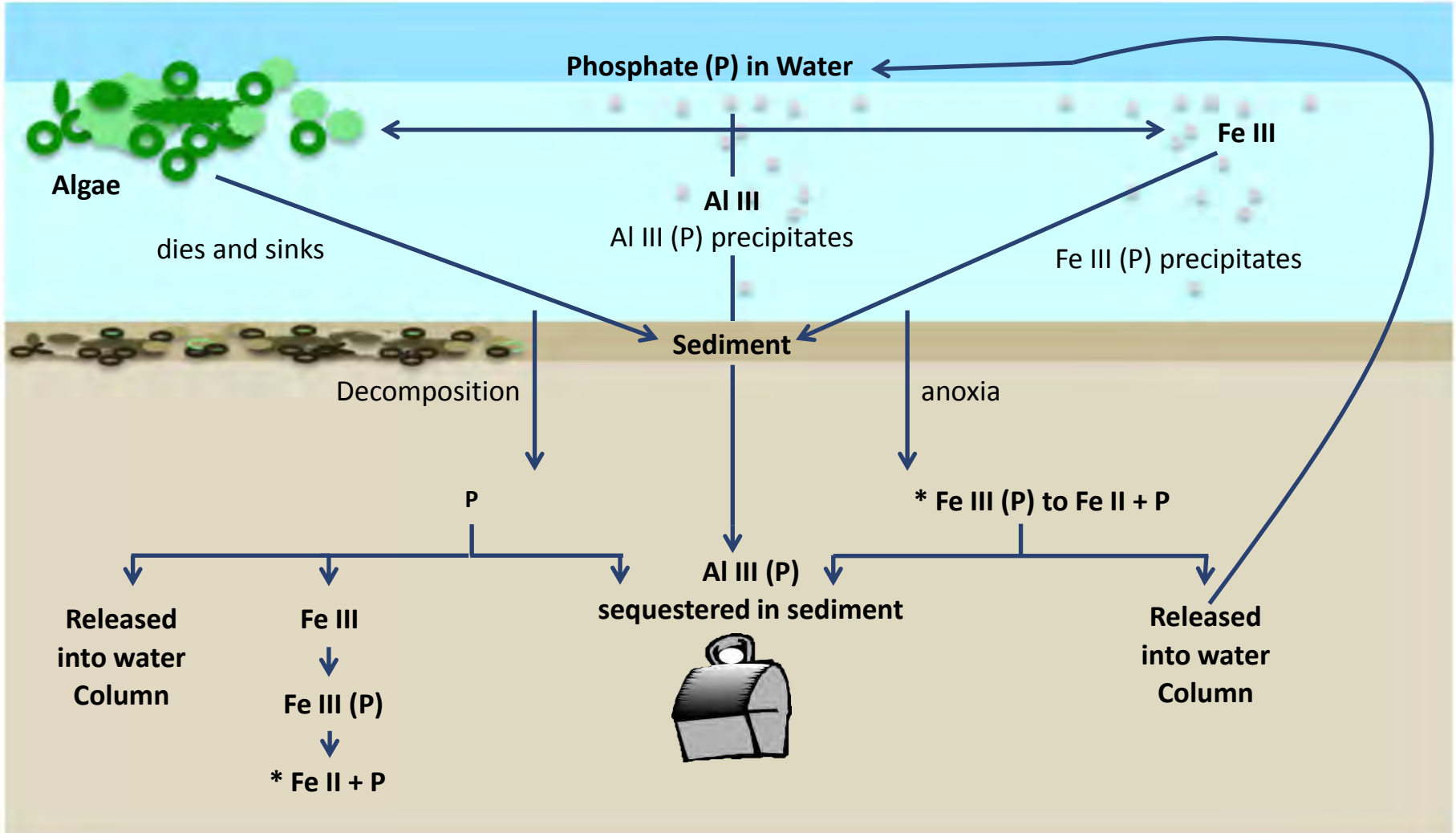


Sediment Chemistry

Objective:

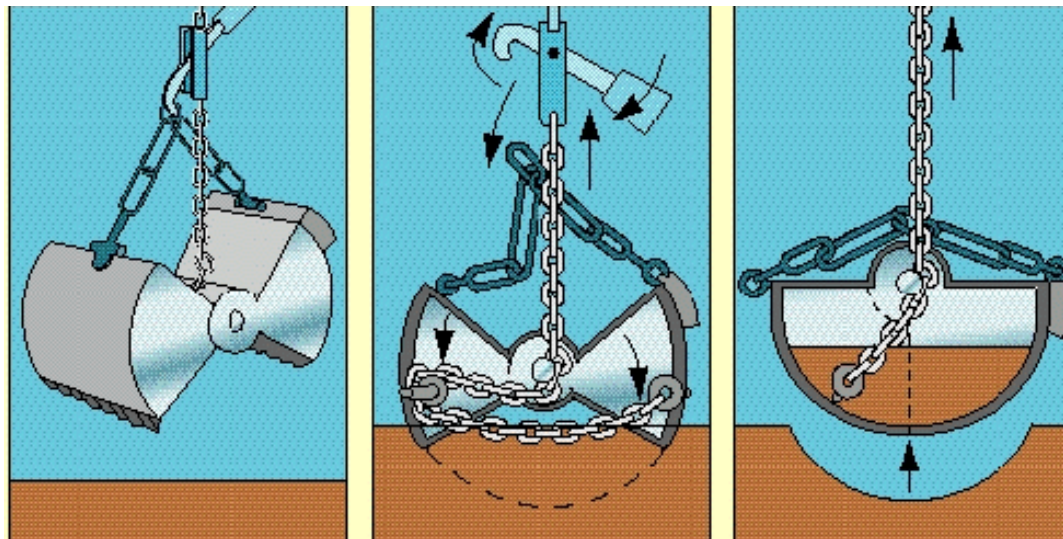
To understand the nutrient cycling in the Serpentine by measuring P, Al, and Fe in the sediment.

The Phosphorous Cycle



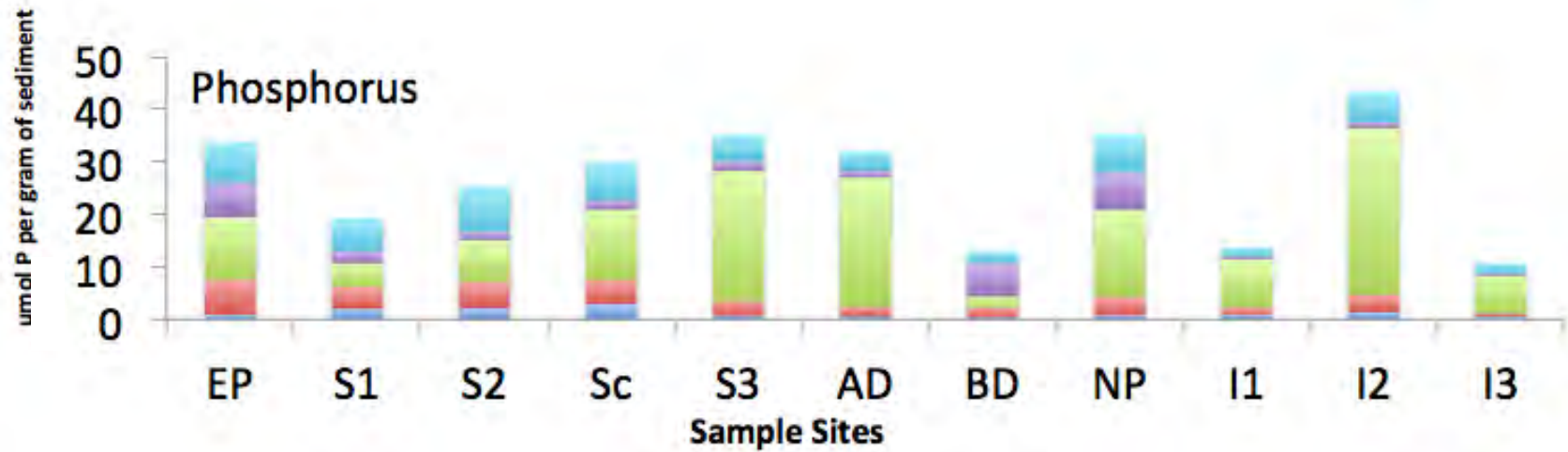
Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal bloom

Methods



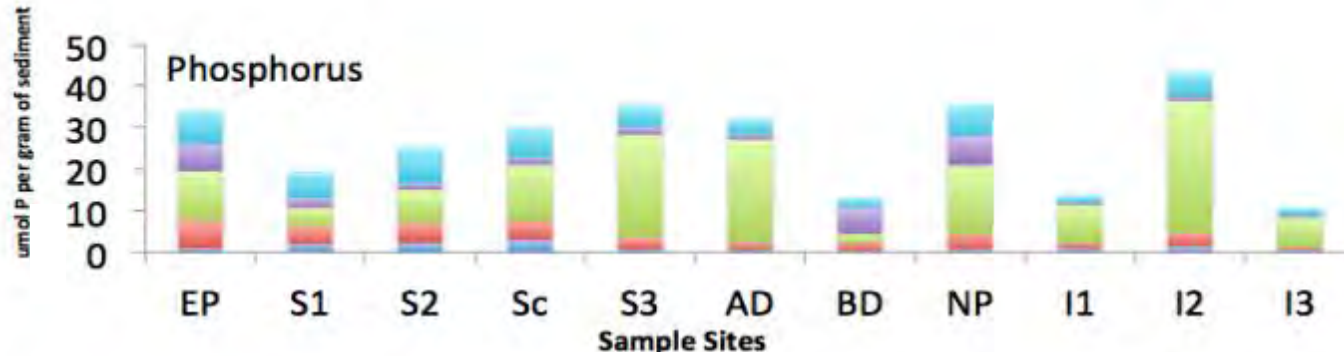
Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Sequential Extraction



Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Sequential Extraction



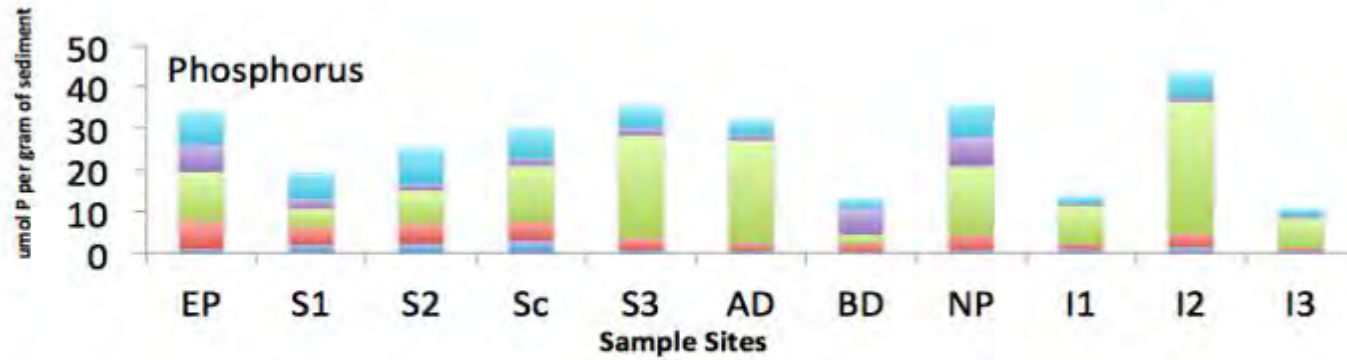
Dissolved in WATER

Step 1



Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Sequential Extraction



Dissolved in WATER

Step 1



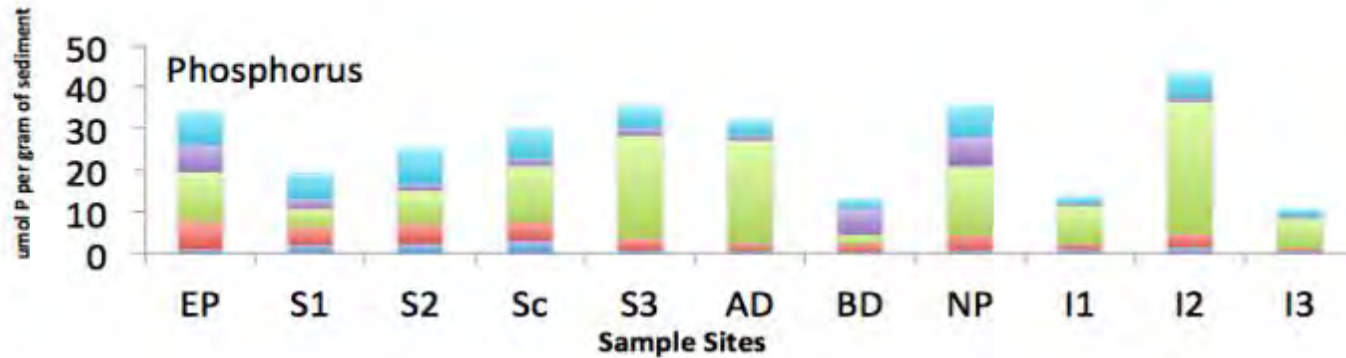
Released during ANOXIC conditions

Step 2



Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Sequential Extraction



Dissolved in WATER

Step 1



Released during ANOXIC conditions

Step 2



Released though decomposition of Organic Matter OR Sequestered in Aluminum

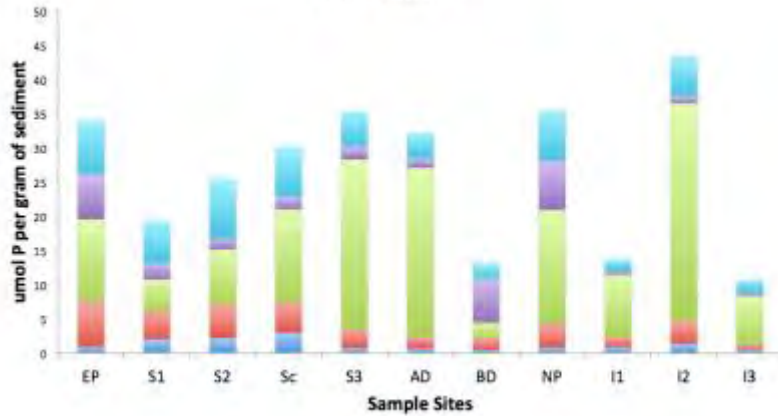
Step 3



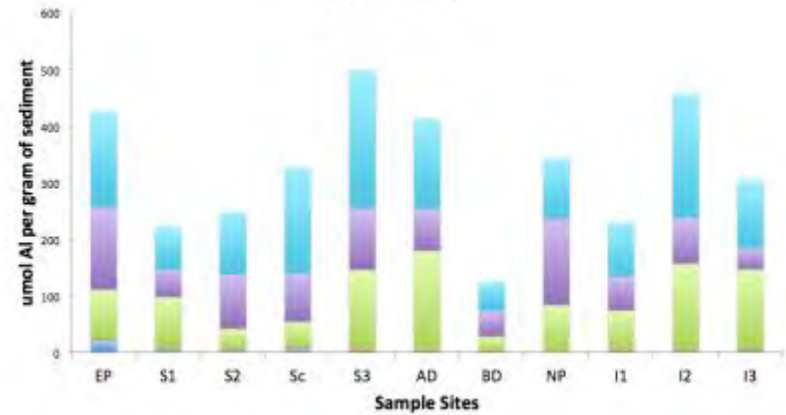
Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Results

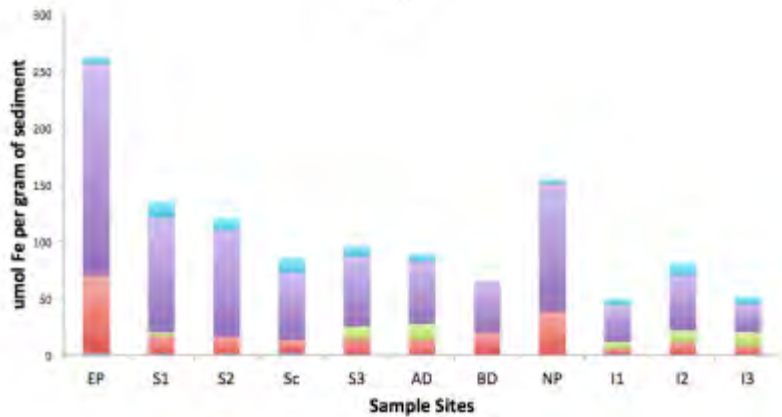
Phosphorus



Aluminum



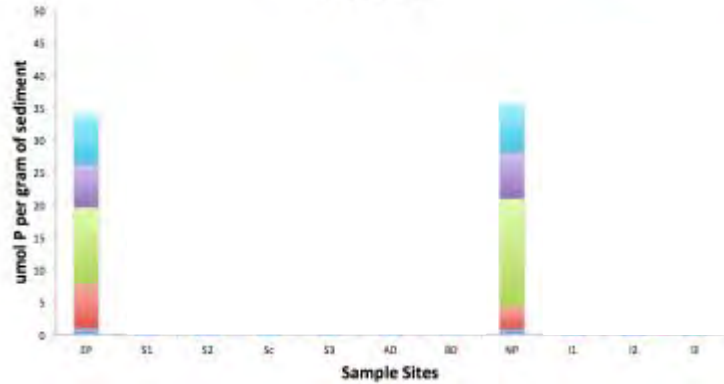
Iron



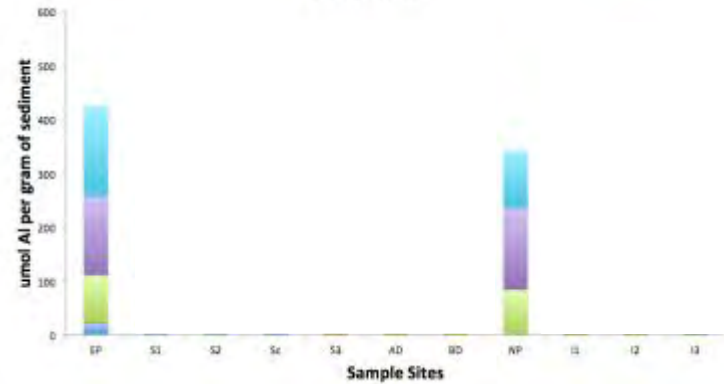
Step 5 Step 4 Step 3 Step 2 Step 1

East Pond vs North Pond

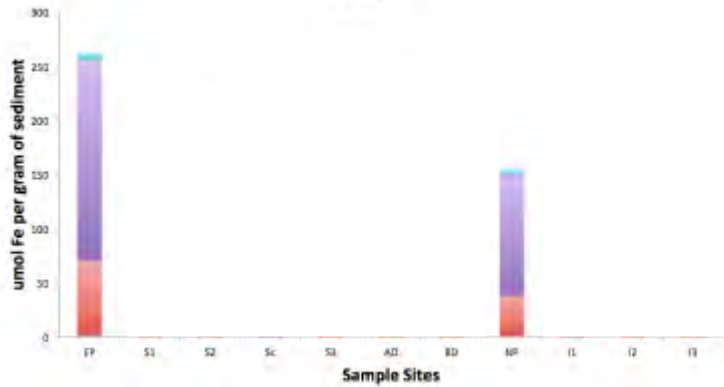
Phosphorus



Aluminum

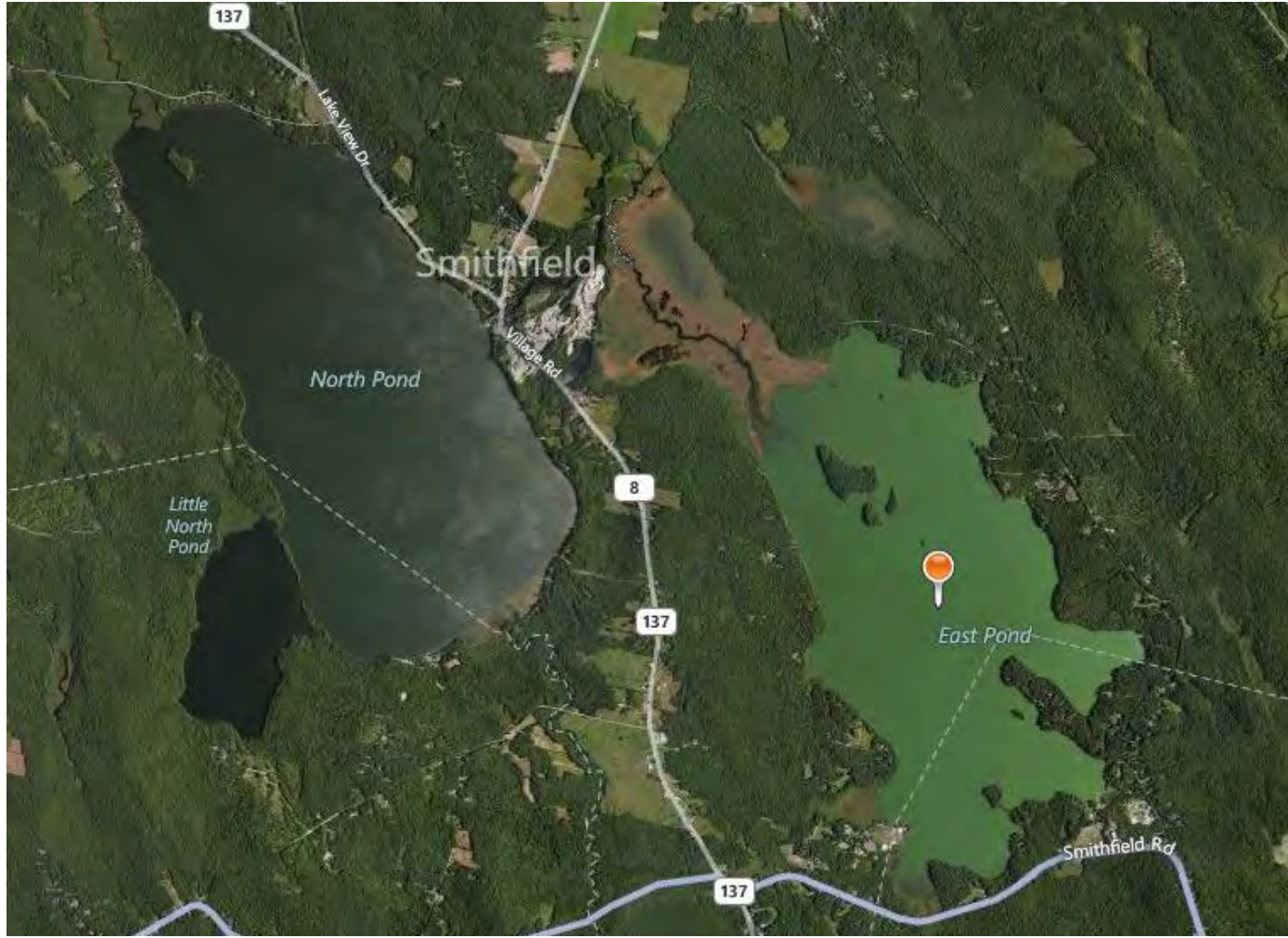


Iron



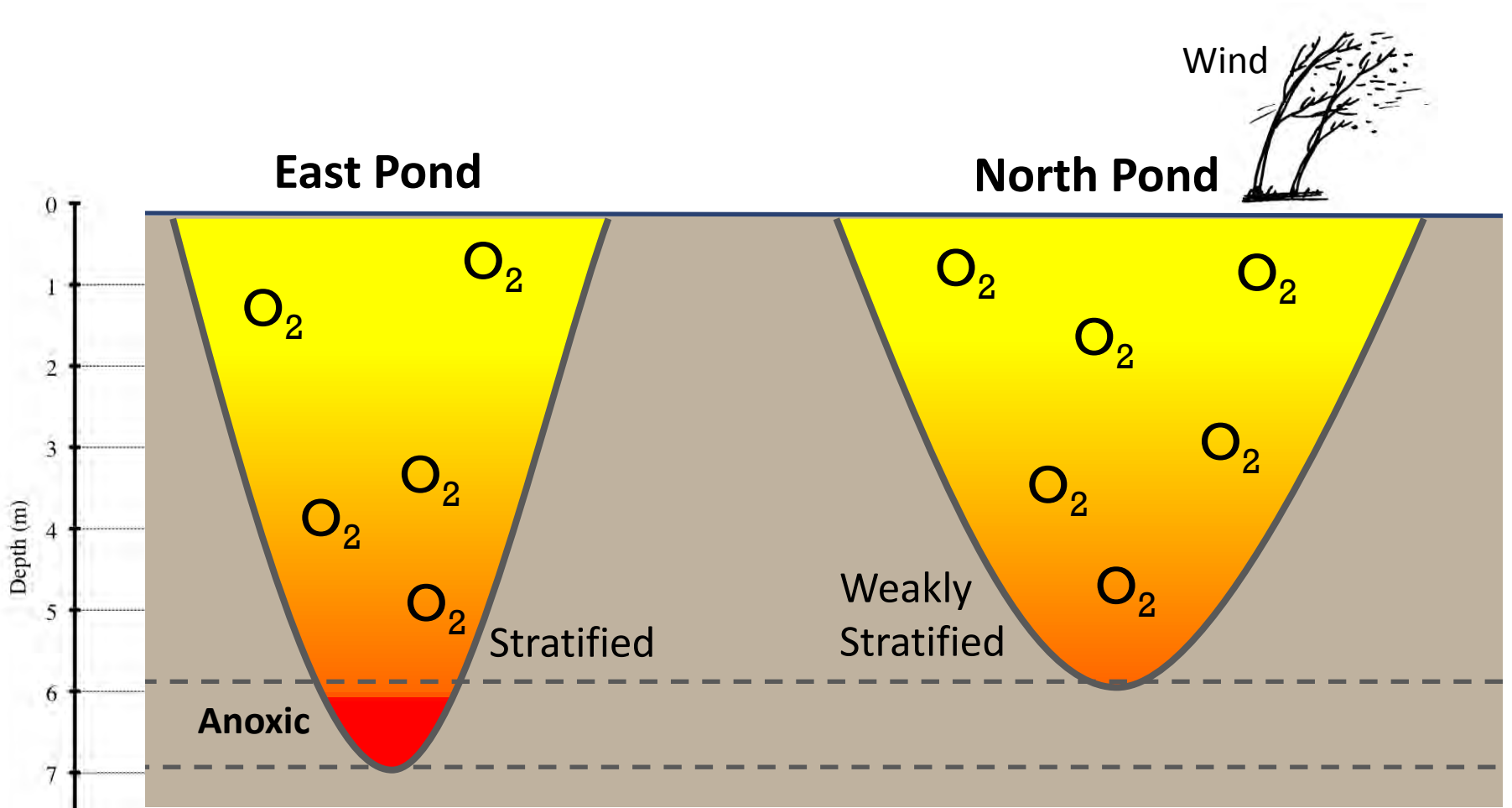
Step 5 Step 4 Step 3 Step 2 Step 1

Why does East Pond bloom?



Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

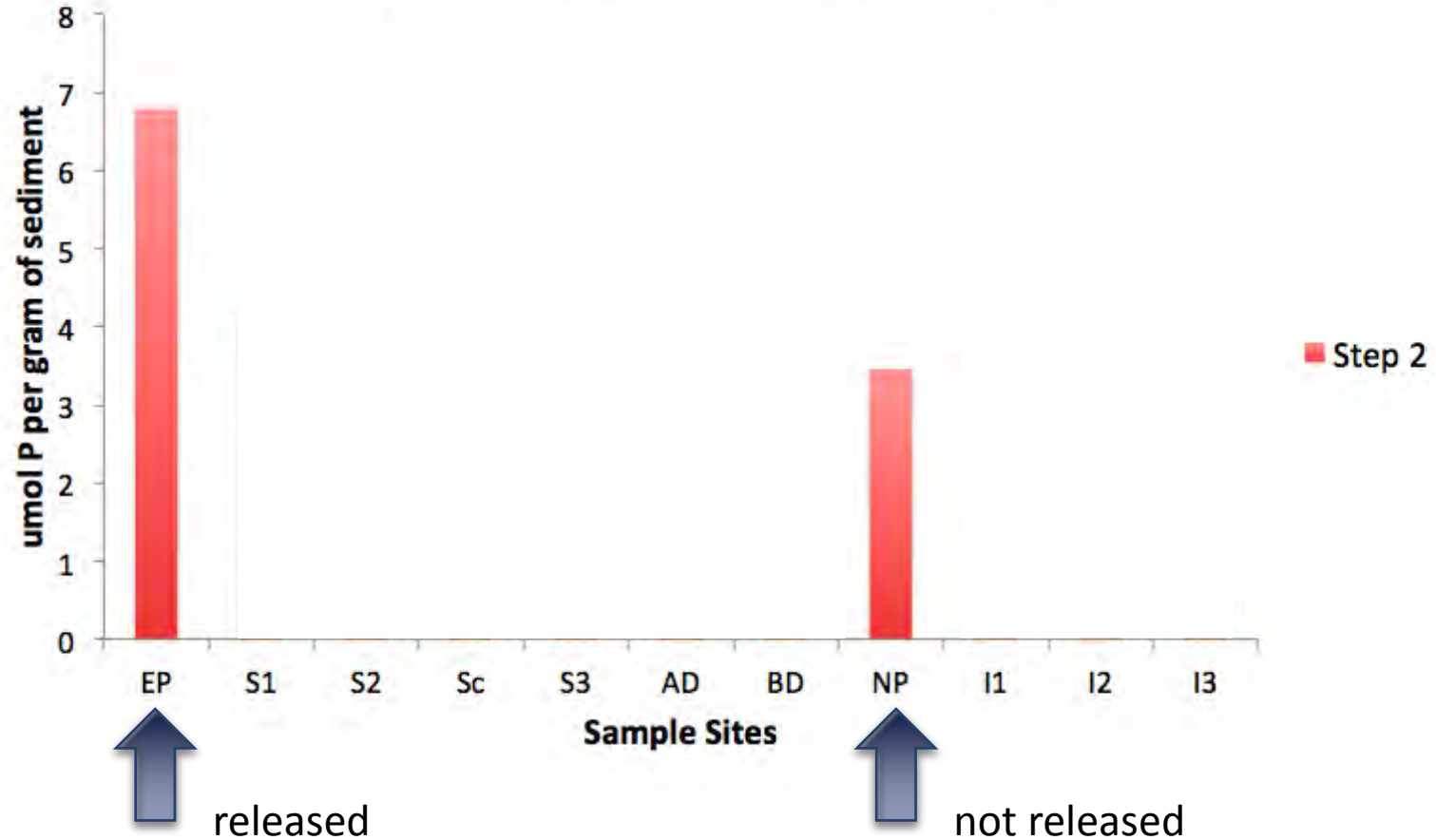
Lake Stratification



Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

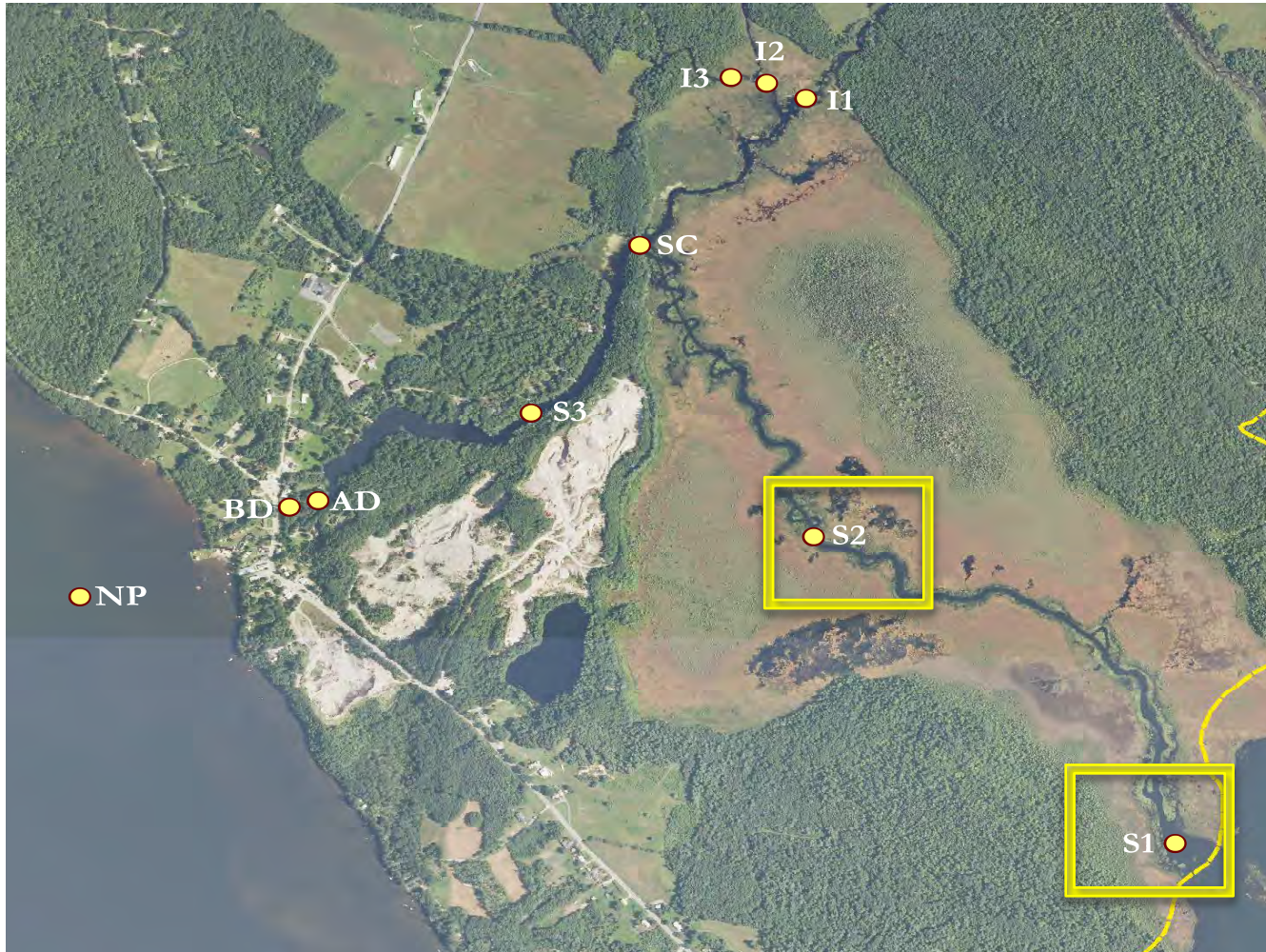
How much more is being released?

P Released in Anoxic Conditions



Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

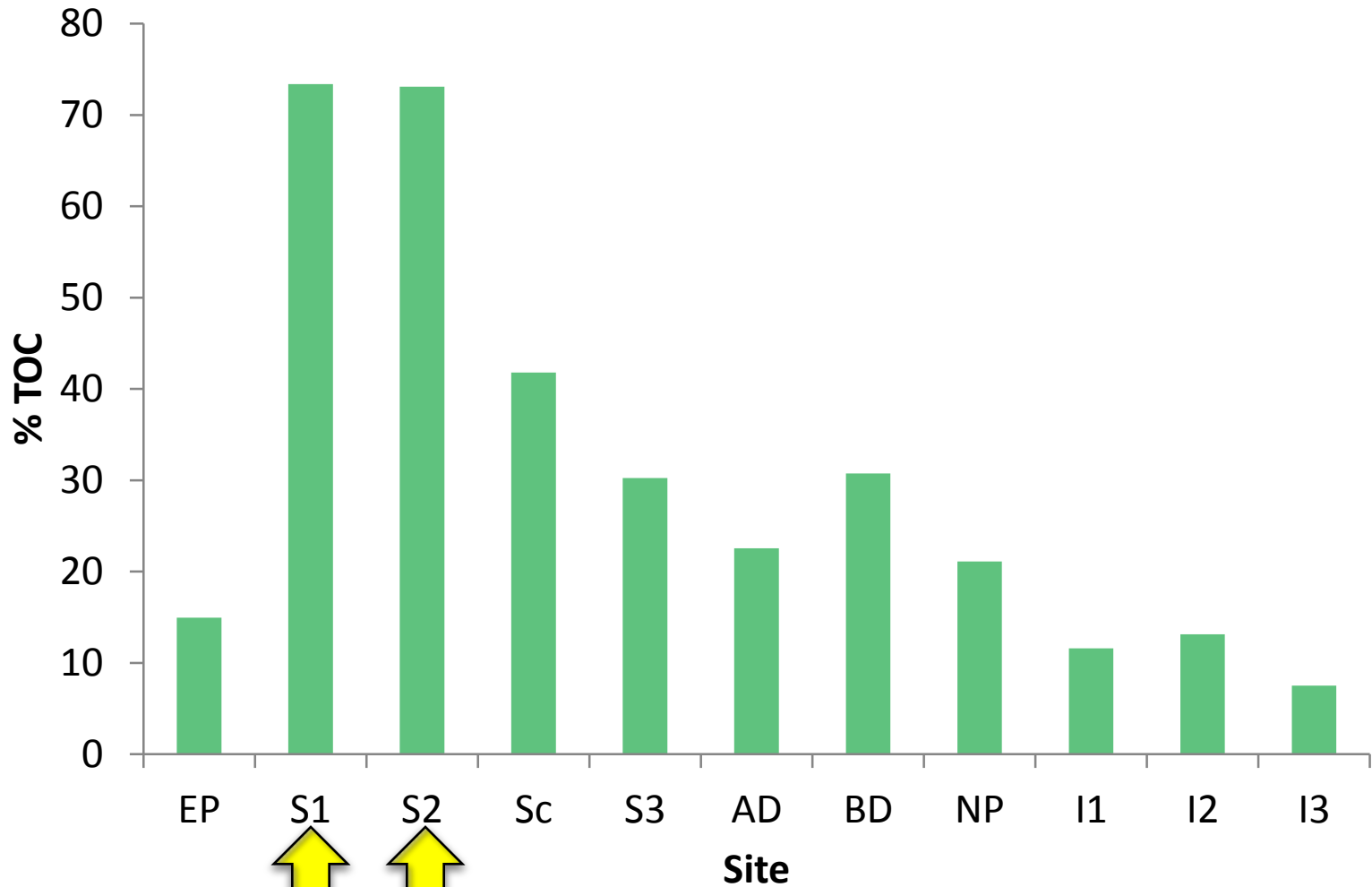
Discussion: S1 and S2



- High total organic carbon
- High dissolved oxygen
- High sphagnum moss presence
- Sphagnum implies low rates of decomposition

Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Total Organic Carbon (TOC)

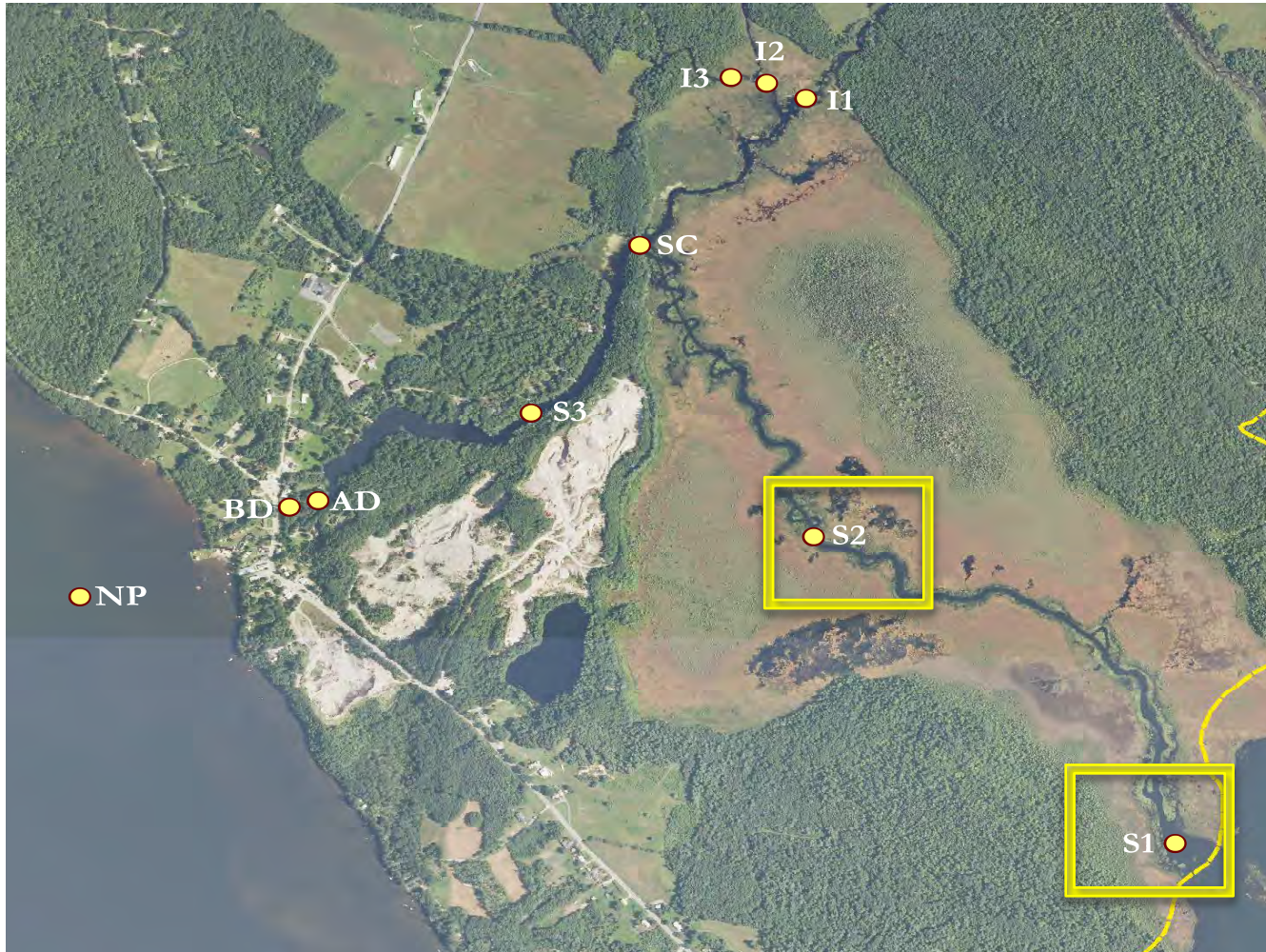


Al sequesters **P**

Fe releases **P** from sediment in ANOXIC environments

P causes algal blooms

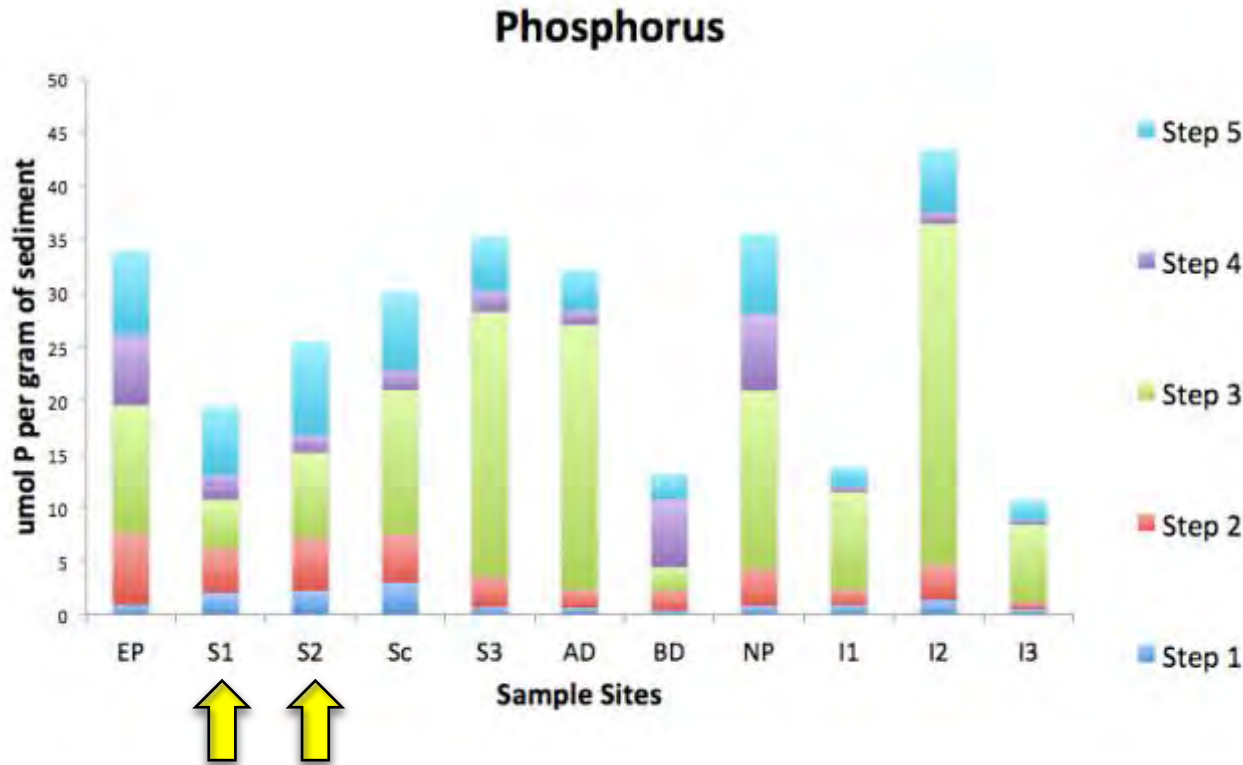
Discussion: S1 and S2



- High total organic carbon
- High dissolved oxygen
- High sphagnum moss presence
- Sphagnum implies low rates of decomposition

Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Discussion: S1 and S2



- High total organic carbon
- High sphagnum moss presence
- **Relatively low nutrients in the water and sediment**
- **Do not expect large nutrient release**

Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

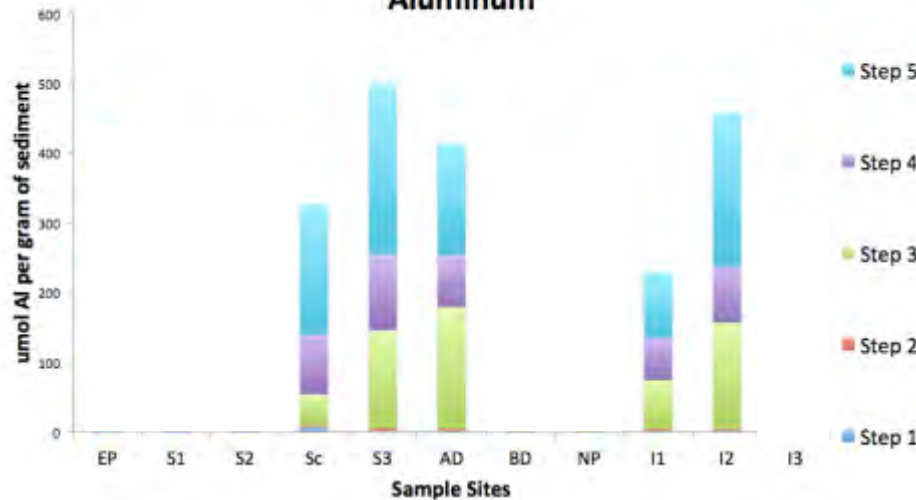
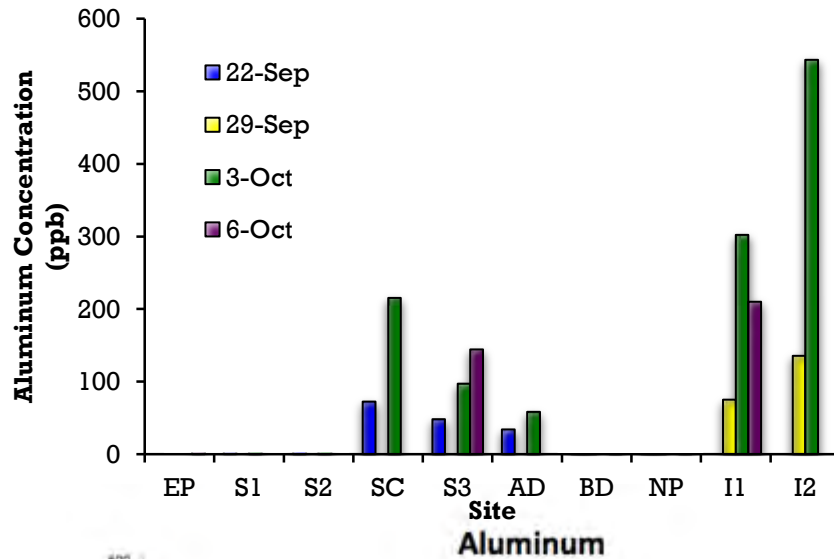
Discussion I2, SC, S3, AD



- **High P, Al, and Fe in water column at I2**
- Appears as though Al and P are precipitating out of water column
- Fe does not show this same correlation

Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

Discussion I2, SC, S3, AD



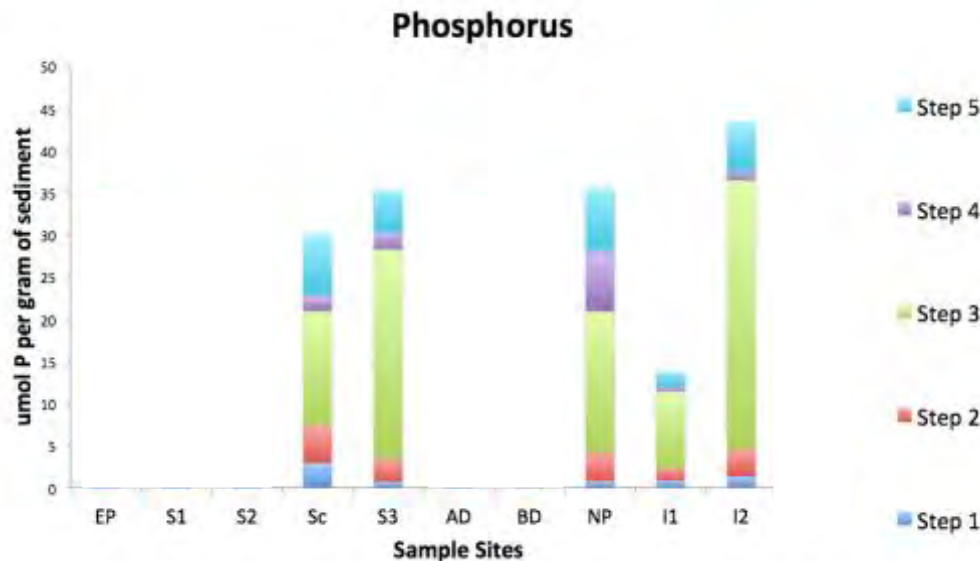
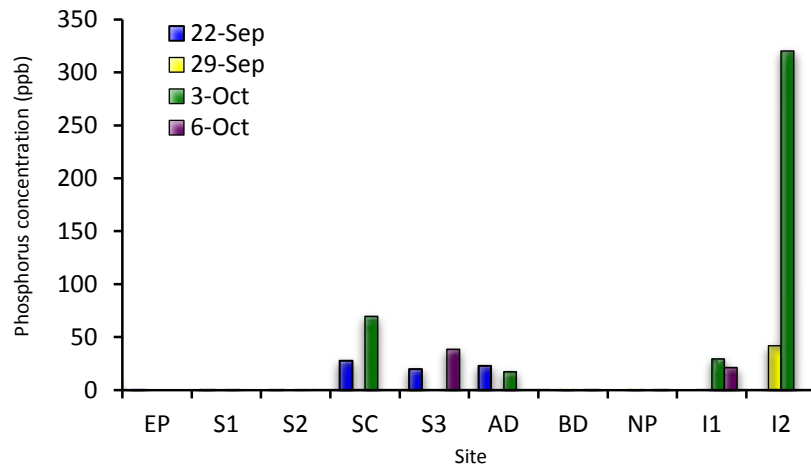
- High P, Al, and Fe in water column at I2

- **Appears as though Al and P are precipitating out of water column**

- Fe does not show this same correlation

Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

Discussion I2, SC, S3, AD



- High P, Al, and Fe in water column at I2
- **Appears as though Al and P are precipitating out of water column**
- **Fe does not show this same correlation**

Al sequesters P **Fe releases P** from sediment in ANOXIC environments **P** causes algal blooms

Discussion I2, SC, S3, AD



- Sediment appears to act as a nutrient sink

Al sequesters **P** **Fe** releases **P** from sediment in ANOXIC environments **P** causes algal blooms

Sediment Take Home Messages

- East Pond blooms because it becomes anoxic
- Sediment is not a source of nutrients in the Serpentine
- The Serpentine may act more as a sink for nutrients than an input

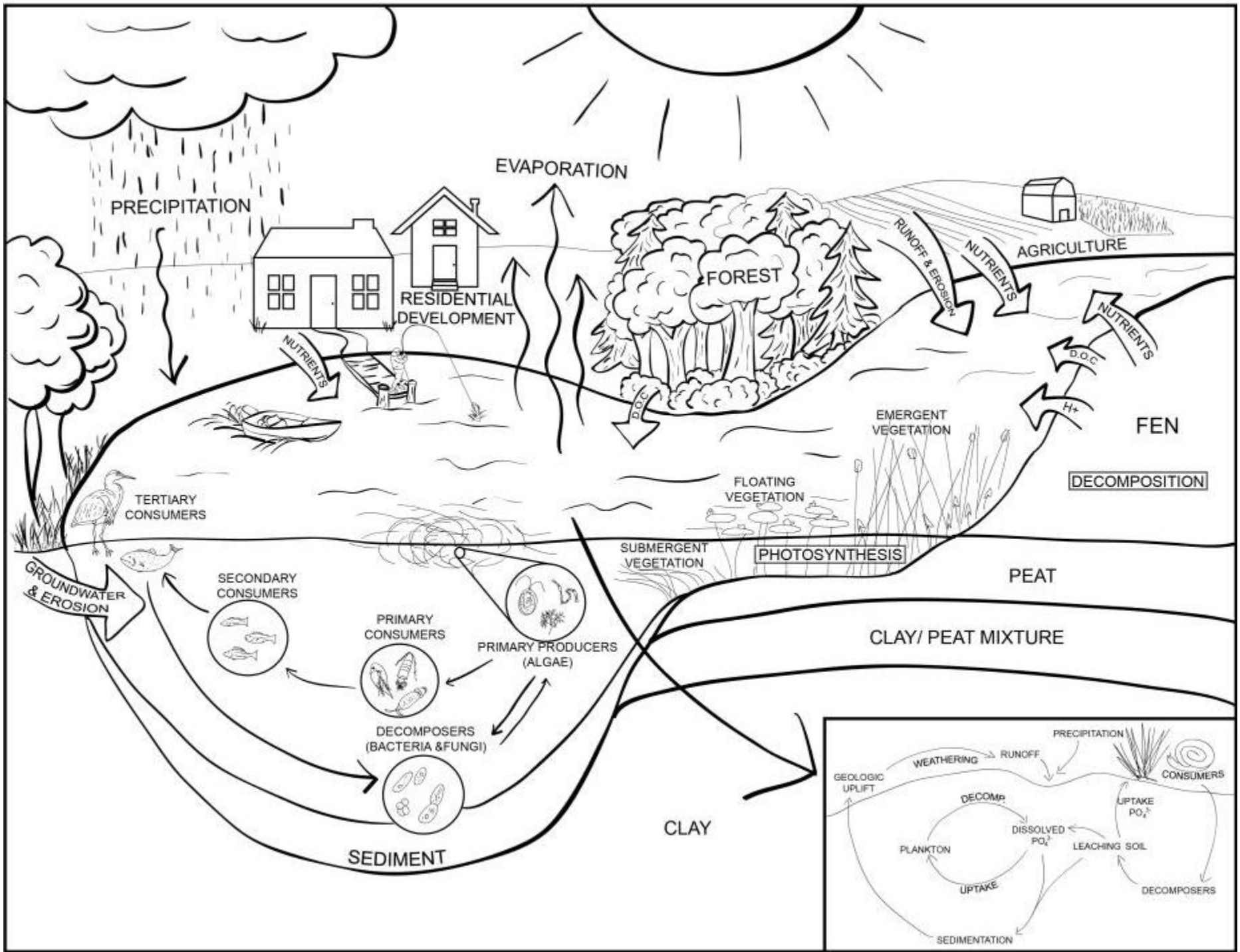




Conclusions

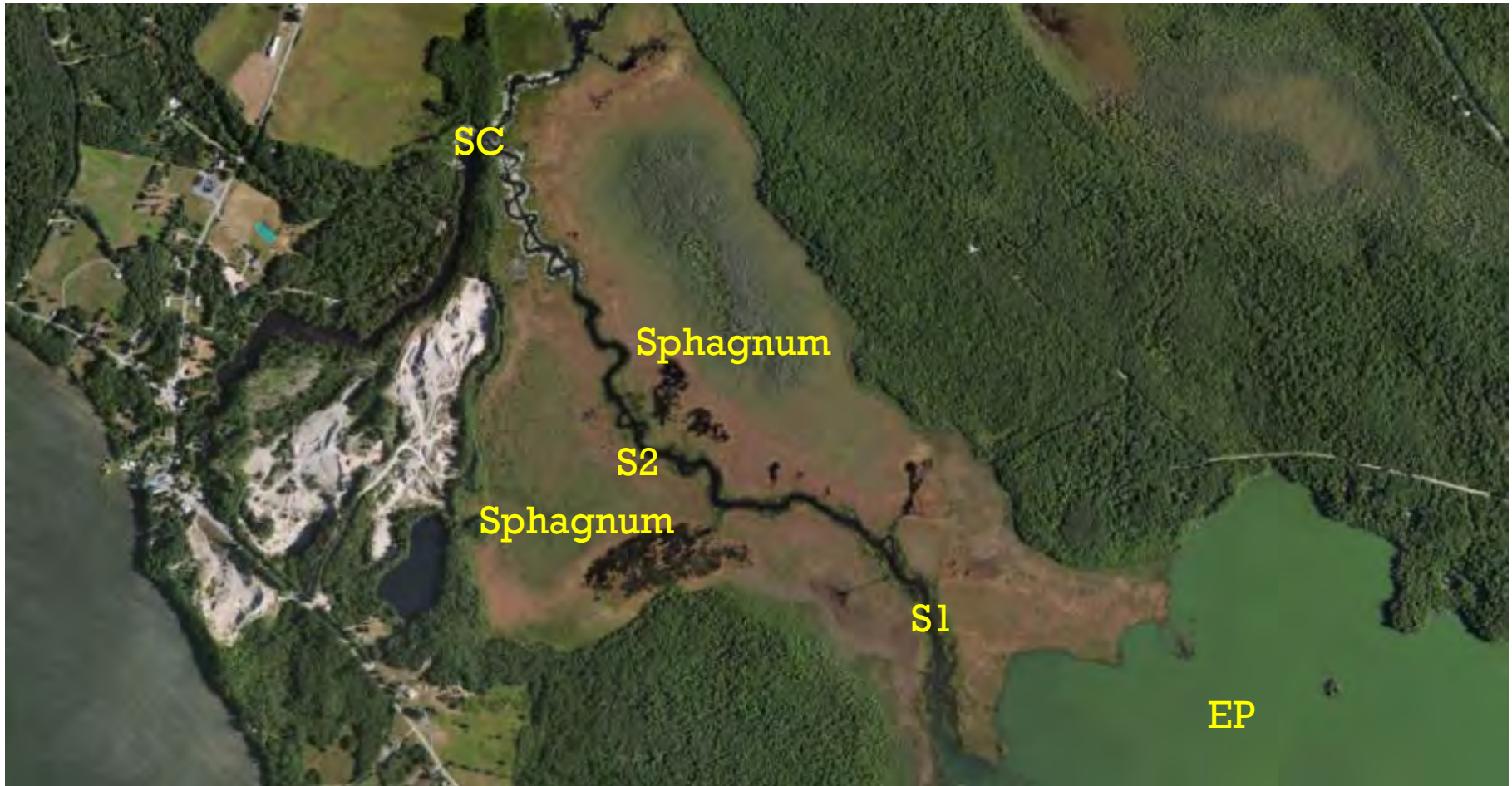
Objective:

- Understand what effects the Serpentine has on East and North Ponds



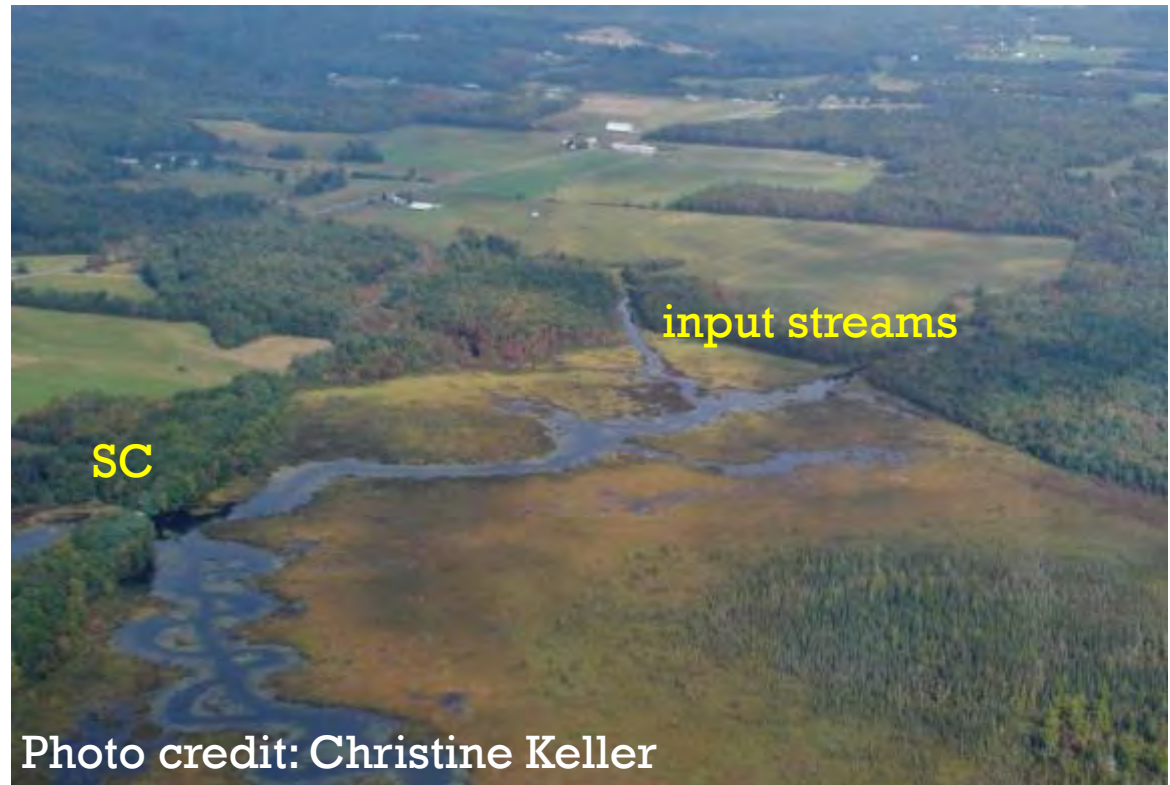
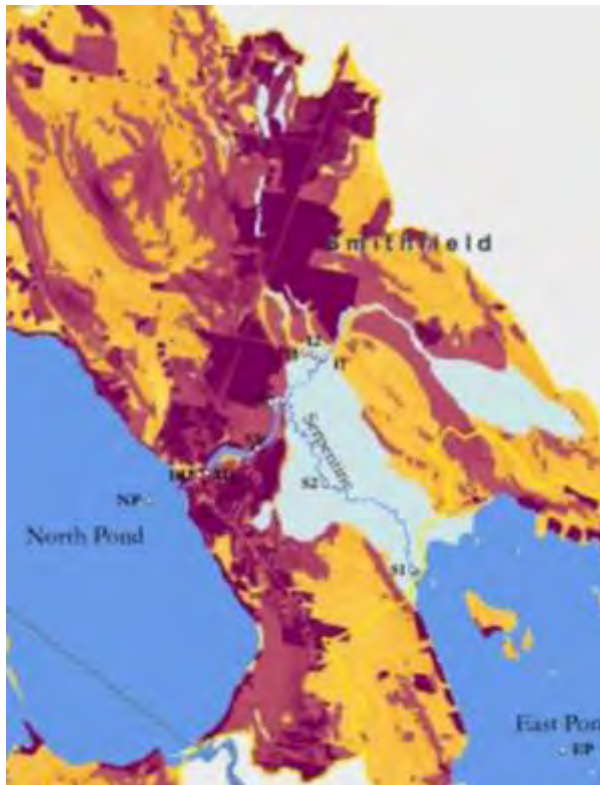
Conclusions

- Sphagnum acting as a decomposition inhibitor and subsequent nutrient sink



Conclusions

- Land-use: High erosion potential and high agricultural area near input streams
- High P, Al, Fe in surface water at input streams and Serpentine confluence (SC)



Conclusions

- Lower P, Al, Fe in surface water from SC down to dam and North Pond



Conclusions

- Lower P, Al, Fe in surface water from SC down to dam and North Pond
 - **H1: Biological uptake**
 - H2: P is binding with Al, Fe and precipitating out into the sediment
 - H3: Dilution



Conclusions

- Lower P, Al, Fe in surface water from SC down to dam and North Pond
 - H1: Biological uptake
 - **H2: P is binding with Al, Fe and precipitating out into the sediment**
 - H3: Dilution



Conclusions

- Lower P, Al, Fe in surface water from SC down to dam and North Pond
 - H1: Biological uptake
 - H2: P is binding with Al, Fe and precipitating out into the sediment
 - **H3: Dilution**



The Take Home Message

Serpentine acting as a sink, rather than
a source for nutrients



Broader Implications

The Serpentine is a missing link ...

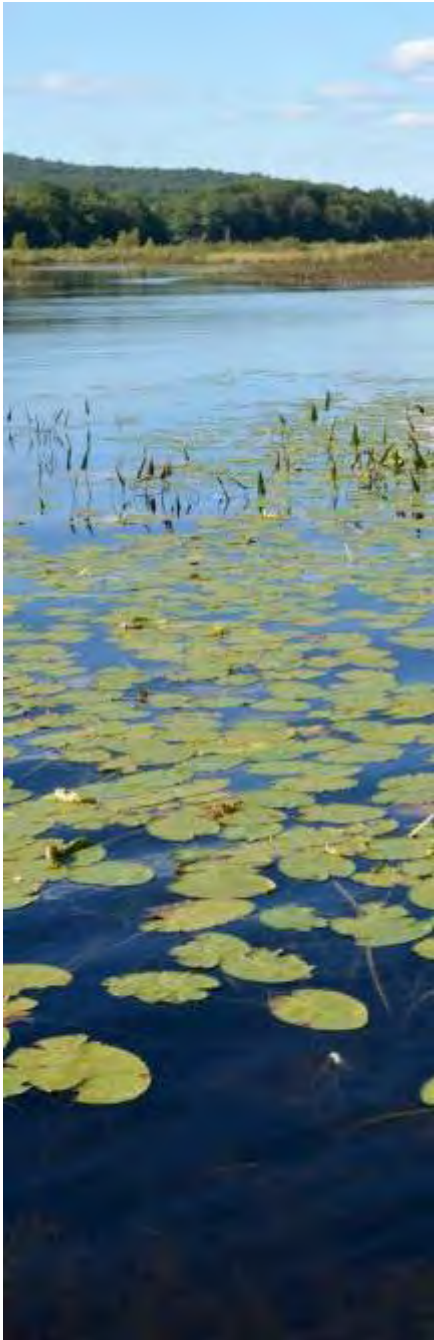
- Important for mitigating nutrient loading from the input streams
- Maintaining health of the Serpentine will help maintain North Pond water quality
- To ensure informed management decisions input streams must be investigated at high resolution
- Collaboration between stakeholders will continue to be necessary in protecting the Belgrade Lake ecosystem



Photo credit: Christine Keller

Acknowledgements

- Christine Keller
- Dr. Stephanie Schmidt
- Abby Pearson
- Dr. Whitney King
- Dr. Manny Gimond
- Dr. Peter Kallin, BRCA
- Sean Boyd
- Bobby van Riper
- Colby College Environmental Studies Department
- Colby College Department of Geology
- Residents of the Serpentine





Questions?